Oligosaccharide Analogues of Polysaccharides

Part 26

Mimics of Cellulose I and Cellulose II: Di- and Monoalkynyl C-Cellosides of 1,8-Disubstituted Anthraquinones

by Kadiyala V. S. N. Murty, Tong Xie, Bruno Bernet, and Andrea Vasella*

Laboratorium für Organische Chemie, ETH Zürich, CH-8093 Zürich

The anthraquinone derivatives T-x-x ($x=2, 4$, and 8), possessing two cellobiosyl, cellotetraosyl, and cellooctaosyl chains, respectively, C-glycosidically bonded at $C(1)$ and $C(8)$ were synthesised as potential mimics of cellulose I. The anthraquinone template enforces a parallel orientation of the cellodextrin chains at a distance corresponding to the one between the crystallographically independent chains of cellulose I, and the ethynyl and buta-1,3-diynyl linker units ensure an appropriate phase shift between them. The H-bonding of the T-x-x mimics was analysed and compared to the one of the mono-chained analogues T-x and of the known cellulose II mimics N-x-x and N-x where one or two cellodextrin chains are O-glycosidically bonded to naphthalene-1,8-diethanol, or to naphthalene-1-ethanol. The OH signals of T-x and T-x-x in solution in (D_6) DMSO were assigned on the basis of DQFCOSY, HSQC, and TOCSY (only of T-4, T-4-4, and T-8-8) spectra and on a comparison with the spectra of N-x and N-x-x. Hydrogen bonding was analysed on the basis of the chemical shift of OH groups and its temperature dependence, coupling constants, SIMPLE 1 H-NMR experiments, and ROESY spectra. **T-4-4** and **T-8-8** in (D_6) DMSO appear to adopt a V-shape arrangement of the cellosyl chains, avoiding inter-chain H-bond interactions. The well-resolved solid-state CP/MAS¹³C-NMR spectra of the mono-chained T-x ($x = 1, 2, 4$, and 8) show that only **T-8** is a close mimic of cellulose II. While the solid-state CP/MAS ¹³C-NMR spectrum of the C_1 symmetric diglucoside T-1-1 is well-resolved, the spectra of T-2-2 and T-4-4 show broad signals, and that of **T-8-8** is rather well resolved. The spectrum of **T-8-8** resembles that of cellulose I_{β} . A comparison of the X-ray powder-diffraction spectra of T-8-8 and T-8 with those of celluloses confirms that T-8-8 is a H-bond mimic of cellulose I and T-8 one of cellulose II.

Surprisingly, there is little difference between the CP/MAS 13C-NMR spectra of the acetyl protected mono-chained C-glycosylated anthraquinone derivatives $\mathbf{A}\text{-}\mathbf{x}$ and the double-chained $\mathbf{A}\text{-}\mathbf{x}\text{-}\mathbf{x}$ ($\mathbf{x}=2, 4$, and 8). The spectra of **A-4** and **A-4-4** resemble strongly the one of cellulose triacetate I (CTA I). The (less well-resolved) spectra of the cellooctaosides A-8 and A-8-8, however, resemble the one of CTA II. The similarity between the solid-state CP/MAS¹³C-NMR spectra of A-4 and A-4-4 to the one of CTA I, and of A-8 and A-8-8 to the one of CTA II is opposite to the observations in the acetylated cellodextrin series.

The mono-chained $\bf{A-x}$ cellulose triacetate mimics 21 ($\bf{A-2}$), 32 ($\bf{A-4}$), and 55 ($\bf{A-8}$) were synthesised by Sonogashira coupling of the cellooligosyl-ethynes 15, 28, and 50, followed by selective deacetylation. Complete deacetylation provided the corresponding T-x mimics. The double-chained A-x-x mimics 24 (A-2-2), 35 (A-4-4), and 58 (A-8-8) were prepared from A-x by triflation and Sonogashira coupling with the cellosyl-buta-1,3-diynes 19, 31, and 53. Their deacetylation provided the corresponding T-x-x mimics 25, 36, and 59. The cellooligosyl-ethynes and cellooligosyl-buta-1,3-diynes required for the *Sono*gashira coupling were prepared by stepwise glycosylation of the partially O-benzylated β -cellobiosylethyne and β -cellobiosyl-buta-1,3-diyne 13 and 17, respectively, with the cellobiosyl donor 2 and the cellohexaosyl donor 47.

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Introduction. – There are at least four polymorphs of cellulose, termed cellulose I–IV [1] [2]. Common to all of them is the ⁴C₁ conformation of the 1,4-linked β -D-glucopyranosyl moieties and the intra-chain, inter-residue $O(3)-H \cdots O(5')$ H-bond, as it was reviewed on several occasions ([3] [4] and refs. quoted there). Most important are the native, metastable cellulose I and cellulose II, the most-stable polymorph [5]. They differ by their crystal packing, *i.e.*, by the relation between adjacent chains of the unit cell. Native cellulose I preparations of different origin show subtle differences in their X-ray diffraction pattern that were rationalised when Atalla and VanderHart found that samples of native cellulose are composed of two crystalline phases, cellulose I_a and cellulose I_b [6]. These phases are also evidenced by solid-state CP/MAS $13C-NMR$, Raman [7], and IR spectroscopy [8], and by electron diffraction [9–11]. Celluloses I_a and I_a are not only found within a single cellulose sample [11], but also along a single microfibril [12]. Their relative amounts vary between samples of different origin. Cellulose I_{α} -rich specimens are found in the cell walls of some algae and in bacterial cellulose, and cellulose I_β -rich specimens in cotton, wood, and ramie fibers [11-13].

The currently accepted three-dimensional structure of celluloses is mainly based on X-ray diffraction data of microcrystals and on computer modeling [14] [15]. The crystal structure of cellulose I_a and I_b , and the H-bond patterns were extensively investigated by Sugiyama et al. [12] and Langan and co-workers [14]. According to Sugiyama et al., cellulose I is a composite of a triclinic cellulose I_a phase with P1 symmetry and a monoclinic cellulose I_β phase with P_1 symmetry [12] [13] [16]. According to the model of Langan et al. [17], cellulose II (obtained from cellulose I by regeneration or mercerisation) differs from cellulose I_β mainly by the antiparallel orientation of the origin and centre chains, the gt conformation of the glucopyranosyl units, and intersheet H-bonds.

The structure of celluloses was also studied by solid-state CP/MAS¹³C-NMR spectroscopy $[18]$. Analysis of selectively ¹³C-labeled celluloses resulted in an unambiguous assignment of the 13 C-NMR peaks [19], while analysis of cellulose fibers from various sources allowed to discriminate between chains in the interior and at the surface of the crystallites. The different conformations of chains in the interior $(tg$ conformation) and at the surface (gt and gg conformation) reveal that the characteristic inter-residue $C(6)OH \cdots OHC(2')$ H-bonds of native celluloses are cleaved at the surface and replaced by inter-residue $C(2)OH \cdots OHC(6')$ and/or intermolecular H-bonds to H₂O or to amorphous cellulose¹). Chains in the interior of the crystallite of native cellulose show a characteristic downfield shift of $C(4)$ (88–91 ppm) and $C(6)$ (65–67 ppm), as compared to $C(4)$ and $C(6)$ of chains at the surface that resonate at 83–85 and 62.9/ 61.7 ppm $[20-24]$. Recent calculations suggest that H-bonding leads to a shortening of the C(4)–O bonds from 1.43 to 1.36 Å and to the downfield shift of C(4) of cellulose I_{α} and I_{β} [25].

Cellotetraose and cellotriose are considered valid models of cellulose II, since single-crystal X-ray analysis of cellooligomers such as cellotetraose $[26-28]$ and cellotriose [29] showed that their structure resembles the one of cellulose II. There are no similar model compounds for cellulose I. To this day, precise structural data defining the atomic details of native cellulose are missing. For this reason, we decided to synthe-

¹) The different conformations of interior and surface chains of the crystallites were also visualised by atomic-force microscopy [9].

sise model compounds of cellulose I, opting for compounds where cellodextrin chains of increasing length are attached in a parallel way to a suitable template.

We already described the design and synthesis of the cellooligomers N-x and N-x-x $(x=2, 3, 4,$ and 8)²) where the cellooligosyl chains are glycosidically attached to naphthalene-1-ethanol and naphthalene-1,8-diethanol (cf. Fig. 2) [30]. However, solid-state CP/MAS ¹³C-NMR spectroscopy characterised the bis-cellooctaoside N-8-8 as a mimic of cellulose II rather than of cellulose I, showing that a parallel attachment of two cellooligosaccharide chains to naphthalene-1,8-diethanol is not sufficient to generate a model for cellulose I_a or I_β [4]. It appeared necessary to design a templated model that assures a parallel orientation of the chains, and that mimicks not only the distance between the origin and centre chains of cellulose I but also the phase shift between them. MM3* Calculations (Macromodel V. 6, gas phase [31]) showed that 1-(buta-1,3-diynyl)-8-ethynylanthraquinone fulfils these conditions. It fixes the chains in a parallel orientation, and leads to a distance between two C-glycosidically attached cellooligosaccharide chains of 6.02 Å and to an appropriate phase shift of 2.59 Å, as discussed before [3]. A further advantage is the anticipated difference of the chemical shift of the NMR signals of the template and of the cellooligosaccharide chains.

We planned to investigate the mono- and the double-chained cellobiosyl, cellotetraosyl, and cellooctaosyl anthraquinonyl derivatives; this should allow to also study the influence of the chain length on the intramolecular, inter-chain interactions. The synthesis of the symmetric and asymmetric β -D-glucopyranosyl-ethynylated and -buta-1,3-diynylated anthraquinones, and the analysis of their H-bonding in solution were published [3]. We now describe the synthesis of the corresponding templated mono- and double-chained cellooligosides and the investigation of their H-bonding in the solid state and in DMSO solution; we also describe the synthesis and spectra of the corresponding analogues of cellulose triacetate (CTA).

Results and Discussion. - 1. Synthesis of β -Cellobiosyl-ethyne and β -Cellobiosylbuta-1,3-diynes. We planned to synthesise the templated even-numbered cellooligosides (exemplified by the cellobiose derivative \bf{A}) by coupling the corresponding ethynyl and butadiynyl C-glycosides, such as the protected cellobiosyl-ethyne and cellobiosyl-buta-1,3-diyne units C and D to an anthraquinone moiety B , as described for the synthesis of the corresponding glucosyl analogue $[3]$ (*Fig. 1*). The required cellobiosyl C-glycosides should be readily available via a protected cellobionolactone E that should be obtained from commercial cellobiose octaacetate 1 [30].

Cellobiose octaacetate (1) was treated with HBr in AcOH to give 94% ³) of the known bromide 2 [32] (*Scheme 1*). Glycosidation of allyl alcohol with 2 in the presence of Hg(CN)₂ yielded 85% of the acetylated β -cellobioside 3 [33]. Following the procedure of Vliegenthart and co-workers [34], 3 was deacetylated (MeONa in MeOH; yield $>98\%$). The resulting allyl cellobioside 4 was transformed into the benzylidene acetal 5 (89%) [35] [36] and further in 73% yield to the benzylated 6 [34] [36]. To obtain higher melting derivatives we also transformed 5 to the 4-chlorobenzyl ether 7 (83%) with a melting point $ca. 20^{\circ}$ higher than that of 6. Regioselective reductive cleavage of

²⁾ For the numbering of these cellosides, see Results and Discussion, Chapt. 4.

³) The yield of **2** was 85–80% when the reaction was performed on a scale >80 g.

Fig. 1. Retrosynthesis of the template-bound cellobiosyl-acetylenes and cellobiosyl-buta-1,3-diynes

the 1,3-dioxane ring of the benzylidene acetal 7 with NaBH₃CN and HCl [30] gave a mixture that was separated by flash chromatography (FC) to provide 69% of the desired secondary alcohol 8. Reduction of 7 with $BH_3 \cdot Me_2NH/BF_3 \cdot Et_2O$ [37] followed by FC gave 8 in a lower yield (60%), while reduction of 7 with Et₃SiH and BF₃ · Et₂O [38] yielded 71% of 8. Methoxymethylation of 8 gave 9 (70%) besides 17% of starting material. However, this derivative could not be converted satisfactorily to the corresponding lactone. Deallylation of 9 under standard conditions [34] followed by treatment with DMSO and Ac₂O [39] or with *Dess-Martin* periodinane [40] gave complex mixtures. We, therefore, isomerised the benzylidenated allyl glycoside 7 according to Baudry et al. [41], and cleaved the resulting prop-1-enyl glycosides with I_2 in THF/ $H₂O$ to obtain 87% of the hemiacetals 10. Oxidation of 10 with DMSO and Ac₂O [39] provided 97% of crude lactone 11.

Stereoselective methods for the preparation of β -D-hexopyranosyl-alkynes from hexonolactones are well-documented [42-46]. Treatment of the cellobionolactone 11 with 1 equiv. of lithium (trimethylsilyl)acetylide ($\text{LiC}\text{ }\equiv \text{CSi}\text{Me}_3$) in THF led to only little conversion (Scheme 2). Slightly better results were obtained by using up to 2 equiv. of LiC \equiv CSiMe₃, in agreement with the reported sluggish addition of alkynyl metal reagents to cellobionolactones [47]. Cerium [45] [48] or aluminium acetylides [49] performed hardly better, even when 2 equiv. of the reagent were used, and the same result was obtained upon adding $LiC \equiv CSiMe_3$ in the presence of *Lewis* acids, such as scandium or ytterbium triflate, in either THF or Et₂O. However, treatment of 11 with 4.5 equiv. of LiC \equiv CSiMe₃ led to complete conversion to 12 (α/β 55:45). Performing this addition on a multigram scale followed by reductive dehydroxylation and concomitant

a) HBr in AcOH, CHCl₃; 94%. b) Hg(CN)₂, allyl alcohol; 85%. c) MeONa, MeOH; >98%. d) PhCH(OMe)₂, MeCN, TsOH·H₂O; 89%. e) BnBr, NaH, DMF; 73%. f) 4-ClC₆H₄CH₂Cl, NaH, DMF ; 83%. g) NaBH₃CN, HCl in Et₂O, THF; 69%. h) MeOCH₂Cl, NaH, DMF; 70%. i) [Ir(MePh₂P)₂- (C_8H_{12})]PF₆, H₂, THF, then I₂ in H₂O; 87%. *j*) Ac₂O, DMSO; 97% of crude 11.

reductive opening of the 1,3-dioxane ring with Et_3SH and $BF_3 \cdot Et_2O$ [38] yielded 72% of the β -configured alcohol 13. C-Desilvlation of 13 to 14 (MeONa in MeOH [50], 89%), followed by acetolytic debenzylation (TMSOTf in Ac₂O) gave the crystalline cellobiosyl-acetylene 15 (76%).

The butadiyne 19 was prepared in a similar way. Addition of the lithium acetylide derived from bis[1,4-(trimethylsilyl)buta-1,3-diyne] to the cellobionolactone 11 gave the hemiketals 16 (α/β 3:2). Concomitant dehydroxylation of 16 and reductive dioxane ring opening yielded 52% of the cellobiosyl-buta-1,3-diyne 17. In contradistinction to the preparation of the alkyne 12, the preparation of the butadiyne 16 required only a slight excess of the alkynyl reagent. C-Desilylation of 17 to 18 (87%) followed by acetolytic debenzylation gave the crystalline cellobiosyl-buta-1,3-diyne 19 (63%).

Sonogashira coupling $[51-53]$ of the anthraquinone triflate 20 [3] and the cellobiosylacetylene 15 at 60° followed by selective deacetylation of the phenolic AcO group $((NH_4)_2CO_3$ in DMF [3] [54] [55]) gave 74% of the C-glycoside 21 that was triflated under standard conditions to yield 96% of the aryl triflate 22 (Scheme 3). Sonogashira coupling of 22 with the cellobiosyl butadiyne 19 at room temperature [52] [53] yielded 87% of the bis-C-cellobiosylated anthraquinone 24. Deacetylation of 21 with MeONa in MeOH gave 95% of the mono-C-glycoside 23. The corresponding bis-C-cellobiosyl derivative 25 was obtained in a yield of 71% by treating 24 with KCN in MeOH/CH₂Cl₂ [3] followed by reversed-phase column chromatography.

The 1 H- and 13 C-NMR data for the glucosyl units of the cellobiosyl-ethynes 13–15, the cellobiosyl-buta-1,3-diynes $17-19$, and their template-bound derivatives $21-25$ are

a) BuLi, Me₃SiC≡CH, THF, -78°. *b*) MeLi·LiBr, Me₃SiC≡C–C≡CSiMe₃, THF, -78°. *c*) Et₃SiH, BF_3 ·OEt₂, CH₂Cl₂/MeCN; 72% of 13 from 11; 52% of 17 from 11. d) MeONa, THF/MeOH; 89% of 14; 87% of 18. e) Ac₂O, Me₃SiOTf, 20-25°; 76% of 15; 63% of 19.

listed in *Tables* 5–8 in the *Exper. Part.* $3J(1, \equiv CH)$ of the ethyne 14 is larger than $5J(1, \equiv$ CH) of the buta-1,3-diyne **18** (2.2 *vs.* 0.6 Hz). $J(1,2)$, $J(2,3)$, $J(3,4)$, and $J(4,5)$ values evidence the 4C_1 conformation of all pyranosyl units. The ethynyl and the buta-1,3-diynyl group have a similar influence upon the ${}^{1}H$ - and ${}^{13}C$ -NMR chemical shifts of the sugar residues, as evidenced by the chemical shift differences ($\Delta\delta$ values) for the correponding pairs 13/17, 14/18, and 15/19 (H-C(1¹): 0.03–0.06, other H-C: \leq 0.03, C(1¹): 0.1–0.4, $C(2^I)$: 0.5–0.7, and other C: \leq 0.2 ppm)⁴). However, the anthraquinonyl moiety of **21** leads to a significant downfield shift, as evidenced by the ¹H-NMR $\Delta \delta$ values for the pair 21 and 16. Large downfield shifts are observed for $H - C(1^1)$ and $H - C(2^1)$ $(0.42 \text{ and } 0.25 \text{ ppm}, \text{resp.})$ and smaller ones for $H - C(3¹)$ to $H - C(6¹) (0.08 - 0.12 \text{ ppm}).$ Even H–C(1^{II}) to H–C(6^{II}) are shifted downfield by 0.04–0.06 ppm. Similarly, the Csignals of 21 are shifted downfield ($\Delta\delta$ for C(1¹): 1.1, C(2¹): 0.2, C(3¹): 0.6, and C(4¹) to $C(6^{II})$: ≤ 0.1 ppm). $C(1^I)$ of the cellobiosyl-ethynes and cellobiosyl-buta-1,3-diynes res-

⁴⁾ According to the 1996 recommendations of the nomenclature of carbohydrates, the monomer units of the cellooligosides are numbered with roman numerals in ascending order starting from the alkynylated end (corresponding to the reducing end).

a) $[Pd(PPh₃)₂][Cl₂, Cl₁, Et₃N/DMF 1:5, 60°, then (NH₄)₂CO₃; 74%. b) As a, but at 24° and without$ $(NH_4)_2CO_3$; 87%. c) Tf₂O, Et₃N, CH₂Cl₂; 96%. d) MeONa, MeOH; 95%. e) KCN, MeOH/CH₂Cl₂; 70%.

onates upfield at $68.3 - 70.4$ ppm, whereas the signal for $C(5^I)$ is shifted downfield to 76.8–79.3 ppm (compare with 74.9 ppm for the β -celloside 7). The position of the ¹³C-NMR signals of the ethynyl and the buta-1,3-diynyl C-atoms is characteristically influenced by the terminal substituent $(Me_3Si, H, or an thraquinonyl)$. The assignment of the 1 H- and 13 C-NMR signals of 24 and 25 to the ethynylated or buta-1,3-diynylated chain is based on the interpretation of DQFCOSY and HSQC spectra and on a comparison with the values of the mono-chained cellobiosyl-ethynes 21 and 23, and the mono- and double-chained glucopyranosyl-ethynes and glucopyranosyl-buta-1,3 diynes (data in [3]). Due to the shorter distance to the anthraquinonyl moiety, the ¹H-NMR signals of the ethynylated chain of 24 (except the one for $H-C(4^I)$) are shifted downfield relative to the corresponding signals of the buta-1,3-diynylated chain; strongest downfield shifts are observed for H–C(1¹), H–C(2¹), and H–C(5¹) ($\Delta\delta$ = 0.25, 0.18, and 0.10 ppm, respectively).

2. Synthesis of the Template-Bound Cellotetraosyl-ethynes and Cellotetraosyl-1,3 butadiynes. We planned to prepare the intermediate acetylated cellotetraose C-glycosides 28 and 31 *via* 26 and 29 that should be obtained by *Koenigs–Knorr* glycosylation [56] with the cellobiosyl bromide 2 of the selectively protected cellobiosyl-ethyne 13 and cellobiosyl-buta-1,3-diyne 17, respectively (Scheme 4). Glycosidation in toluene/ CH₂Cl₂ 1 :1 with 4 equiv. of 2 at -35° to 25° for 2 days and in the presence of excess AgOTf yielded 84% of the crystalline cellotetraosyl-ethyne 26, while the AgOTf-promoted glycosidation of 13 by 2 in ClCH₂CH₂Cl at -35° gave only traces of 26, and changing the solvent to CH_2Cl_2 or MeCN hardly improved the conversion. C-Desilyla-

a) AgOTf, 3-Å mol. sieves, toluene/CH₂Cl₂: 84% of **26**; 82% of **29**, b) Bu_{NF}·3 H₂O, THF; 86% of 27; 94% of 30. c) Me₃SiOTf, Ac₂O; 76% of 28; 65% of 31.

tion of 26 with Bu₄NF \cdot 3 H₂O afforded 86% of 27. Acetolytic debenzylation of 27 gave the crystalline peracetylated cellotetraosyl-ethyne 28 (76%).

The cellotetraosyl-buta-1,3-diyne 31 was similarly synthesised from the bromide 2 and the cellobiosyl-buta-1,3-diyne 17. Glycosylation of 17 with 2 under the same conditions that we used for the synthesis of 26 yielded 82% of 29. Desilylation of 29 gave the terminal butadiyne 30 (94%) that was transformed by acetolysis to the peracetate 31 (65%).

Sonogashira coupling of the triflated anthraquinone 20 with the cellotetraosyl-acetylene 28 under the conditions used for the preparation of the cellobiosyl-ethyne 21 and

the cellobiosyl-buta-1,3-diyne 24 produced a complex mixture (Scheme 5). Only little conversion resulted from performing the coupling at ambient temperature. Addition of $Bu₄NI$ (1.5 equiv.) proved beneficial, as observed by *Yamaguchi* and co-workers [57], and the coupling of 20 with the cellotetraosyl-acetylene 28 occurred cleanly at ambient temperature⁵). Selective deacetylation of the crude coupling product and FC gave 75% of the cystalline cellotetraosylated 1-hydroxyanthraquinone 32. Triflation under standard conditions transformed 32 to 33 (95%). The Bu₄NI-promoted coupling of this triflate with the cellotetraosyl-1,3-butadiyne 31 gave the crystalline bis-cellotetraoside 35 (59%). The mono-cellotetraoside 32 was deacetylated with MeONa in MeOH to provide 95% of 34, while complete deacetylation of the bis-cellotetraoside 35 to 36 proved difficult. Standard methods, such as treatment with MeONa in MeOH, K_2CO_3 in MeOH, or KCN in MeOH failed to provide the desired product. Deacetylation with aqueous solutions of trialkylamines, such as $Et₃N$ led to a very poor conversion even after 7 days, while aqueous KOH (30 equiv.) provided only small amounts of 36. Finally, 78% of the completely deacetylated 36 were obtained by using $28-29$ equiv. of Bu₄NOH in H₂O.

The ¹H- and ¹³C-NMR data for the glucosyl units of the cellotetraosyl-ethynes and -buta-1,3-diynes $26 - 36$ are listed in Tables $9 - 12$ in the Exper. Part. The acetoxylated and benzyloxylated centres of 26, 27, 29, and 30 show the expected relative chemical shifts (¹H-NMR: downfield shift for acetoxylated centres, ¹³C-NMR: downfield shift for benzyloxylated centres). The ¹H- and ¹³C-NMR chemical shifts for unit I and IV of the cellotetraosyl-ethynes and -buta-1,3-diynes 28 and 31– 36 agree well with the corresponding values for unit I and II of the peracetylated cellobiosyl-ethynes and -buta-1,3-diynes. Unit II and III of 28 and $31-33$ show slightly shifted ¹H- and ¹³C-NMR signals (¹H-NMR: $\Delta\delta \leq 0.03$ ppm with the exception of 0.06–0.08 ppm for H–C(2) of 28 and 31–33 and for H–C(5) of 32 and 33; ¹³C-NMR: $\Delta\delta$ < 0.2 ppm). Individual ¹H- and 13 C-NMR signals are assigned to unit I of the ethynylated and buta-1,3-diynylated chain of 35, whereas the signals for units II to IV of these chains overlap increasingly with the distance of the centres from the anthraquinonyl moiety.

3. Synthesis of the Template-Bound Cellooctaosyl-ethynes and Cellooctaosyl-buta-1,3-divnes. The preparation of the cellooctaosyl-alkynes according to the route described above for 36 requires a cellohexaosyl donor. Such a donor should be accessible in sufficiently large amounts by two sequential glycosidations, starting with cellobiosyl derivatives. The synthesis of the $HO-C(4^{IV})$ unprotected cellotetraoside 43 from the cellobiosyl bromide 2 and the cellobiosyl acceptor 38 [34] [36] was described [30] (Scheme 6). The key step of this synthesis – the AgOTf-promoted Koenigs – Knorr glycosidation of 38 with 1.2 equiv. of 2 in ClCH₂CH₂Cl at -30° – afforded 89% of the cellotetraoside 39. We could not reproduce the high yield by following this procedure. Even using 3 equiv. of 2 and changing the solvent to toluene/ CH_2Cl_2 yielded only 50% of 39. However, 39 was obtained in 87% yield by a TMSOTf-promoted glycosidation of 36 with the trichloroacetimidate 37 [61] in CH₂Cl₂ at -40° . The next steps, *i.e.*, deacetylation to 40, benzylidenation to 41, benzylation to 42, and regioselective reduction to the alcohol 43 were performed according to [30], and resulted in an overall yield

⁵⁾ For other Bu4NI-promoted cross couplings, see [58] [59] and refs. cited therein. For a CuI-promoted Sonogashira coupling, see [53] [60].

a) $[Pd(PPh₃)₂][C₁, CuI, Bu₄NI, Et₃N/DMF 1:5, 24^o, then (NH₄)₂CO₃; 75%. b) As a, but without$ (NH_4) ₂CO₃; 59%. c) Tf₂O, Et₃N, CH₂Cl₂; 95%. d) MeONa, MeOH; 95%. e) Bu₄NOH, H₂O; 78%.

of 71%. The TMSOTf-promoted glycosidation of 43 with the trichloroacetamidate 37 in CH₂Cl₂ at -60° gave the cellohexaoside 44 (94%). Deallylation of 44 followed by hydrogenolytic debenzylation and acetylation yielded 75% of α/β -45 2:3⁶). Selective deacetylation of the anomeric AcO group of α/β -45 with $(NH_4)_2CO_3$ in DMF [54] gave the hemiacetals α/β -46 2 : 1 (78%) [66] that were treated with Cl₃CCN in the presence of DBU. Crystallisation of the product from AcOEt/hexane yielded 71% of a 94 : 6 mixture of the trichloroacetimidates α/β -47 [66]. The cellooctaoside 48 was obtained in 88% by a BF_3 OEt₂-promoted glycosylation of a slight excess of the cellobiosyl-ethyne 13 with α/β -47 94:6, while the TMSOTf-promoted glycosylation of 13 with a slight excess of α/β -47 94:6 at -78° yielded only 40% of 48. C-Desilylation of 48 with $Bu_4NF \cdot 3 H_2O$ to 49 followed by acetolytic debenzylation provided the peracetylated cellooctaosyl-ethyne 50 (75%). Similarly, the cellooctaosyl-buta-1,3-diyne 53 was synthesised via 51 and 52 from α/β -47 94:6 and the cellobiosyl-buta-1,3-diyne 17 (42%) overall yield).

The Bu₄NI-promoted Sonogashira coupling at $0-28^\circ$ of the triflate 20 with the cellooctaosyl-ethyne **50** gave 83% of the octaosylated anthraquinone **54** (Scheme 7). Selective deacetylation of the aromatic AcO group of 54 gave the hydroxy-anthraquinone 55 (88%) which was triflated to 56 (59%). The Bu₄NI-promoted coupling of 56 with the cellooctaosyl-1,3-butadiyne 53 gave 40% of the bis-cellooctaoside 58. The

For the synthesis and NMR data of α -45, see [62-65].

a) Me₃SiOTf, 4-Å mol. sieves, CH₂Cl₂, -40 (38) or -60° (43); 87% of 39; 94% of 44. b) MeONa, MeOH; 99%. c) ZnCl₂, PhCHO; 84%. d) BnBr, NaH, DMF; 93%. e) NaBH₃CN, HCl, 3-Å mol. sieves, THF; 92%. f) H_2 , $[Ir(MePh_2P)_2(C_8H_{12})]PF_6$, THF, then I_2 , H_2O ; Pd(OH)₂/C, 6 bar of H_2 , AcOEt/MeOH; Ac₂O, pyridine; 75% of α/β -45 2:3. g) (NH₄)₂CO₃, DMF; 78% of α/β -46 2:1. h) Cl₃CCN, DBU, CH₂Cl₂; 71% of α/β -47 94 :6. i) BF₃ · Et₂O, 4-A mol. sieves, CH₂Cl₂, $\leq -18^{\circ}$; 88% of 48; 72% of 51. j) Bu₄NF·3 H₂O, THF; 92% of 49; 83% of 52. k) BF₃·OEt₂, Ac₂O; 82 of 50; 70% of 53.

a) Pd(PPh₃)₂Cl₂, CuI, Bu₄NI, Et₃N/DMF 1:5; 83% of 54; 40% of 58. b) (NH₄)₂CO₃, DMF; 88%. c) Tf₂O, Et₃N, CH₂Cl₂; 59%. d) Bu₄NOH, H₂O; 43% of 57; 55% of 59.

mono- and the bis-octaosides 55 and 58 were deacetylated with aqueous Bu₄NOH, similary to 35, but sonification was required to dissolve the octaosides. This provided the deprotected mono-cellooctaoside 57 in 43% and the deprotected bis-cellooctaoside 59 in 55% yield.

The 1 H- and 13 C-NMR chemical shifts for unit I and VIII of the cellooctaosylethynes and cellooctaosyl-buta-1,3-diynes 50, 53-55, and 57-59 resemble those of the terminal units of the corresponding cellobiosyl and cellotetraosyl analogues, whereas the 1 H- and 13 C-NMR signals for units II to VII of the cellooctaosyl derivatives mostly overlap (see Tables 13 and 14 in the Exper. Part).

4. Are the Template-Bound Cellooligosyl C-Glycosides Mimics of Cellulose I? Comparative Analysis of the Template-Bound Cellooligosyl-ethynes and Cellooligosyl-buta-1,3-diynes. To facilitate the discussion, we have replaced the compound numbers of the oligosides of interest, designating the mono- and double-chained anthraquinonyl derivatives as **T-x** and **T-x-x** with **x** and **x-x** ($\mathbf{x} = 2, 4$, and 8) denoting the number of glucosyl residues (Fig. 2,a). Similarly, the peracetates are labeled $\bf{A-x}$ and $\bf{A-x-x}$. The deprotected 2-naphthylethyl cellooligosides [4], of interest as reference compounds, are designated as N-x and N-x-x. The ethynylated chain of $\bf{A-x-x}$ and T-x-x is termed E and the buta-1,3-diynylated chain B. According to carbohydrate nomenclature, the glucopyranosyl units of the cellooligosyl chains are labeled with roman numerals, as shown in Fig. 2, b, for the cellotetraoside $T-4-4$ ⁴). However, as the NMR signals of the internal units overlap strongly, particularly in the octaoside derivatives, we desig-

Fig. 2. a) Numbering of the template-bound single- and double-chained cellodextrins, and labeling of the chains of T-x-x, and b) labeling of the glucopyranosyl units of template-bound cellodextrins

nate units I as a , the terminal units as c , and the sum of the internal units (units II and III of tetraosides and units II–VII of octaosides) as b . These designations of the glucosyl residues reflect regularities in the NMR spectra, as discussed below. To further simplify the notation, the O-atoms are numbered in the same way as C-atoms.

The template-bound cellooligosaccharides **T-x** and **T-x-x** ($x = 2, 4$, and 8) are yellowto-brownish solids that decompose upon melting. The double-chained T-x-x possess higher melting points than their mono-chained **T-x** analogues. The shorter **T-x** and **T-x-x** $(x=2 \text{ and } 4)$ are well soluble in DMSO, DMF, and N,N-dimethylacetamide $(>40 \text{ mmol/l})$, slightly soluble in H₂O ($<$ 5 mmol/l), and nearly insoluble in MeOH, EtOH, CF₃CH₂OH, AcOH, THF, dioxane, and CHCl₃. The solubility of T-x and T-x-x in $H₂O$ decreases with increasing number of sugar residues. The cellooctaosides **T-8** and T-8-8 are only soluble in DMSO. Hence, most of the NMR studies were conducted in (D_6) DMSO solution.

4.1. Investigation of the H-Bonding of T-x and T-x-x in DMSO Solution. The 1 H-NMR chemical shift of the OH groups $(\delta(OH))$, the vicinal coupling constant $(J(H, OH))$, and the temperature dependence of the OH signals $(\Delta \delta(OH)/\Delta T)$ are useful parameters for the investigation of H-bonding of alcohols and polyols in (D_6) DMSO [67] [68]. Fully solvated OH groups acting as H-donors in an intermolecular H-bond to (D_6) DMSO are characterised by a downfield shift of the OH signal, $J(H,OH)$ values of 4.5 – 5.5 Hz for equatorial and of 4.2 – 4.4 Hz for axial OH groups, and a strong temperature dependence $(|\Delta \delta (\text{OH})/\Delta T| > 4.5 \text{ pb/K}$). OH Groups acting as H-donors in an intramolecular H-bond are readily detected by an upfield shift of the OH signals, by $J(H,OH)$ values deviating from those of fully solvated OH groups, and by a weak temperature dependence $(|\Delta \delta(\text{OH})/\Delta T| < 3 \text{ pbb/K}$) [67] [68]. The $\delta(\text{OH})$ values for OH groups of monosaccharides and of the terminal units of oligosaccharides are a useful reference for the interpretation of $\delta(OH)$ values for OH groups of the internal units of oligosaccharides. The H-bonding of β -cellobiose and methyl β -cellobioside [69] in (D_6) DMSO has been analysed [68]. All OH groups, with the exception of $HO(3^1)$ are more or less fully solvated. A completely persistent inter-residue $O(3) - H \cdots O(5')$ H-bond of methyl β -cellobioside is evidenced by $J(3,\text{OH})=1.7$ Hz, $\delta(\text{HO}(3))=4.68$ ppm, and $\Delta\delta(HO(3))/\Delta T = -2.6$ ppb/K.

In the templated cellooligosaccharides $N-x$ and $N-x-x$ ($x=2, 3, 4,$ and 8) one or two glycosidically bonded cellodextrin chains are attached to naphthalene-1-ethanol and naphthalene-1,8-diethanol. The H-bonding of solutions of these oligosaccharides in (D_6) DMSO was analysed [4]. HO(3a) and HO(3b) (δ (OH)=4.59-4.73 ppm, $J(H, OH) < 2 Hz$, $\Delta\delta(OH)/\Delta T = -1.7$ to -2.2 ppb/K) are engaged in inter-residue Hbonds to $O(5')$ of the neighbouring glucosyl unit, whereas the other OH groups are involved in intermolecular H-bonds to the solvent $(J(2,OH)=J(3c,OH))$ $J(4c,OH) = 4.5 - 5.3$ Hz, $J(6,OH) = 5.4 - 6.5$ Hz, $\Delta\delta(OH)/\Delta T = -4.2$ to -7.2 ppb/K). $HO(2b)$ resonates at lowest field (5.37 ppm), followed by $HO(2c)$ (5.20 ppm), HO(2a) (5.13–5.14 ppm), HO(3c) (4.99 ppm), and HO(4c) (4.96 ppm), whereas HO(6b) resonates at $4.63-4.69$ and HO(6a) and HO(6c) at $4.52-4.61$ ppm. Only small shift differences ($\Delta\delta$ (OH) values) were observed for the OH groups of the internal units b ($\Delta\delta(OH) \leq 0.01$ ppm with the exception of 0.08 ppm for HO(3^{III}) of the cellotetraosides and $HO(3^{VII})$ of the cellooctaosides). Weak interchain H-bond interactions in $N-x-x$ were only observed for unit *a* closest to the template.

No concentration dependence of the $\delta(OH)$ and $\delta(H-C(1))$ values of **T-4-4** was observed in (D_6) DMSO solution at concentrations between 18 and 62 mmol (*Fig. 3*). Since all 1 H-NMR spectra of **T-x** and **T-x-x** were recorded at low concentration (10) mmol/l for **T-x** and 5 mmol/l for **T-x-x**), solute–solute interactions can be neglected.

All OH groups of the monoglucoside **T-1** and the diglucoside **T-1-1** are solvated in $(D₆)$ DMSO [3]. A weakly persistent flip-flop H-bond between the two primary OH groups of the diglucoside T-1-1 is suggested by the upfield shift of both OH groups $(0.04$ ppm for HO(6E) and 0.09 ppm for HO(6B) relative to HO(6) of **T-1**).

The unambiguous assignment of the OH signals of **T-x** and **T-x-x** ($\mathbf{x} = 2$, 4, and 8) in $(D₆)$ DMSO is based on the interpretation of DQFCOSY, HSQC, and TOCSY (only of T-4, T-4-4, and T-8-8) spectra and a comparison with the spectra of N-x and N-x-x (Table 1). The OH groups of T-x and T-x-x show similar chemical shifts as the corresponding OH groups of N-x and N-x-x except for $HO(2aE)$, $HO(2aB)$, and $HO(6a)$

Fig. 3. Concentration-dependence of the ¹H-NMR spectra of **T-4-4** in $(D_6)DMSO$: H–C(1) and OH signals

which are shifted downfield by 0.43, 0.54, and 0.10 ppm, respectively, due to the alkynyl substituent [70]. $HO(3a)$ and $HO(3b)$ of **T-x** and **T-x-x** are involved in an inter-residue H-bond to $O(5')$ as evidenced by the upfield shift $(4.75-4.65 \text{ ppm})$ and a small $J(3,\text{OH})$ value $(<1.5$ Hz). The other OH groups of **T-x** and **T-x-x** are engaged in intermolecular H-bonds to (D_6) DMSO. As it was already observed for N-4 and N-4-4, HO(3^{III}) of T-4 and **T-4-4** resonates downfield to $HO(3^{II})$ ($\Delta\delta$ = 0.06 – 0.07 ppm). A similar downfield shift is expected for $HO(3)^{VII}$ of **T-8** and **T-8-8**, but overlapping $HO(3)$ and $HO(6)$ signals prevent an exact assignment.

A comparison of $\delta(OH)$ of the ethynylated chain E of the di-cellosides T-x-x (x=2, 4, and 8) with $\delta(OH)$ of **T-x** ($x = 2, 4$, and 8; $\Delta\delta \le 0.04$ ppm; Table 1) shows a weak influence of the buta-1,3-diynylated chain of T-x-x on the chemical shift of the OH groups of the E chain. These small shift differences suggest at best very weak inter-chain H-bond interactions. The OH groups of the buta-1,3-divnylated chain B of T-x-x ($x = 2, 4$, and 8) possess similar chemical shifts as the OH groups of the ethynylated chain $E (\Delta\delta \le 0.03$ ppm) with the exception of $HO(2a)$ which shows the expected downfield shift due to the buta-1,3-diynyl substituent (0.21 vs. predicted 0.16 ppm [70]). Duplication of OH signals of T-x-x is only observed for those OH groups that are close to the template, i.e., OH groups of unit a and $HO-C(2^{II})$ (part of units b). This restricts the analysis by ¹H-NMR spectroscopy to inter-chain H-bond interactions of these OH groups of T-x-x.

HO(2*a*) of **T-x** ($\mathbf{x} = 2, 4$, and 8) resonates as broad *singlet* at 5.56–5.58 ppm, whereas $HO(2aE)$ and $HO(2aB)$ of T-x-x give rise to sharp *doublets* at 5.56–5.57 and 5.77–5.78 ppm, respectively (*Table 1*). That $HO(2a)$ of **T-x** and $HO(2aE)$ of **T-x-x** display the

	$T-2$ E chain	$T-4$ E chain	$T-8$ E chain
HO(2a) 5.58 ^a)		5.58a)	5.56a)
	$HO(3a)$ 4.82 (<1.5)	4.76 (< 1.5)	$4.74 - 4.66$
	$HO(6a)$ 4.71 (5.8)	4.68(5.1)	$4.74 - 4.66$
HO(2b)		5.42(5.0), 5.43(5.0)	5.41 (4.9), 5.39 (ca. 4.4; 6 H)
HO(3b)		4.75 ($<$ 1.5), 4.69 ($<$ 1.5)	$4.74 - 4.66$
HO(6b)		4.71(5.2)	$4.74 - 4.66$
	$HO(2c)$ 5.27 (4.7)	5.24(4.7)	5.21(4.8)
	$HO(3c)$ 5.05 (4.6)	5.04(3.8)	5.01(4.8)
	$HO(4c)$ 5.02 (4.7)	5.00(4.9)	4.98(5.3)
	$HO(6c)$ 4.64 (5.2)	4.61(5.1)	4.58(5.1)
	$T-2-2$	$T-4-4$	$T-8-8$
	E chain, B chain	E chain, B chain	E chain, B chain
	$HO(2a)$ 5.57 (5.5), 5.78 (5.8)	5.56 (5.7) , 5.77 (5.9)	5.56 (5.5) , 5.77 (5.8)
	HO(3 <i>a</i>) 4.80 (<1.5), 4.81 (<1.5) 4.74 (<1.5)		$4.74 - 4.65$
	$HO(6a)$ 4.71 (6.0)	$4.72 - 4.65$	$4.74 - 4.65$
HO(2b)		5.39 (2 OH; 4.9), 5.40 (4.8), 5.43 (5.0)	$5.43 - 5.38$
HO(3b)		$4.74 \left(\langle 1.5 \rangle, 4.67 \left(\langle 1.5 \rangle \right)$	$4.74 - 4.65$
HO(6b)		$4.72 - 4.65$	$4.74 - 4.65$
	$HO(2c)$ 5.25 (4.9), 5.28 (5.0)	5.22(4.9)	5.22(4.7)
	$HO(3c)$ 5.03 (4.6)	5.00(4.9)	5.01(4.4)
	$HO(4c)$ 5.00 (5.4), 5.01 (6.0)	4.98(5.4)	4.98(5.0)
	$HO(6c)$ 4.62 (5.5), 4.63 (5.5)	4.59(5.3)	4.58(5.2)

Table 1. ¹H-NMR $\delta(OH)$ [ppm] and J(H,OH) [Hz] Values (in parentheses) of **T-x** and **T-x-x** (**x** = 2, 4, 8) in (D_6) DMSO (assignments based on DQFCOSY and HSQC spectra and on TOCSY spectra of T-4, T-4-4, and T-8-8)

same chemical shift is not in agreement with an intramolecular H-bond of $HO(2a)$ of T- \bf{x} to $C(9')=O$, nor is such a H-bond expected, since this C=O group acts already as Hacceptor of HO(8') that resonates at the expected low field (12.46 – 12.50 ppm). A fast H/H exchange of $HO(2a)$ of T-x is responsible for the broadening of the signals; it is probably caused by the nearby phenolic $HO(8')$. The vicinal $J(H,OH)$ of the other OH groups of the E chain of **T-x-x** are similar to the corresponding coupling constants of T-x (ΔJ mostly smaller than 0.3 Hz). The values of $J(2b,OH)$ and $J(2c,OH)$ (4.7–5.0 Hz), and of $J(6b, OH)$ and $J(6c, OH)$ (5.1 – 5.2 Hz) of **T-x** and **T-x-x** do not hint at persistent intramolecular H-bonds. The larger $J(2a,OH)$ value of **T-x-x** ($\mathbf{x} = 2$, 4, and 8; 5.5 – 5.9 Hz) and $J(6a, OH)$ values of **T-2** and **T-2-2** (5.8 – 6.0 Hz) may point at weakly persistent intramolecular H-bonds. Since a small J(2,OH) value is expected for intraresidue $O(2)$ -H··· $O(3)$ or $O(2)$ -H··· $O(1)$ H-bonded glucopyranosyl species ($J < 2$) Hz [71]), these larger $J(2a,OH)$ values of T-x-x evidence either an electronic influence of the π -orbitals of the buta-1,3-diynyl substituent (*cf.* [72]) or weakly persistent interchain H-bonds of HO(2a). That both factors are operating is suggested by $J(2,OH)$ of the mono- and double-chained glucopyranoside analogues (5.5–5.6 vs. 6.0 Hz [3]).

SIMPLE ¹H-NMR experiments [73] with (D_6) DMSO solutions of **T-4, T-4-4, T-8**, and **T-8-8** did not show any splitting of OH signals upon titrating with D_2O or CD_3OD , corroborating the absence of strongly persistent intramolecular H-bonds between OH groups.

The temperature dependence of the OH signals of T-x and T-x-x ($x = 2, 4$, and 8) in (D_6) DMSO was determined from 298 to 348 K in 10-K intervals. $\Delta \delta$ (OH)/ ΔT values are listed in Table 2. Similarly as observed for $N-x$ and $N-x-x$, a weak temperature dependence of $\delta(HO(3a))$ and $\delta(HO(3b))$ values of **T-x** and **T-x-x** (**x**=2, 4; -2.0 to -2.3 ppb/K) confirms the inter-residue H-bond of HO(3a) and HO(3b) to O(5) of the adjacent glucopyranosyl moiety. The weak temperature dependence of the $\delta(HO(8'))$ value of T-x (-1.5 to -1.9 ppb/K) evidences an intramolecular H-bond of HO(8') to C(9')=O. $\Delta\delta$ (OH)/ ΔT Values in the range of -4.7 to -7.2 ppb/K for all other OH groups of T-x and T-x-x evidence more or less completely solvated OH groups. The $\Delta\Delta\delta(OH)/\Delta T$ values for corresponding OH groups of T-x and the E chain of T-x-x are small (≤ 0.4 ppb/K), except for HOC(2a) with $\Delta\Delta\delta(OH)/\Delta T$ values of 0.4–0.7 ppb/K). The slightly higher $|\Delta \delta(HO(2aE)/\Delta T)|$ values of **T-x-x** may suggest that $HO(2aE)$ acts as H-acceptor of a weakly persistent inter-chain H-bond.

		T-4	$T-8$	$T-2-2$ E chain, B chain	T-4-4 E chain, B chain	$T-8-8$ E chain, B chain
$HO-C(8')$	-1.9	-1.9	-1.9	$\overline{}$		
HO(2a)		$-7.2 -6.8$		-6.0 $-7.9, -6.6$	$-7.2, -6.2$	$-7.4, -6.4$
HO(3a)	-2.2	-2.3	b)	$-2.0, -2.3$	-2.2	$-2.4, -2.4$
HO(6a)	-5.6	-5.0	b)	-5.6	-5.1	b)
HO(2b)		$-5.8, -6.1$	-6.0	\equiv	-5.8 to -6.0	-5.8 to -6.2
HO(3b)		$-2.3, -2.2$	b)		-2.2	$^{\rm b})$
HO(6b)	Contract Contract	-4.7	b)	$\overline{}$	-5.0 to -5.2	b)
HO(2c)	-6.0	-5.7		$-5.9 -6.1, -6.2$	-5.7	-5.8
HO(3c)	-6.7	-6.3		$-5.8 -6.6, -6.2$	-6.3	-6.2
HO(4c)		$-6.0 -5.9$		$-5.0 -5.9, -6.2$	-5.7	-6.2
HO(6c)	$-5.1 -5.3$			$-5.3 -5.2, -5.5$	-5.2	-5.3

Table 2. Temperature Coefficients $\Delta\delta(OH)/\Delta T$ [ppb/K] of **T-x** and **T-x-x** in $(D_6)DMSO^a$)

H-Bonding of N-x and N-x-x ($x = 4$ and 8) was investigated by analysis of ROESY spectra $[4]^7$). The analysis was restricted to the cross-peaks between $H-C(1)$ and OH signals, as the signals for $H-C(2)$ to $H-C(6)$ overlapped. Cross-peaks between the HO(3ⁿ) and H-C(1ⁿ⁺¹) signals (n=1-3 for N-4 and N-4-4, and 1-7 for N-8 and $N-8-8$) confirme the inter-residue H-bond of $HO(3^n)$. Cross-peaks between the HO(2) and H-C(1) signals of the same unit evidence a weakly persistent $O(2ⁿ)$ - $H \cdots O(6^{n-1})$ H-bond. Inter-chain H-bonding of N-4-4 and N-8-8 is restricted to unit I;

 $7)$ See there for the differentiation between 'true' and relayed ROEs, such as TOCSY/ROE and exchange/ROE.

a weak inter-chain H-bond between the two $HO(2^1)$ groups is possible, but could not be unambiguously proven.

ROESY Spectra of T-4, T-4-4, T-8, and T-8-8 were recorded under identical conditions. Signal overlap prevents the assignment of the cross-peaks involving CH groups other than $H-C(1)$. Therefore, the analysis was again limited to interactions between $H-C(1)$ and OH groups, and between OH groups.

The partial ROESY spectrum of T-4 shows seven positive cross-peaks between 4.0–6.0 ppm (Fig. 4). All HO(2) signals show cross-peaks with the $H-C(1)$ signal of the same residue (peaks #1 to #4). These intra-residue interactions reveal weakly persistent inter-residue H-bonds of $HO(2^{IV})$ to $HOC(6^{III})$, $HO(2^{III})$ to $HOC(6^{II})$, and $HO(2^{II})$ to $HO(6^{I})$ – as also observed for **N-4** – and a weakly persistent bifurcated H-bond of $HO(2^1)$ to the acetylene and $C(9')=O$ groups (marked in blue in Fig. 5). Additional positive cross-peaks between the signals of $HO(3^{III})$ and $H-C(1^{IV})$ (peak #5), HO(3^{II}) and H-C(1^{III}) (peak #6), and HO(3^I) and H-C(1^{II}) (peak #7) confirm the inter-residue H-bonds of $HO(3)$ to $O(5)$ of the next residue (marked in red in Fig. 5).

Fig. 4. ROESY Spectrum of T-4: cross-peaks between $H-C(1)$ and OH signals

The partial ROESY spectrum of **T-4-4** between $4.0-6.0$ ppm is depicted in Fig. 6. $H - C(1^I), H - C(1^{II}), HO(2^I),$ and $HO(3^I)$ of the *B* chain resonate at a position that differs from the one for the corresponding H-atoms of the E chain. The unambiguous assignment of the signals of T-4-4 is based on the stronger downfield shift of $HO(2^1)$ by the buta-1,3-diynyl group [70], and on the analysis of a DQFCOSY and a TOCSY spectrum. The ROESY spectrum of T-4-4 shows cross-peaks analogous to those observed for **T-4** (peaks $#1 - #7$); they evidence the completely persistent inter-residue

Fig. 5. Interpretation of the ROESY spectrum of T-4; arrows indicate ROE cross-peaks and hashed lines H-bonds

 $O(3)$ –H··· $O(5)$ H-bonds and weakly persistent inter-residue H-bonds $O(2)$ –H··· $O(6)$ (tautomer A in Fig. 7). The cross-peak pairs $#4B/#4E$ and $#7B/#7E$ reveal only intrachain contacts; there are no cross-peaks for the corresponding inter-chain interactions. An additional cross-peak (#8) is observed between the signals of $H-C(1^V)$ and one HO(6) signal of the *multiplet* for HO(6^{I-III}). For geometric reasons, it is best assigned to an intra-chain inter-residue close contact between $H-C(1^V)$ and $HO(6^{III})$. Analogous close contacts between $H-C(1^{III})$ and $HO(6^{II})$, and between $H-C(1^{II})$ and $HO(6^I)$ may contribute to the cross-peaks #6 and #7. They evidence weakly persistent inter-residue H-bonds from HO(6) to HO(2) (interactions and H-bonds in green in tautomer \mathbf{B} ; Fig. 7). It is noteworthy that the cross-peaks between $H-C(1)$ and OH signals reveal inter-residue flip-flop H-bonds between $HO(6)$ and $HO(2)$ of the double-

Fig. 6. ROESY Spectrum of T-4-4: cross-peaks between $H-C(1)$ and OH signals

Fig. 7. Interpretation of the ROESY spectrum of **T-4-4**; arrows indicate ROE cross-peaks and hashed lines H-bonds

chained T-4-4 and N-4-4, but not of the mono-chained T-4 and N-4 (only cross-peaks for unidirected $O(2') - H \cdots O(6)H$ H-bonds).

There are several positive cross-peaks between OH groups (peaks #9 to #14) in the ROESY spectrum of T-4-4, whereas the ROESY spectrum of N-4-4 showed only a single positive cross-peak between $HO(2^{III})$ and $HO(3^{III})$. The inter-residue H-bonds $\overrightarrow{O(6)}$ -H··· $O(2')$ of tautomer **B** of **T-4-4** suggest a close intra-residue contact of $HO(2)$ and $HO(3)$. This is confirmed by cross-peaks between the signals of $HO(2^{\text{IV}})$ and $HO(3^{IV})$ (#9), $HO(2^{III})$ and $HO(3^{III})$ (#10), $HO(2^{III})$ and $HO(3^{II})$ (#11E), $HO(2^{IIB})$ and $HO(3^{II})$ (#11B), $HO(2^{I E})$ and $HO(3^{I})$ (#12E), and finally $HO(2^{I B})$ and $HO(3¹)$ (#12B); they evidence an intra-residue close contact between HO(2) and $HO(3)$ also for units I (marked with pink arrows in tautomer **B**). There are two additional positive cross-peaks between the broad *singlet* of $HO(3^I)/HO(3^{III})$ and, on the one hand, the *triplet* of $HO(6^{IV})$ (#13) and, on the other hand, the *doublet* of $HO(2^{\text{IV}})$ (#14). Cross-peak #13 may be assigned to a close intra-chain contact between $HO(6^{IV})$ (gg conformation for $HOC(6^{IV})$) and $HO(3^{III})$, whereas cross-peak #14 cannot be assigned to an intra-chain contact (MM3^{*} modeling predicts a H ··· H distance of 4.75 Å distance between $HO(2^{IV})$ and $HO(3^{III})$). A close inter-chain contact is possible between $HO(2^{IVE})$ and $HO(3^{IIIB})$ but considered improbable since there are no additional cross-peaks suggesting close contacts between OH groups of the other units, especially of unit I.

Fig. 8. ROESY Spectrum of T-8: cross-peaks between $H-C(1)$ and OH signals (analogous numbering of cross-peaks as in T-4-4)

The ROESY spectrum of **T-8** (Fig. 8) shows similar cross-peaks as the ROESY spectrum of **T-4-4** for the E chain, confirming the inter-residue H-bond $O(3)$ $H \cdot \cdot \cdot O(5')$ and the inter-residue flip-flop H-bond between $HO(2)$ and $HO(6')$. The ROESY spectrum of **T-8-8** (*Fig.* 9) shows the same cross-peaks as the ROESY spectrum of **T-4-4** if one assumes that the cross-peaks $#6, #7, #9$, and $#14$ are hidden by strong negative absorptions. Hence, we expect that T-4-4 and T-8-8 adopt in $(D₆)$ DMSO conformations which avoid inter-chain H-bond interactions. This is feasible in a V-shape arrangement of the cellooligosyl chains, as illustrated by the MM3*-calculated $O(6)-H \cdots O(2')$ H-bonded tautomer of **T-8-8** in Fig. 10 (considered a model of cellulose almost completely dissolved in DMSO). This structure suggests at best an inter-chain H-bond between $HO(2^{1B})$ and $HO(6^{1E})$. However, there are no crosspeaks between these OH signals in the ROESY spectra of T-4-4 and T-8-8.

4.2. Comparison of the CP/MAS¹³C-NMR Spectra of **T-x** and **T-x-x** ($x=1, 2, 4$, and 8) with Those of Cellulose I_a , I_β , and II. The solid-state CP/MAS ¹³C-NMR chemical shifts of **T-x** and **T-x-x** ($x = 1, 2, 4$, and 8) are listed in *Table 3*⁸). The assignment of the signals is based on a comparison with the data in $D₂O$ solution (especially of the glucosides **T-1** and **T-1-1**), and with the data of N -x and N -x-x in the solid state. The

⁸⁾ We thank Prof. Dr. Beat H. Meier, Ashwin Vorhofen, and Matthias Ernst, Laboratory of Physical Chemistry, ETH Zurich, for the CP/MAS 13C-NMR spectra of T-x, T-x-x, A-x, and A-x-x.

Fig. 9. ROESY Spectrum of T-8-8: cross-peaks between $H-C(1)$ and OH signals (analogous numbering of cross-peaks as in T-4-4)

intensity of the anthraquinonyl and ethynyl signals decreases with increasing saccharide chain length; the spectra of T-8 and T-8-8 essentially show only signals of the central units b.

The solid-state CP/MAS¹³C-NMR spectra of the mono-chained T-x ($x=1, 2, 4$, and 8) are well-resolved (*Fig. 11*). The spectrum of the glucoside **T-1** shows separate peaks for all C-atoms of the glucopyranosyl-ethynyl moiety; the δ values are very similar to those of the solution spectrum in (D_6) DMSO [3] ($\Delta \delta \leq 2.0$ ppm, with the exception of 4.6 ppm for C \equiv C \sim C (1)). Noteworthy are the downfield shift for C (5) resonating at 79.0 ppm, the upfield shift for C(1) appearing at 70.6 ppm, and signals for the ethynyl moiety at 94.1 and 88.6 ppm. The signals for the ethynyl group and for $C(4c)$ of **T-2** and **T-4** are unambiguously assigned by a comparison with the corresponding values of T-1 and by the decrease of their intensity in **T-4**. The duplication of the signals for $C(1c)$ and $C(4a)$ of **T2** (107.8/106.4 and 84.6/83.8 ppm, resp.; Table 3) hints at the presence of two different molecules in the unit cell. One signal is observed for each $C(6a)$ and $C(6c)$ (62.1/ 61.1 ppm). The spectrum of **T-4** exhibits the expected three peaks for $C(1b)$ and $C(1c)$ (109.3, 108.0, and 106.3 ppm), but in a ratio of ca. 3:2:1. Similarly, $C(4a)$ and $C(4b)$ of **T-4** give rise to two peaks at 82.4 and 83.0 ppm in a ratio larger than 3:1,

Fig. 10. MM3*-Modeled, V-shaped conformer of T-8-8 possessing inter-residue O(6)-H--O(2) Hbonds (H-C atoms are omitted for enhanced clarity)

HELVETICA CHIMICA ACTA – Vol. 89 (2006) 697

whereas $C(6a)$, $C(6b)$, and $C(6c)$ resonate as a single peak at 61.5 ppm with a weak shoulder at 61.0 ppm. The ratios for the $C(1)$ and $C(4)$ signals again suggest two different molecules in the unit cell. There is no downfield shift for $C(4a)$ and $C(6)$ of **T-2** and **T-4**, and of $C(4b)$ of **T-4**; this clearly evidences that the cellobioside **T-2** and the cellotetraoside T-4 are not H-bonding models for cellulose II. At best, they mimic cellulose chains at the surface of the crystallites (see *Introduction*). In contradistinction, the striking similarity of the solid-state CP/MAS¹³C-NMR spectrum of **T-8** and the one of cellulose II, especially the downfield shift of $C(4)$ and $C(6)$ at 89.9/88.7 and 63.8 ppm, respectively, reveals that the cellooctaoside T-8 is an excellent mimic of cellulose II in the interior of the crystallite.

The solid-state CP/MAS ¹³C-NMR spectrum of the C_1 -symmetric diglucoside **T-1-1** is well-resolved (Fig. 12 and Table 3). The assignment of the peaks of **T-1-1** is based on a comparison with its solution spectrum in (D_6) DMSO [3] and the solid-state spectrum of T-1. The solid-state CP/MAS¹³C-NMR spectra of T-2-2 and T-4-4 show only broad signals, probably due to a low degree of crystallinity. According to the upfield shift of $C(4b)$ and $C(4c)$ (<82 ppm), **T-2-2** and **T-4-4** do not mimic the H-bonds of cellulose I. The solid-state CP/MAS 13C-NMR spectrum of T-8-8 is much better resolved and

	$C(1)$ of units c and <i>b</i>	$C(4)$ of units a and b^a)	$C(2)$, $C(3)$, $C(5)$, $C(1)$ of unit a, and $C(4)$ of unit c	C(6)
T-1			79.0, 76.4, 72.5, 70.6, 68.7	61.5
$T-2$	107.8, 106.4	84.6, 83.8	78.5, 77.6, 77.0, 75.6, 74.3, 72.6, 68.8	62.1, 61.1
$T-4$	109.3, 108.0, 106.3	83.0, 82.4	78.5, 77.7, 76.4, 74.4, 72.8, 68.7	61.5
$T-8$	108.3, 106.2	89.9, 88.7	77.7, 76.0, 73.7	63.8
Cellulose II	108.3, 106.2	89.9, 88.7	77.8, 75.9, 73.8	64.2, 63.6
$T-1-1$			79-77, 74.2, 72.9, 71.9, 67.8	59.9, 59.1
$T-2-2$	104.2 (br.)	84.0 (br.)	76.72 (br.), 72.0 (br.)	62.7 (br.)
$T-4-4$	103.9 (br.)	83.5 (br.)	75.0 (br.)	62.8 (br.)
$T-8-8$	107.0, 105.7	89.9	76.1, 73.0	66.3, 64.3
Cellulose I_β	107.0, 105.2	90.1, 89.3	76.1, 75.3, 73.7, 72.5	66.8, 66.1
Cellulose I_a	106.3	91.0, 90.2	75.8, 73.8, 73.0, 72.0	66.5
	$C(9)$ and $C(10)$ of the template	$C(8)$ of the template of $T-x$	other signals of the template	$C\equiv C$ and $C\equiv C-C\equiv C$
T-1	185.9, 178.3	163.2	140.9, 116.0	94.1, 88.6
$T-2$	187.0, 179.1	160.1	144.0, 116.0	92.7, 85.6
$T-4$	186.9, 178.9	160.0	143.4, 116.0	92.3, 86.1
$T-1-1$	184.3, 180.8		143.6, 122.0	94.92, 86.1, 82.0, 80.7, 67.8
$T-2-2$	180.1 (br.)		141.3, 121.8	92.7 (br.)
$T-4-4$			143.0, 122.0	93.9 (br.)

Table 3. CP/MAS ¹³C-NMR Chemical Shifts [ppm] of the Mono- and Bis-C-glucoside T-1 and T-1-1, the Mono- and Bis-C-Cellodextrins T-x and T-x-x (x=2, 4, and 8), and the Cellulose Polymorphs I_{ω} , I_{β} [19], and II [6]

resembles that of cellulose I_{β} , especially with regard to the two peaks for C(1) at 107.0 and 105.7 ppm. A broad peak for $C(4)$ of **T-8-8** at 89.9 ppm replaces the double-headed signal for C(4) of cellulose I_β at 90.1/89.3 ppm. We conclude that **T-8-8**, unlike N-8-8, is indeed a H-bond mimic of cellulose I. While a comparison of T-8-8 with N-8-8 strongly suggests that implementation of the phase shift is crucial to mimic cellulose I, it is not yet clear which one of the three factors – the rigid linkers, the optimised distance between the chains, and the phase shift between the chains – is mostly responsible to render **T-8-8** a good mimic of cellulose I_{β} .

4.3. Comparison of the X-Ray Powder-Diffraction Spectra of T-8 and T-8-8 with Those of Celluloses. As several attempts to grow single crystals of **T-8 and T-8-8** suitable for X-ray diffraction failed, we analysed these compounds by X-ray powder diffraction and compared the spectra with those of celluloses [5] (*Fig. 13*). The powder-diffraction spectra of **T-8-8** and **T-8**⁹) resemble strongly that of cellulose I and cellulose II, respectively, confirming that T-8-8 is an oligomeric mimic of cellulose I and T-8 of cellulose II.

⁹) We thank Prof. Dr. Reinhard Nesper and Dr. Qinxing Xie, Laboratory of Inorganic Chemistry, ETH Zurich, for the powder diffraction spectra of T-8 and T-8-8.

Fig. 11. CP/MAS ¹³C-NMR Spectra of **T-x** ($\mathbf{x} = 1, 2, 4$, and 8) and cellulose II (from [6])

5. Comparison of the CP/MAS¹³C-NMR Spectra of **A-x** and **A-x-x** with Those of Cellulose Triacetates I and II. Cellulose triacetates I (CTA I) and II (CTA II) are prepared by acetylating native cellulose I and mercerised cellulose II, respectively, and maintain the polymorphic character of cellulose [74]. Similar to cellulose I and II, CTA I can be transformed into CTA II, while the reverse process is unknown. Remarkably, both CTA I and CTA II were obtained from native celluloses by homogeneous or heterogeneous acetylation, respectively, while the same conditions transformed cellulose II only into CTA II. Based on the X-ray fiber-diffraction patterns and stereochemical model analysis, a parallel chain orientation was proposed for CTA I [75] and an antiparallel one for **CTA II** [76]. The unit cell of **CTA I** (orthorhombic, $P2₁$) contains two parallel chains, whereas the unit cell of **CTA II** (orthorhombic, $P2_12_12_1$) contains

Fig. 12. CP/MAS¹³C-NMR Spectra of T-x-x (x=1, 2, 4, and 8), and cellulose I_a and I_β (from [6])

two parallel and two antiparallel chains. However, there are experimental results in conflict with Sprague's suggested analogy between celluloses and CTAs (see the excellent discussion in [77]) casting some doubt on the unambiguous assignment of the polarity of the chains in CTA I and CTA II.

Fig. 13. X-Ray powder-diffraction patterns of T-8, T-8-8, cellulose I to IV, and amorphous cellulose

The CP/MAS¹³C-NMR spectra of **CTA I** and **CTA II** show characteristic differences [78]. Analysis of 13C-enriched samples [79] led to a complete assignment of the signals for $C(2)$ to $C(5)$. The spectrum of **CTA I** displays a single set of signals, whereas **CTA II** shows a double set of signals for $C(2)$ to $C(6)$ evidencing two different 2,3,6tri-O-acetyl- β -D-glucopyranosyl units [77] (*Fig. 14*). C(1) of **CTA I** resonates at 102.4 ppm, $C(1)$ of **CTA II** at 100.4 ppm, $C(6)$ of **CTA I** at 62.5 ppm, and $C(6)$ of **CTA II** at 65.2 and 67.3 ppm (*Table 4*). The different values for $C(6)$ may indicate a different orientation of the (acetoxy)methyl group, or the effect of the different environment. If Horii's rule [80] should be applicable to acetylated glucopyranosides, then CTA I addopts the gg , and **CTA II** the tg and gt conformation.

Fig. 14. CP/MAS¹³C-NMR Spectra of $A-x$, $A-x-x$ $(x=2, 4, and 8)$, CTA I, and CTA II (from [78])

	$C(1)$ of units c C \equiv C and and b	$C=C-C\equiv C$	$C(4)$ of units a and b	$C(2)$, $C(3)$, $C(5)$, $C(1)$ of unit <i>a</i> , and $C(4)$ of unit <i>c</i>	C(6)
$A-2$ $A-4$	103.0 103.1 (2 C),	95.7, 83.6 94.8, 85.0	79.6, 77.6 79.9 (2 C),	73.3, 72.0, 70.4, 69.6 75.4, 73.8, 72.5, 70.7, 69.1	63.8, 62.9 63.9, 62.1
$A-8$	100.1 103.4 (br.)		77.6 81.2 (br.)	76.9 (br.)	(3 C) 69.9, 66.9
$A-2-2$	101.3	92.6, 85.6, 84.3, 81.9	78.0 (br.)	73.1, 70.6, 69.9, 68.0	64.8, 61.3
$A-4-4$	103.0 $(2 C)$, 99.8	92.2	80.0 (2 C), 78.2	75.5, 73.8, 72.5, 72.0, 70.7, 69.1, 68.3, 67.4	65.3, 62.4 (3 C)
$A-8-8$	103.4 (br.)		81.6 (br.)	76.7 (br.)	69.6, 67.0
CTAI	102.4		80.3	76.2 (2 C), 72.9 ^a)	62.5
CTAII	100.4 (2 C)		80.1, 78.7	76.6, 75.8 (2 C), 74.8, 73.5, $72.5^{\rm a}$)	67.3, 65.2
	$C = O$ of Ac	Me of Ac	the template	$C(9)$ and $C(10)$ of $C(8)$ of the template of $\bf{A-x}$ Other signals of	the template
$A-2$	171.6, 171.1, 170.5, 169.9	22.4, 21.9, 20.9, 19.9	185.4, 181.3	162.4	139.0, 115.0
$A-4$	172.0, 170.5, 169.8	21.7, 20.4	185.7, 180.6	163.3	139.0, 115.0
$A-8$	173.5	24.3			137.7
$A-2-2$	170.5	22.0, 20.6	$182.6, 182.0, 179.9 -$		141.0, 119.0
$A-4-4$	172.5, 172.0, 170.8, 169.9	21.8, 20.5	181.5		141.0, 119.0
$A-8-8$	173.6	24.2			
	171.9, 169.4	22.0, 20.9			
CTAI					

Table 4. CP/MAS¹³C-NMR Chemical Shifts [ppm] of A -x, A -x-x (x = 2, 4, and 8), CTA I [79], and CTA II [79]

Surprisingly, according to the studies of the CP/MAS ¹³C-NMR and X-ray diffraction spectra of a series of peracetylated cello-oligomers by VanderHart et al. [78] and Kono et al. [64] [81], the X-ray diffraction spectra of peracetylated cellopentaose and higher cello-oligosides are similar to those of **CTA I**. In view of these results, we considered that analysis of the X-ray diffraction spectra of $\bf{A-x}$ and $\bf{A-x-x}$ ($\bf{x} = 2, 4$, and 8) may contribute to understand the (apparent) opposite chain orientation in the solid state of free and acetylated higher cellodextrins.

The CP/MAS¹³C-NMR spectra of the template-bound peracetylated **A-x** and **A-x-x** $(x=2, 4, and 8)$, and of **CTA I** and **CTA II** are depicted in Fig. 14, and their chemical shift values are compiled in Table 4. Surprisingly, the spectra of the mono-chained Ax (especially of the tetraoside and octaoside) resemble the corresponding spectra of the double-chained A -x-x. $C(1)$ of the tetraosides $A-4$ and $A-4-4$ resonates as two peaks in a $2:1$ ratio at $103.4 - 103.1$ and $101.3 - 99.8$ ppm; the former peak is assigned to $C(1b)$ and the latter one to $C(1c)$. $C(6)$ of **A-4** and **A-4-4** resonates as two peaks in a 1:3 ratio at $65.3 - 63.9$ and $62.4 - 62.1$ ppm; the weaker peak was assigned to $C(6c)$, and the stronger one to $C(6a)$ and $C(6b)$. With the exception of the signals for unit c that find no equivalent in the spectra of **CTA I** and **CTA II**, the spectra of A -4 and A-4-4 resemble the spectrum of CTA I. The (less well resolved) spectra of the cellooctaosides A-8 and A-8-8, however, resemble strongly the spectrum of CTA II, particularly when considering the two peaks for $C(6)$, the broad signals for $C(4)$ and for $C(2)/C(3)/C(5)$, and the single peak for the Me groups. The similarity between the solid-state CP/MAS 13 C-NMR spectra of A-4 and A-4-4 and the one of CTA I, and of A-8 and A-8-8 with the one of CTA II is opposite to the observations in the cellodextrin series. At first view, this result is surprising especially for A-8-8 which possesses parallel cellosyl chains. However, the antiparallel orientation of two molecules of A-8-8 leads to two parallel and two antiparallel chains, as required for CTA II. Any further analysis must resolve the question about the origin of the signal-doubling in the solid-state 13 C-NMR spectrum of **CTA II**. Is it due to different glucopyranosyl units in the repeating cellobiosyl moiety of all chains, to different parallel chains, or to different antiparallel chains? Better resolved diffraction analyses of CTA II and its oligomeric models should allow to answer this fundamental question.

6. Conclusions. Solid-state CP/MAS 13C-NMR spectroscopy and X-ray powder-diffraction analysis reveal that **T-8** and **T-8-8** are oligomeric mimics of cellulose II and I_{β} , respectively. A combined analysis of the ¹H-NMR data (δ (OH), J(H,OH), $\Delta\delta$ (OH)/ ΔT , ROESY spectra) of the mono- and double-chained template-bound cellosides **T-4, T-8, T-4-4,** and **T-8-8** in (D_6) DMSO solution reveals strong intra-strand inter-residue C(3)OH ··· O(5') H-bonds, weakly persistent inter-residue flip-flop H-bonds between HO(2) and HO(6'), and at best a weakly persistent inter-strand H-bond between $HO(2^{I})$ of the ethynylated chain E and $HO(6^{I})$ of the buta-1,3-diynylated chain B of **T-4-4** and **T-8-8**. The solid-state CP/MAS ¹³C-NMR spectra of \mathbf{A} -4 and \mathbf{A} -4-4 resemble that of CTA I, and those of A-8 and A-8-8 resemble that of CTA II, opposite to the findings in the cellodextrin series.

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Experimental Part

General. See [3]. Lewis acids such as AgOTf, CdCO₃, HgCl₂, Hg(CN)₂, and BF₃ \cdot OEt₂ were used directly without purification. IR Spectra (ca. 3% soln. in CHCl₃ or CH₂Cl₂ soln. or 1% KBr pills): Per kin -Elmer 298 and 1600 FT-IR spectrometer. NMR Spectra: Varian XL-300 (1 H: 300 MHz, 13 C: 75 MHz), or a Bruker AMX-500 or -600 (500 or 600 MHz, resp.) in deuteriated solvents (CDCl₃ and $(D₆)$ DMSO, from Dr. Glaser AG, Basel); chemical shifts δ in ppm and couplings constants J in Hz. In ambiguous cases, ¹ H-assignment is based on selective homonuclear decoupling experiments and 2D experiments. The CP/MAS ¹³C-NMR experiments were performed on a *Bruker* NMR spectrometer with 500-MHz resonance frequency for ${}^{1}H$ and 125 MHz for ${}^{13}C$. The spectrometer is equipped with a 2.5-mm MAS double resonance probe from *Chemagnetics* (Ft. Ciollins, Colorado, USA). The sample spinning frequency was 20 kHz for all samples. The contact time for cross-polarisation is 3 ms and the data-acquisition time is 25 ms. The radio-frequency field strengths are 100 and 84 kHz for the 1 H and the ¹³C channel, resp. The numbers of scans collected vary from 10000 to 20000 with a repetition of 8 s for all samples.

Allyl $4,6$ -O-Benzylidene-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 4)-2,3,6-tris-O-(4chlorobenzyl)- β -D-glucopyranoside (7). A soln. of 5 [36] (36 g, 76.5 mmol) and 4-ClC₆H₄CH₂Cl (93.7 g, 0.58 mol) in DMF (324 ml) was added dropwise to a cooled (0 $^{\circ}$), stirred suspension of NaH (55-65% in oil, 27.8 g) in dry DMF (396 ml). The mixture was gradually warmed to 25° , stirred overnight, cooled to 0°, treated dropwise with dry MeOH (30 ml), and poured into diluted AcOH (36 ml of AcOH in 6.0 l of H₂O). After stirring for 1 h at $0-5^\circ$, the precipitate was filtered off and washed with hexane. A soln. of the crude solid (110 g) in CH₂Cl₂ (600 ml) was washed with brine, dried (Na₂SO₄), and filtered through Celite. The filtrate was diluted with MeOH (2.5 l). The precipitate was filtered off, washed with MeOH and Et₂O, and dried overnight in high vacuum affording pure 7 (69.1 g, 83%). R_f (AcOEt/hexane 1:4) 0.40. M.p. 178.2°. $[\alpha]_D^{25} = +2.7$ (c=1.0, CHCl₃). IR (CHCl₃): 3008w, 2869w, 1600w, 1492m, 1408w, 1361w, 1308w, 1277w, 1162w, 1092s, 1015m, 931w, 839w. ¹H-NMR (300 MHz, CDCl₃): see Table 5; additionally, 7.50 – 7.10 (m, 25 arom. H); 5.93 (dddd, J = 17.1, 10.6, 5.9, 5.3, CH=CH₂); 5.49 (s, PhCH); 5.32 $(dq, J=17.1, 1.6)$, 5.21 $(dq, J=10.6, 1.6)$ (CH=CH₂); 4.86 $(d, J=11.8)$, 4.85 $(d, J=11.2)$, 4.67 $(d, J=12.1)$

Table 5. Selected ¹H-NMR Chemical-Shift Values [ppm] and Coupling Constants [Hz] of the 4-Chlorobenzylated Cellobiose Derivatives $7, 8, 11, 13, 14, 17,$ and 18 in CDCl₃

	7	8	11	$13a$)	14	$17a$)	18
$C\equiv CH$					2.52		2.22
$H - C(1^1)$	4.40	4.40		3.99	3.98	4.04	4.01
$H - C(2^{I})$	3.39	3.385	4.07	3.50	3.53	3.50	3.51
$H - C(3I)$	3.48	3.49	3.92	3.43	3.44	3.45	3.44
$H - C(4^1)$	3.91	3.91	4.16	3.93	3.93	3.93	3.92
$H - C(5^1)$	$3.33 - 3.25$	$3.34 - 3.20$	$4.52 - 4.47$	$3.26 - 3.21$	$3.28 - 3.21$	$3.28 - 3.24$	$3.28 - 3.22$
$H_a-C(6^I)$	3.76	3.76	3.69	3.75	3.76	3.77	3.75
$Hb-C(6I)$	3.63	3.65	3.61	3.63	3.63	3.63	3.61
$H - C(1H)$	4.48	4.42	4.48	4.39	4.39	4.41	4.39
$H - C(2H)$	$3.33 - 3.25$	3.23	3.34	3.20	3.21	3.22	3.21
$H - C(3H)$	$3.59 - 3.52$	3.325	$3.64 - 3.57$	3.28	3.29	3.31	3.29
$H - C(4H)$	$3.59 - 3.52$	3.63	$3.64 - 3.57$	3.60	3.61	3.62	3.59
$H - C(5H)$	$3.24 - 3.14$	$3.34 - 3.20$	$3.32 - 3.24$	$3.26 - 3.21$	$3.28 - 3.21$	$3.28 - 3.24$	$3.28 - 3.22$
$H_a-C(6H)$	4.21	3.56	4.22	3.56	3.55	3.58	3.57
$H_b-C(6H)$	3.46	3.43	3.67	3.42	3.42	3.45	3.43
$HO-C(4H)$	\overline{a}	3.01		2.95	2.95	3.00	2.99
$J(1^I,\equiv CH)$	$\overline{}$	$\overline{}$	$\overline{}$	$\overline{}$	2.2	$\overline{}$	0.6
$J(1^{\rm I},2^{\rm I})$	7.8	7.8	$\overline{}$	9.6	9.7	9.4	9.3
$J(2^I,3^I)$	9.0	9.0	5.8	9.1	9.0	9.0	9.0
$J(3^I,4^I)$	9.2	9.0	5.0	8.9	8.7	8.9	8.7
$J(4^{\rm I},5^{\rm I})$	9.2	9.6	8.1	9.7	> 9.3	9.7	9.7
$J(5^{\rm I},6{\rm a}^{\rm I})$	4.0	4.0	3.8	3.4	3.3	3.5	3.3
$J(5^{\rm I},6{\rm b}^{\rm I})$	2.3	2.5	2.8	1.6	1.9	1.6	1.6
$J(6a^{\rm I}, 6b^{\rm I})$	10.9	10.9	11.2	11.0	11.1	11.2	11.2
$J(1^{\rm II},2^{\rm II})$	7.8	7.8	7.8	7.9	7.9	7.8	7.8
$J(2^{\text{II}}, 3^{\text{II}})$	b	9.0	8.8	9.1	9.0	9.0	9.0
$J(3^{\text{II}},4^{\text{II}})$	b	9.5	b)	8.9	8.7	8.8	9.0
$J(4^{\text{II}},5^{\text{II}})$	\mathbf{b}	9.9	$^{b})$	9.1	8.9	9.0	9.0
$J(5^{\text{II}},6\text{a}^{\text{II}})$	5.0	4.7	4.7	5.8	5.9	5.8	6.2
$J(5^{\text{II}},6b^{\text{II}})$	10.6	5.9	10.6	4.8	3.4	4.8	4.7
$J(6a^{\text{II}}, 6b^{\text{II}})$	10.6	9.7	10.6	10.0	10.0	9.9	10.0
$J(4^{\text{II}}\text{,OH})$		1.9		2.0	1.9	1.8	2.2

(3 PhCH); 4.66 (d, J=12.4, 2 PhCH); 4.65 (d, J=11.8), 4.62 (d, J=11.5), 4.52 (d, J=12.5) (3 PhCH); 4.36 – 4.44 (m, 1 allyl. H); 4.36 (d, J = 12.4, 2 PhCH); 4.12 (ddt, J = 13.1, 5.9, 1.4, 1 allyl. H). ¹³C-NMR (75 MHz, CDCl₃): see *Table 6*; additionally, 137.38, 137.07, 136.88, 136.83, 136.54, 136.38 (6s); 133.84 $(d, CH=CH₂)$; 133.46, 133.45, 133.37, 133.34, 133.11 (5s); 128.24 – 129.29 (several d); 125.99 (2d); 117.35 (t, CH=CH₂); 101.20 (d, PhCH); 74.54, 74.41, 74.14, 73.98, 72.52 (5t, 5 PhCH₂); 70.21 (t, CH₂= CHCH₂). HR-MALDI-MS: 1113.2074 (57, $[M + Na]^+$, C₅₇H₅₅Cl₅NaO⁺₁; calc. 1113.2085). Anal. calc. for $C_{57}H_{55}Cl_5O_{11}$ (1093.32): C 62.62, H 5.07; found: C 62.66, H 5.22.

Allyl 6-O-Benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 4)-2,3,6-tris-O-(4-chloroben zyl - β -D-glucopyranoside (8). A suspension of 7 (16 g, 14.6 mmol) and NaBH₃CN (10.12 g, 161 mmol) in dry THF (240 ml) was cooled to 0° , treated portionwise with sat. HCl in Et₂O (160 ml), and stirred for 4 h. After dilution with CH_2Cl_2 , the yellow suspension was filtered through a pad of silica gel. The filtrate was washed with H₂O, sat. NaHCO₃ soln., and brine, dried (Na_3SO_4) , and evaporated. Crystallisation from CH₂Cl₂/hexane and FC of the mother liquor (toluene/acetone 20:1) gave 8 (11.0 g, 69%). R_f $(ACOE$ t/hexane 1:4) 0.08. M.p. 102–103.4°. $[a]_D^{25} = +14.9$ $(c=1.0, CHCl_3)$. ¹H-NMR (300 MHz, CDCl₃): see Table 5; additionally, 7.40 – 7.10 (m, 25 arom. H); 5.94 (dddd, J = 17.2, 10.3, 5.9, 5.3, CH=CH₂); 5.33 $(dq, J=17.2, 1.6), 5.24 (dq, J=10.3, 1.6)$ (CH=CH₂); 4.885 (d, J=11.2), 4.85 (d, J \approx 11.2), 4.835 (d, $J=11.8$), 4.75 (d, $J=11.8$), 4.70 (d, $J=11.8$), 4.64 (d, $J=11.5$) (6 PhCH); 4.63 (br. d, $J\approx 11.2$, 2 PhCH); 4.55 (d, J = 12.1, PhCH); 4.435 (br. s, PhCH₂); 4.43 - 4.36 (m, 1 allyl. H); 4.39 (d, J = 12.1, PhCH); 4.12 (ddt, J = 13.2, 5.9, 1.4, 1 allyl. H). Anal. calc. for $C_{57}H_{57}Cl_5O_{11}$ (1095.33): C 62.50, H 5.25; found: C 62.38, H 5.34.

Allyl 6-O-Benzyl-2,3-bis-O-(4-chlorobenzyl)-4-O-(methoxymethyl)- β -D-glucopyranosyl-(1 \rightarrow 4)-2,3, 6-tris-O-(4-chlorobenzyl)- β -D-glucopyranoside (9). A soln. of 8 (1.2 g, 1.09 mmol) in dry DMF (30 ml) was cooled to 0° , treated with 55% NaH in oil (94 mg, 2.2 mmol) and MOMCl (176 μ l, 2.2 mmol), warmed to r.t., and stirred for 4 h. The mixture was poured into H_2O . After extraction with CH_2Cl_2 , the org. layer was washed with H₂O and brine, dried (Na_5SO_4) , and evaporated. Crystallisation from CH₂Cl₂/hexane and FC of the mother liquor (toluene/acetone 25:1) gave 9 (880 mg, 70%) and 8 (200 mg, 16%). R_f (AcOEt/hexane 1:4) 0.28. M.p. 102.0–104.1°.

4,6-O-Benzylidene-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 4)-2,3,6-tris-O-(4-chloro $benzyl$ -D-glucopyranose (10). At 25°, a suspension of bis(methyl(diphenyl)phosphine)(cycloocta-1,5-

Table 6. Selected ¹³C-NMR Chemical-Shift Values [ppm] of the 4-Chlorobenzylated Cellobiose Derivatives 7, 11, 13, 14, 17, and 18 in CDCl₃

	7	11	$13a$)	14	$17a$)	18
C(4')					86.29	67.21
C(3')					88.29	70.29
C(2')			91.40	74.33	71.08	68.81
C(1')			102.13	80.72	74.15	72.80
$C(1^I)$	102.47	168.44	70.26	69.64	70.17	70.01
$C(2^I)$	81.35	77.91	81.54	81.32	80.97	80.88
$C(3^I)$	$82.24b$)	$81.35b$)	83.88	83.97	83.94	83.96
$C(4^1)$	77.00	76.56	76.32	76.22	76.15	76.14
$C(5^I)$	74.88	76.68	79.19	79.28	79.28	79.31
$C(6^I)$	67.71	67.79	67.77	67.84	67.76	67.70
C(1 ^H)	102.84	103.78	102.28	102.35	102.26	102.35
$C(2^{\text{II}})$	81.63	80.88	81.77	81.79	81.73	81.76
C(3 ^H)	$82.60b$)	$81.58b$)	84.39	84.41	84.33	84.39
C(4 ^H)	81.17	80.22	73.79	73.80	73.65	73.67
C(5 ^H)	65.85	66.08	72.88	72.91	72.91	72.96
C(6 ^{II})	68.65	68.49	71.05	71.05	70.94	70.95

diene)iridium(I) hexafluorophosphate (1.16 g, 1.37 mmol) in dry THF (500 ml) was degassed and stirred under H_2 for 1 min (red suspension turns into pale yellow soln.). H_2 was replaced by Ar. The yellow soln. was treated with a soln. of 7 (50 g, 45.7 mmol) in THF (11), stirred for 1 h, treated with H₂O (500 ml) and I_2 (23.2 g, 91.4 mmol), stirred for 1 h, and diluted with cold 5% aq. Na₂S₂O₃ soln. (11). After evaporation of the pale yellow soln., a soln. of the residue in AcOEt was washed with brine, dried (MgSO₄), and evaporated. Crystallisation from Et₂O/MeOH gave 10 (41.8 g, 87%). R_f (AcOEt/hexane 1:3) 0.27. M.p. 125.6° .

4,6-O-Benzylidene-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 4)-2,3,6-tris-O-(4-chlorobenzyl)-D-glucono-1,5-lactone (11). At 26 $^{\circ}$ and under Ar, a suspension of 10 (10 g, 9.5 mmol) in DMSO (24 ml) was treated with Ac₂O (20 ml) and stirred for 6 h. The clear yellow soln. was poured into cold H₂O. The separated oil was decanted, dissolved in Et₂O, washed with H₂O and brine, and dried (Na_2SO_4) . After evaporation, drying of the residue under high vacuum for several hours gave crude 11 (9.7 g, 97%) as a pale brownish oil, which was used for the next step without further purification. FC (AcOEt/hexane 1:4) of a small sample gave pure 11 for microanalysis and optical rotation. R_f $(ACOEt/hexane 1:3)$ 0.46. $[\alpha]_D^{25} = +29.0$ $(c=1.0, CHCl_3)$. IR $(CHCl_3)$: 3034w, 2876w, 1758m, 1600w, 1493m, 1408w, 1365w, 1308w, 1276w, 1174w, 1090s, 1016m, 844m. ¹H-NMR (300 MHz, CDCl₃): see Table 5; additionally, 7.50 – 7.10 (m, 25 arom. H); 5.54 (s, PhCH); 4.87 (d, J=11.8), 4.85 (d, J=11.8) (2 PhCH); 4.70 (d, J=11.8, 2 PhCH); 4.66 (d, J=12.5), 4.63 (d, J=11.2), 4.58 (d, J=11.5), 4.56 (d, $J=11.8$). 4.47 (d, $J=11.5$), 4.37 (d, $J=12.2$) (6 PhCH). ¹³C-NMR (75 MHz, CDCl₃): see *Table 6*; additionally, 136.91, 136.64, 136.25, 135.97, 135.71, 135.08(6s); 134.03, 133.77, 133.48(3s); 133.64 (2s); 128.28 – 129.50 (several d); 125.94 (2d); 101.30 (d, PhCH); 74.59, 74.15, 72.70, 72.55, 72.37 (5t, 5 PhCH₂). Anal. calc. for $C_{54}H_{49}Cl_5O_{11}$ (1051.24): C 61.70, H 4.70; found: C 61.59, H 4.88.

6-O-Benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 6)-3,7-anhydro-4,5,8-tris-O-(4chlorobenzyl)-1,2-dideoxy-1-C-(trimethylsilyl)-D-glycero-D-gulo-oct-1-ynitol (13). Under Ar, a soln. of (trimethylsilyl)acetylene (11.6 ml, 82 mmol) in THF (276 ml) was cooled to -78° , treated dropwise with 1.6M BuLi in hexane (51.2 ml) over a period of 0.5 h, and stirred for 1 h. The resulting soln. was transferred into a cooled (-78°) soln. of 11 (19.1 g, 18.2 mmol) in THF (300 ml) over a period of 10 min with a canula. The pale brown soln. was stirred for 4 h, treated with sat. aq. NH₄Cl soln. (20 ml), warmed to r.t., stirred for 0.5 h, and worked up (AcOEt). Evaporation gave crude 12 (20.9 g of a pale yellow syrup, α/β 55:45) that was used for the next step without further purification.

A chilled (-40°) soln. of BF₃ · Et₂O (11.4 ml, 91 mmol) and Et₃SiH (23.1 ml, 145 mmol) in CH₂Cl₂/ MeCN 1:1 (210 ml) was transferred *via* a canula to a stirred and chilled (-40°) soln. of crude 12 (20.9 g, ca. 18.2 mmol) in CH₂Cl₂/MeCN 1 :1 (210 ml) over a period of 0.5 h. The mixture was stirred at -40° for 1 h, allowed to warm to -10° to -15° , stirred for 16 h, treated with sat. aq. NaHCO₃ soln. (200 ml), stirred for 1 h, warmed to r.t., and worked up (AcOEt). Evaporation and FC (AcOEt/hexane 1:7) gave 13 (14.8 g, 72%). Pale yellow oil. R_f (AcOEt/hexane 1:3) 0.39. $\left[\alpha\right]_D^{25} = 11.8$ ($c = 1.0$, CHCl₃). IR (CHCl₃): 3500w, 3036w, 2916w, 2873w, 2169w, 1702w, 1600w, 1493m, 1409w, 1362w, 1308w, 1275w, 1090s, 1016s, $847m$, $814w$. 1 H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY and a HSQC spectrum): see Table 5; additionally, 7.40 – 7.10 (m, 25 arom. H); 4.91 (d, $J=11.4$), 4.89 (d, $J=10.8$), 4.82 (d, $J=11.6$), 4.75 (d, $J=11.6$), 4.69 (d, $J=10.8$), 4.66 (d, $J=11.5$), 4.61 (d, $J=11.4$), 4.60 (d, $J=11.4$), 4.56 $(d, J=12.2)$, 4.46 $(d, J=12.3)$, 4.43 $(d, J=12.3)$, 4.37 $(d, J=12.5)$ (12 PhCH); 0.17 (s, Me₃Si). ¹³C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see Table 6; additionally, 137.57, 137.44, 137.13, 136.64, 136.55, 136.28(6s); 133.53 (2s); 133.51, 133.37, 132.95 (3s); 127.58– 129.38 (several d); 74.57, 74.45, 74.24, 74.10, 73.69, 72.58 (6t, 6 PhCH₂); -0.32 (q, Me₃Si). HR-MALDI-MS: 1155.2366 $(21, [M+Na]^+, C_{59}H_{61}Cl_5NaO_{10}Si^+$; calc. 1155.2374). Anal. calc. for $C_{59}H_{61}Cl_5O_{10}Si$ (1135.47): C 62.41, H 5.41; found: C 62.42, H 5.54.

6-O-Benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 6)-3,7-anhydro-4,5,8-tris-O-(4chlorobenzyl)-1,2-dideoxy-D-glycero-D-gulo-oct-1-ynitol (14) . A soln. of 13 $(6.3 g, 5.6 mmol)$ in THF $(30 g, 5.6 g,$ ml) was cooled to 0° , treated with a soln. of MeONa (0.6 g, 11 mmol) in MeOH (50 ml), and stirred for 1 h. After neutralisation with Amberlist IR 120 (H⁺ form), filtration, evaporation, and FC (AcOEt/hexane 1:11) gave 14 (5.28 g, 89%). Solid foam. R_f (AcOEt/hexane 1:3) 0.34. M.p. 112.9°. $[a]_D^{25} = +21.9$ $(c=1.0, \text{CHCl}_3)$. UV (CHCl₃): 267 (2948), 276 (1771). IR (CHCl₃): 3502w, 3306w, 3007w, 2910w, 2870w, 1600w, 1493s, 1457w, 1409w, 1362w, 1308w, 1298w, 1276w, 1089s, 1016s, 8 45w. ¹ H-NMR (300

MHz, CDCl₃): see *Table 5*; additionally, 7.40–7.10 (*m*, 25 arom. H); 4.93 (*d*, $J=11.2$), 4.86 (*d*, $J=11.2$), 4.82 (d, J=11.8), 4.78 (d, J=12.5), 4.74 (d, J=11.8), 4.67 (d, J=10.6) (6 PhCH); 4.60 (d, J=11.5, 2 PhCH); 4.56 (d, J=12.2, PhCH); 4.45 (d, J=11.8, 2 PhCH); 4.36 (d, J=12.5, PhCH). ¹³C-NMR (75 MHz, CDCl₃): see Table 6; additionally, 137.60, 137.57, 137.25, 136.76, 136.55, 136.42 (6s); 133.66 (2s); 133.70, 133.51, 133.14 (3s); 127.69 – 129.43 (several d); 74.64, 74.46, 74.33, 74.12, 73.70, 72.65 (6t, 6 PhCH₂). HR-MALDI-MS: 1083.1973 (56, C₅₆H₅₃Cl₅NaO₁₀, [M+Na]⁺; calc. 1083.1979). Anal. calc. for $C_{56}H_{53}Cl_5O_{10}$ (1063.29): C 63.26, H 5.02; found: C 63.32, H 5.22.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 6)$ -4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-Dglycero-D-gulo-oct-1-ynitol (15). A soln. of TMSOTf (17 ml, 94 mmol) in Ac₂O (10 ml) was added dropwise to a chilled (-40°) soln. of 14 (5.0 g, 4.7 mmol) in Ac₂O (92 ml). The mixture was warmed to 20^o, stirred for 16 h, cooled to 0° , treated dropwise with sat. aq. NaHCO₃ soln. (40 ml), stirred for 0.5 h, and worked up (AcOEt). Evaporation and FC (AcOEt/hexane 1:3) gave a colourless solid, which was recrystallised in AcOEt/hexane to afford 15 (2.3 g, 76%). Colourless needles. R_f (AcOEt/hexane 1:1) 0.26. M.p. 202[°]. $[\alpha]_D^{25} = -11.2$ (c=1.0, CHCl₃). UV (CHCl₃): 266 (1143). IR (CHCl₃): 3307w, 3034w, 2953w, 2865w, 2133w, 1755s, 1603w, 1457w, 1429w, 1368s, 1249s, 1169m, 1049s, 905w, 651w. ¹ H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY and a HSQC spectrum): see *Table 7*; additionally, 2.48 (d, J = 1.9, C = CH); 2.12, 2.07, 2.05, 2.01, 2.00, 1.99, 1.96 (7s, 7 AcO). ¹³C-NMR (75 MHz, CDCl₃; assignments based on a HSQC spectrum): see Table 8; additionally, 170.43, 170.28, 170.14, 169.74, 169.44, 169.26, 169.02 (7s, 7C=O); 20.81, 20.61, 20.57 (3q, 3 Me); 20.49 (q, 4 Me). HR-MALDI-MS: 667.1839 (65, $[M + Na]^+$, $C_{28}H_{36}NaO_{17}^+$; calc. 667.1850). Anal. calc. for $C_{28}H_{36}O_{17}$ (644.58): C 52.17, H 5.63; found: C 52.17, H 5.62.

6-O-Benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 8)-5,9-anhydro-6,7,10-tris-O-(4chlorobenzyl)-1,2,3,4-tetradeoxy-1-(trimethylsilyl)-D-glycero-D-gulo-deca-1,3-diynitol (17). Under Ar, a soln. of 1,4-bis(trimethylsilyl)buta-1,3-diyne (0.40 g, 2.0 mmol) in THF (10 ml) was cooled to 0° , treated dropwise with 1.4M MeLi·LiBr in Et₂O (1.26 ml, 1.9 mmol) over a period of 0.5 h, and stirred for 4 h at 27° . The deep-brown-coloured soln. was transferred into a cooled (-78°) soln. of 11 (1.0 g, 0.95 mmol) in THF (10 ml) over a period of 10 min with a canula. The resulting brown soln, was stirred for 4 h at -78° . treated with sat. aq. NH₄Cl soln. (10 ml), warmed to r.t., stirred for 0.5 h, and worked up (Et₂O). Evaporation gave crude 16 (1.15 g of a brownish syrup, α/β 3 : 2), which was used for the next step without further purification.

A chilled (-40°) soln. of BF₃ · OEt₂ (1.2 ml, 9.5 mmol) and Et₃SiH (2.4 ml, 152 mmol) in CH₂Cl₂/ MeCN 1:1 (10 ml) was added to a stirred and chilled (-40°) soln. of crude 16 (1.15 g, ca. 0.95 mmol) in CH₂Cl₂/MeCN 1:1 (10 ml) over a period of 0.5 h and with a canula. The mixture was stirred at -40° for 1 h and at -10 to -15° for 16 h, treated with sat. aq. NaHCO₃ soln. (10 ml), warmed to r.t., stirred for 1 h, and worked up (AcOEt). Evaporation and FC (AcOEt/hexane 1:6) gave 17 (0.58 g, 52%). Pale yellow oil. R_f (AcOEt/hexane 1:3) 0.36. $\left[\alpha\right]_D^{25} = -13.5$ ($c = 1.0$, CHCl₃). IR (CHCl₃): 3504w, 3034w, 2918w, 2874w, 2111w, 1703w, 1600m, 1493m, 1457w, 1409w, 1362w, 1308w, 1253m, 1090s, 1016s, $847m$, $634w$. 1 H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY and a HSQC spectrum): see Table 5; additionally, 7.40 – 7.10 (m, 25 arom. H); 4.94 (d, $J=11.4$), 4.84 (d, $J=10.9$), 4.83 (d, $J=11.7$), 4.76 (d, $J=11.6$), 4.71 (d, $J=10.9$), 4.68 (d, $J=11.5$) (6 PhCH); 4.62 (br. d, $J=11.1$, 2 PhCH); 4.56 (d, J = 12.2), 4.47 (d, J = 12.6), 4.45 (d, J = 12.6), 4.37 (d, J = 12.2) (4 PhCH); 0.23 (s, Me₃Si). ¹³C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see *Table 6*; additionally, 137.41, 137.39, 137.08, 136.58, 136.25, 136.16 (6s); 133.47 (2s); 133.62, 133.32, 132.99 (3s); 127.53 – 129.57 (several d); 74.63, 74.41, 74.29, 74.07, 73.67, 72.65 (6t, 6 PhCH₂); -0.53 (q, Me₃Si). HR-MALDI-MS: 1179.2376 $(14, [M+Na]^+, C_{61}H_{61}Cl_5NaO_{10}Si^+$; calc. 1179.2374). Anal. calc. for $C_{61}H_{61}Cl_5O_{10}Si$ (1159.50): C 63.19, H 5.30; found: C 63.23, H 5.41.

6-O-Benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 8)-5,9-anhydro-6,7,10-tris-O-(4 $chlorobenzyl$ -1,2,3,4-tetradeoxy-D-glycero-D-gulo-deca-1,3-diynitol (18). A soln. of 17 (50 mg, 43 µmol) in THF (0.3 ml) was cooled to 0° , treated with a soln. of MeONa (2 mg, 86 µmol) in MeOH (0.5 ml), stirred for 1 h, and passed through a pad of Amberlite IR 120 (H⁺ form, 0.5 ml wet). Evaporation of the pale yellow eluent and FC (AcOEt/hexane 1:5) gave 18 (50 mg, 87%). Pale white solid turning red on standing (it was used immediately for the next step). R_f (AcOEt/hexane 1:3) 0.45. M.p. 112.9°. $\left[\alpha\right]_D^{25} = -10.0$ $(c=1.0, CHCl₃)$. UV (CHCl₃): 267 (3697), 277 (2430). IR (CHCl₃): 3504w, 3300w, 3007w, 2910w, 2871w,

2075w, 1600w, 1492s, 1409w, 1363w, 1296w, 1089s, 1016s, 844w, 814w. ¹H-NMR (300 MHz, CDCl₃): see Table 5; additionally, 7.40 – 7.10 (m, 25 arom. H); 4.94 (d, $J=11.1$), 4.83 (d, $J=11.5$), 4.80 (d, $J=10.9$), 4.75 (d, J=11.5), 4.70 (d, J=10.6), 4.67 (d, J=11.5), 4.61 (d, J=11.2), 4.60 (d, J=11.5), 4.56 (d, $J=12.1$), 4.47 (d, $J=12.5$), 4.43 (d, $J=12.5$), 4.36 (d, $J=11.8$) (12 PhCH). ¹³C-NMR (75 MHz, CDCl₃): see Table 6; additionally, 137.56, 137.49, 137.21, 136.71, 136.32, 136.20 (6s); 133.85, 133.70, 133.62, 133.44, 133.15 (5s); 127.65 – 129.64 (several d); 74.72, 74.44, 74.31, 74.10, 73.67, 72.68(6t, 6 PhCH2). $HR\text{-}MALDI\text{-}MS: 1107.1965 (56, $[M + Na]^+$, $C_{58}H_{53}Cl_5NaO_{10}^+$; calc. 1107.1979). Anal. calc. for$ $C_{58}H_{53}Cl_5O_{10}$ (1087.31): C 64.07, H 4.91; found: C 64.17, H 5.13.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 8)$ -6,7,10-tri-O-acetyl-5,9-anhydro-1,2,3,4-tetradeoxy-D-glycero-D-gulo-deca-1,3-diynitol (19). A soln. of TMSOTf (1.7 ml, 9.2 mmol) in Ac₂O (3.8 ml) was added dropwise to a chilled (-40°) soln. of 18 (0.50 g, 0.46 mmol) in Ac₂O (9 ml). The mixture was warmed to 25°, stirred for 24 h, cooled to 0°, treated dropwise with sat. aq. NaHCO₃ soln. (4 ml), stir-

Table 7. Selected ¹H-NMR Chemical-Shift Values [ppm] and Coupling Constants [Hz] of the Peracetylated Cellobiosylalkynes 15, 19, 21, 22, and 24 in CDCl₃

	$15a$)	$19a$)	$21a$)	$22^{b})$	24°)	
					E chain ^b)	B chain
$H - C(1^1)$	4.14	4.20	4.56	4.56	4.66	4.41
$H - C(2^I)$	5.07	5.05	5.32	5.28	5.32	5.14
$H - C(3I)$	5.13	5.11	5.25	5.23	5.25	5.20
$H - C(4I)$	3.76	3.76	3.87	3.85	3.86	3.89
$H - C(5^I)$	3.56	3.55	3.71	3.71	3.76	3.68
$H_a-C(6^I)$	4.46	4.46	4.57	4.53	4.58	4.53
$H_b - C(6^1)$	4.07	4.06	4.15	4.16	4.16	4.15
$H - C(1H)$	4.49	4.48	4.55	4.55	4.63	4.55
$H - C(2H)$	4.90	4.89	4.95	4.95	4.95	4.94
$H - C(3H)$	5.12	5.12	5.16	5.16	5.19	5.17
$H - C(4H)$	5.04	5.04	5.08	5.08	5.10	5.09
$H - C(5H)$	3.64	3.64	3.69	3.69	3.73	3.70
$H_a-C(6^{\text{II}})$	4.35	4.35	4.39	4.39	4.42	4.39
$H_b-C(6II)$	4.03	4.03	4.07	4.07	4.08	4.07
$J(1^I, \equiv CH)$	2.0	0.9	$\qquad \qquad -$	-	$\overline{}$	$\overline{}$
$J(1^{\rm I},2^{\rm I})$	9.6	9.7	9.6	9.8	9.7	9.7
$J(2^{\rm I},3^{\rm I})$	9.3	9.6	9.4	c)	c)	9.4
$J(3^{\rm I},4^{\rm I})$	8.8	8.9	9.1	9.4	9.4	8.9
$J(4^{\rm I},5^{\rm I})$	9.8	9.8	9.8	9.9	9.9	9.9
$J(5^{\rm I},6{\rm a}^{\rm I})$	5.2	5.1	5.1	5.2	4.8	5.6
$J(5^{\rm I},6{\rm b}^{\rm I})$	1.9	2.0	1.9	1.9	1.8	2.0
$J(6a^{\rm I}, 6b^{\rm I})$	12.1	12.1	12.2	12.2	12.2	12.2
$J(1^{\rm II},2^{\rm II})$	7.9	7.9	8.0	7.9	7.9	7.9
$J(2^{\text{II}},3^{\text{II}})$	9.3	9.3	9.3	9.3	9.4	9.4
$J(3^{\text{II}},4^{\text{II}})$	9.5	9.4	9.4	9.4	9.4	9.4
$J(4^{\text{II}},5^{\text{II}})$	9.9	9.7	9.8	9.9	10.0	9.9
$J(5^{\text{II}},6\text{a}^{\text{II}})$	4.4	4.4	4.4	4.6	4.2	4.6
$J(5^{\text{II}},6b^{\text{II}})$	2.3	2.3	2.3	2.3	2.3	2.4
$J(6a^{\text{II}}, 6b^{\text{II}})$	12.5	12.5	12.5	12.5	12.5	12.4

^a) Assignments based on a DQFCOSY and a HSQC spectrum. ^b) H–C(2¹) and H–C(3¹) of 22 and the E chain of 24 form a narrow AB system leading to virtual couplings with H–C(1¹) and H–C(4¹). ^c) Not assigned.

Table 8. Selected ¹³C-NMR Chemical-Shift Values [ppm] of the Peracetylated Cellobiosylalkynes **15, 19, 21**, 22, and 24 in CDCl₃, and of the Deprotected Cellobiosylalkynes 23 and 25 in $(D_6)DMSO$

	$15a$)	$19a$)	21°)	22	24°)		$23^{\rm a}$)	$25^{\rm a})$	
					E chain B chain			E chain	B chain
C(4')		66.82				78.74 ^b)			78.82
C(3')		69.84				$78.17b$)			77.30
C(2')	75.31	69.46	85.62	84.93	85.87	71.92	83.89	84.20	69.65
C(1')	77.80	71.10	90.14	89.99	90.17	$79.27b$)	94.51	94.02	84.20
$C(1^I)$	68.31	68.69	69.39	69.39	69.48	69.09	70.71	70.69	70.38
$C(2^I)$	71.20	70.91	71.44	71.35	71.14	71.05	73.44	73.56	73.23
$C(3^I)$	72.90	72.88	73.53	73.24	73.15	73.37	76.38	75.80	75.70
$C(4^I)$	76.14	76.02	76.28	76.38	76.31	76.09	80.04	79.98	79.81
$C(5^I)$	76.84	76.92	76.93	76.89	76.76	77.05	78.88	78.92	78.78
$C(6^I)$	62.07	61.94	62.18	62.37	62.22	62.29	60.21	60.15	60.15
C(1 ^H)	100.77	100.75	100.84	100.76	100.73	100.73	103.01	103.00	103.00
C(2 ^H)	71.56	71.55	71.64	71.67	71.68	71.64	73.23	73.19	73.11
C(3 ^H)	73.09	73.03	72.97	73.03	73.01	72.96	76.54	76.35	76.35
C(4 ^H)	67.76	67.75	67.81	67.93	67.95	67.86	69.95	69.93	69.91
C(5 ^H)	71.97	71.98	72.02	72.04	72.03	71.92	76.71	76.67	76.67
C(6 ^{II})	61.53	61.52	61.58	61.67	61.56	61.70	60.95	60.94	60.94

^a) Assignments based on a HSQC spectrum. ^b) Assignments may be interchanged.

red for 0.5 h, and worked up (AcOEt). Evaporation and FC (AcOEt/hexane 1 : 2) gave a colourless solid, which was recrystallised in AcOEt/hexane affording 19 (0.2 g, 63%). Colourless needles. R_f (AcOEt/hexane 1:1) 0.22. M.p. 200.7°. $[a]_D^{25} = -21.6$ ($c = 1.0$, CHCl₃). UV (CHCl₃): 265 (1062). IR (CHCl₃): 3300w, 3034w, 2954w, 2867w, 1756s, 1603w, 1428w, 1368m, 1249s, 1168w, 1051s, 908w. ¹ H-NMR (500 MHz, CDCl₃; assignments based on a DOFCOSY and a HSOC spectrum): see *Table 7*; additionally, 2.20 (d, $J=0.9$, C \equiv CH); 2.12, 2.07, 2.05, 1.99, 1.96 (5s, 5 AcO); 2.01 (s, 2 AcO). ¹³C-NMR (75 MHz, CDCl₃, assignment based on a HSQC spectrum): see Table 8; additionally, 170.43, 170.25, 170.14, 169.72, 169.30, 169.26, 169.01 (7s, 7C=O); 20.80, 20.60, 20.51 (3q, 3 Me); 20.48(q, 4 Me). HR-MALDI-MS: 691.1843 (43, $[M + Na]^+$, $C_{30}H_{36}NaO_{17}^+$; calc. 691.1850). Anal. calc. for $C_{30}H_{36}O_{17}$ (668.60): C 53.89, H 5.43; found: C 53.89, H 5.51.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 6)$ -4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-1-C-(8-hydroxy-9,10-dioxoanthracen-1-yl)-D-glycero-D-gulo-oct-1-ynitol (21). Under Ar, a stirred suspension of 20 (0.9 g, 2.2 mmol), $[Pd(PPh_3)_2]Cl_2$ (75 mg, 0.10 mmol), and CuI (60 mg, 0.31 mmol) in degassed Et₃N/ DMF 1:5 (10 ml) was warmed to 60°, treated dropwise with a soln. of 15 (1.9 g, 2.9 mmol) in Et₃N/DMF 1:5 (6 ml) over a period of 3 h, stirred for 4 h, treated with solid (NH₄),CO₃ (0.45 g, 5.8 mmol), cooled to 23° , stirred for 12 h, treated with additional (NH₄)₂CO₃ (0.22 g, 2.8 mmol), and stirred for 6 h. Workup (AcOEt), evaporation, and FC (AcOEt/hexane/CH₂Cl₂ 2:3:3) gave bright yellow 21 (1.4 g, 74%). R_f (AcOEt/hexane 1:1) 0.17. M.p. 263-265° (dec.). $[a]_D^{25} = -4.6$ ($c = 1.0$, CHCl₃). UV (CHCl₃): 276 (16427), 373 (4481), 412 (7039), 436 (3840). IR (CHCl3): 3586w, 3008w, 2954w, 2866w, 1756s, 1673m, 1640m, 1603w, 1581w, 1456m, 1435w, 1367m, 1318w, 1283m, 1166w, 1150w, 1094w, 1042m, 920w, 8 99w, 850w. ¹H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY and a HSQC spectrum): see Table 7; additionally, 12.49 (d, J=0.3, HO-C(8')); 8.32 (dd, J=7.8, 1.4, H-C(4')); 7.89 (dd, J=7.7, 1.4, H-C(2')); 7.80 (dd, J=7.5, 1.2, H-C(5')); 7.73 (t, J=7.8, H-C(3')); 7.67 (td, J \approx 7.9, 0.3, H- $C(6')$); 7.32 (dd, J \approx 8.4, 1.2, H $-C(7')$); 2.16, 2.12, 2.11, 2.07, 2.06, 2.02, 1.99 (7s, 7 AcO). ¹³C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see Table 8; additionally, 187.32 (s, $C(9')$); 181.79 (s, $C(10')$); 170.52, 170.37, 170.22, 169.92, 169.76, 169.32, 169.10 (7s, 7 OC=O); 162.58 (s, $C(8')$; 140.87 (d, $C(2')$); 136.60 (d, $C(6')$); 134.47 (s, $C(9'a)$); 133.47 (d, $C(3')$); 133.35 (s, $C(4'a)$); 132.73 (s, C(10'a)); 128.24 (d, C(4')); 124.82 (d, C(7')); 122.03 (s, C(1')); 119.33 (d, C(5')); 116.3 (s, $C(8'a)$); 20.93, 20.91, 20.68, 20.62 (4q, 4 Me); 20.56 (q, 3 Me). HR-MALDI-MS: 889.2163 (60, $[M+Na]^+$, $C_{42}H_{42}NaO_{20}^+$; calc. 889.2167). Anal. calc. for $C_{42}H_{42}O_{20}$ (866.78): C 58.20, H 4.88; found: C 58.26, H 4.97.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 6)$ -4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-1-C-{8-[(trifluoromethyl)sulfonyloxy]-9,10-dioxoanthracen-1-yl}-D-glycero-D-gulo-oct-1-ynitol (22). A soln. of 21 (0.5 g, 0.58 mmol) in dry CH₂Cl₂ (8.5 ml) was treated with Et₃N (160 μ l, 1.2 mmol) and cooled to -78° . The resulting red suspension was treated with freshly distilled Tf₂O (120 μ l, 2 mmol), stirred for 1 h, allowed to warm to 0° , stirred for 30 min, and diluted with CH₂Cl₂ (250 ml). Workup, evaporation, and FC (AcOEt/hexane 2:3) gave 22 (0.55 g, 96%) as a pale yellow solid, which was recrystallised in AcOEt/hexane. R_f (AcOEt/hexane 1:1) 0.17. M.p. 234.1–235.4° (dec.). $\left[\alpha\right]_D^{25} = +29.7$ (c=1.0, CHCl₃). UV (CHCl₃): 274 (17609), 368 (3831). IR (CHCl₃): 3068w, 2950w, 2870w, 1756s, 1682m, 1599w, 1434m, 1368m, 1323m, 1249s, 1157w, 1140m, 1094w, 1048m, 987w, 898w, 836w. ¹H-NMR (500 MHz, CDCl₃): see Table 7; additionally, 8.37 (dd, J = 7.8, 1.2, H – C(4')); 8.27 (dd, J = 7.8, 1.3, H – C(5')); 7.89 (dd, $J=7.8$, 1.3, H-C(2')); 7.85 (t, $J\approx8.0$ H-C(6')); 7.73 (t, $J=7.8$, H-C(3')); 7.63 (br. d, $J\approx8.2$, H- $C(7')$; 2.05 (s, 2 AcO); 2.16, 2.12, 2.11, 2.02, 1.99 (5s, 5 AcO). ¹³C-NMR (125 MHz, CDCl₃): see *Table* 8; additionally, 181.10 (s, C(9')); 179.82 (s, C(10')); 170.54, 170.46, 170.22, 170.02, 169.79, 169.33, 169.10 $(7s, 7 \text{ OC}=0); 147.49 \text{ (s, C}(8)); 140.95 \text{ (d, C}(2)); 134.95 \text{ (d, C}(6)); 134.77 \text{ (s, C}(9a)); 134.48 \text{ (s, C}(9a))$ $C(10'a)$; 133.23 (d, C(3')); 133.19 (s, C(4'a)); 128.89 (d, C(4')); 127.71 (d, C(7')); 127.62 (d, C(5')); 126.63 (s, C(8'a)); 122.03 (s, C(1')); 118.80 (q, 1 J(C,F)=321.0, CF₃); 20.84, 20.72, 20.67, 20.62, 20.30 $(5q, 5 \text{ Me})$; 20.55 $(q, 2 \text{ Me})$. ¹⁹F-NMR (282 MHz, CDCl₃): -73.63 (s, CF_3) . HR-MALDI-MS: 1021.1643 (12, $[M+Na]^+$, $C_{43}H_{41}F_3NaO_{22}S^+$; calc. 1021.1660). Anal. calc. for $C_{43}H_{41}F_3O_{22}S$ (998.84): C 51.71, H 4.14; found: C 51.70, H 4.24.

 β -D-Glucopyranosyl-(1 \rightarrow 6)-3,7-anhydro-1,2-dideoxy-1-C-(8-hydroxy-9,10-dioxoanthracen-1-vl)-Dglycero-p-gulo-*oct-1-ynitol* (23). A stirred suspension of 21 (0.25 g, 0.29 mmol) in dry MeOH (18 ml) was treated with a soln. of 0.02M MeONa (18 ml), stirred for 6 h at 20° , treated with H₂O (10 ml), stirred for 1 h, and neutralised with *Amberlite IR-120* (H^+ form). After filtration, evaporation of the filtrate gave 23 (158 g, 95%). Bright yellow solid. R_f (RP-18 silica gel; MeCN/H₂O 2 : 1) 0.33. M.p. 152.8° (dec., \rightarrow green residue). $[\alpha]_D^{25} = +13.2$ (c=1.0, DMSO). ¹H-NMR (500 MHz, (D₆)DMSO; assignments based on a DOFCOSY and a HSOC spectrum): see *Table 1*: additionally, 12.46 (br. s, $HO-C(8')$); 8.22 (dd, $J=7.8$, 1.4, H-C(4')); 8.00 (dd, $J=7.7$, 1.4, H-C(2')); 7.91 (t, $J=7.7$, H-C(3')); 7.79 (t, $J=7.9$, H- $C(6')$); 7.69 (br. d, J \approx 7.5, H $-C(5')$); 7.39 (br. d, J \approx 8.2, H $-C(7')$); 4.31 (br. d, J $=7.9$, H $-C(3¹)$, H $C(1^{11})$; 3.81 (dd, J \approx 11.8, 5.0, H – $C(8^1)$); 3.73 (br. d, J = 9.9, H – $C(6^{11})$); 3.66 (dd, J \approx 11.0, 5.3, H' – $C(8¹)$); 3.46–3.30 (m, H–C(4¹), H–C(5¹), H–C(6¹), H–C(7¹), H'–C(6^{I1})); 3.24 (ddd, J \approx 9.6, 6.2, 2.2, $H-C(5^{II})$; 3.19 (br. t, $J=9.0$, $H-C(3^{II})$); 3.09 (td, $J\approx 9.4$, 3.4, $H-C(4^{II})$); 3.03 (td, $J\approx 8.6$, 4.1, $H-J$ $C(2^{II})$). ¹³C-NMR (125 MHz, (D_6) DMSO; assignments based on a HSQC spectrum): see *Table 8*; additionally, 186.83 (s, C(9')); 181.44 (s, C(10')); 161.45 (s, C(8')); 140.94 (d, C(2')); 136.70 (d, C(6')); 134.04 (s, C(9'a)); 133.92 (d, C(3')); 132.61 (s, C(4'a)); 132.54 (s, C(10'a)); 127.26 (d, C(4')); 124.28(d, $C(7')$); 122.06 (d, C(1')); 118.50 (d, C(5')); 116.27 (s, C(8'a)). MALDI-TOF: 573.5 ([M+H]⁺, C₂₈H₂₉O⁺₁₃; calc. 573.2), 595.5 ([M+Na]⁺, C₂₈H₂₈NaO⁺₁₃; calc. 595.1), 611.4 ([M+K]⁺, C₂₈H₂₈KO⁺₁₃; calc. 611.1). Anal. calc. for $C_{28}H_{28}O_{13} \cdot H_2O$ (590.53): 56.95, H 5.12; found: C 56.98, H 5.22.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 8)$ -6,7,10-tri-O-acetyl-5,9-anhydro-1,2,3,4-tetrade oxy -1-C-{8-[2,3,4,6-tetra-O-acetyl- β -D-glucopyranosyl-(1 \rightarrow 6)-4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-D-glycero-D-gulo-oct-1-ynitol-1-yl]-9,10-dioxoanthracen-1-yl}-D-glycero-D-gulo-deca-1,3-diynitol (24). Under Ar, a suspension of 21 (0.4 g, 0.40 mmol), $[Pd(PPh₃)₂]Cl₂$ (14 mg, 0.02 mmol), and CuI (11.5) mg, 0.06 mmol) in degassed Et₃N/DMF 1:5 (4 ml) was treated with a soln. of 19 (0.54 g, 0.80 mmol) in Et₃N/DMF 1:5 (4 ml) over a period of 12 h, and stirred at 24° for additional 12 h. Workup (AcOEt), evaporation, and FC (CH₂Cl₂/AcOEt 5:1 \rightarrow 2:1) gave **24** (0.53 g, 87%) as a pale yellow solid, which was recrystallised in AcOEt/MeOH. R_f (AcOEt/hexane 2:1) 0.20. M.p. 302.7-304.1° (dec.). $[\alpha]_D^{25}$ = +14.8 (c = 1.0, CHCl₃). UV (CHCl₃): 277 (30643), 289 (14415), 311 (12508), 373 (8107). IR (CHCl3): 3034w, 2947w, 2867w, 1755s, 1677w, 1603w, 1574w, 1431w, 1368m, 1334w, 1249s, 1094w, $1048m$, $908w$, $836w$. $\rm ^1H\text{-}NMR$ (500 MHz, CDCl₃; assignments based on a DQFCOSY and a HSQC spectrum): see Table 7; additionally, 8.31 $(dd, J=7.8, 1.4, H-C(5'))$; 8.29 $(dd, J=7.8, 1.4, H-C(4'))$; 7.95 $(dd,$ $J=7.8, 1.4, H-C(7'))$; 7.92 (dd, $J\approx7.8, H-C(2'))$; 7.72 (t, $J=7.8, H-C(6'))$; 7.70 (t, $J\approx7.8, H-C(3'))$; 2.19, 2.18, 2.16, 2.13, 2.11, 2.10, 2.07, 2.061, 2.059, 2.05, 2.02, 2.01, 2.00, 1.98 (14s, 14 AcO). 13C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see Table 8; additionally, 181.78 (s, C(10')); 179.82 (s, C(9')); 170.51 (s, 2 OC=O); 170.43, 170.31, 170.25, 170.22, 170.19, 169.87, 169.77, 169.52, 169.36, 169.32, 169.17, 169.11 (12s, 12 OC=O); 142.01 (d, C(2')); 141.37 (d, C(7')); 135.50 (s, $C(9'a)$; 133.81 (s, $C(8'a)$); 133.72 (s, $C(4'a)$); 133.49 (s, $C(10'a)$); 132.88 (d, $C(6')$); 132.77 (d, $C(3')$); 128.12 (d, C(4')); 127.75 (d, C(5')); 122.37 (s, C(8')); 121.47 (s, C(1')); 20.99, 20.93, 20.90, 20.73, 20.69, 20.67, 20.65, 20.58, 20.52 (9q, 9 Me); 20.56 (q, 3 Me); 20.55 (q, 2 Me). HR-MALDI-MS: 1539.4000 (12, $[M+Na]^+$, $C_{72}H_{76}NaO_{36}^+$; calc. 1539.4013). Anal. calc. for $C_{72}H_{76}O_{36}$ (1517.37): C 56.99, H 5.05; found: C 57.03, H 5.17.

 β -D-Glucopyranosyl-(1 \rightarrow 8)-5,9-anhydro-1,2,3,4-tetradeoxy-1-C-{8-{ β -D-glucopyranosyl-(1 \rightarrow 6)-3, 7-anhydro-1,2-dideoxy-D-glycero-D-gulo-oct-1-ynitol-1-yl]-9,10-dioxoanthracen-1-yl}-D-glycero-D-gulodeca-1,3-diynitol (25). Under Ar, a stirred suspension of KCN (4 mg, 0.06 mmol) in dry MeOH (4.0 ml) was treated with a soln. of 24 (150 mg, 99 µmol) in CH₂Cl₂ (8 ml), stirred at 25 \degree for 36 h, treated with H₂O (10 ml) , stirred for 20 min, and diluted with H₂O (10 ml) . Evaporation of the org. solvents and lyophilisation gave a pale greenish-yellow residue, which was subjected to reversed-phase (RP) FC (Lichroprep $RP-18$, 40–63 µm, MeCN/H₂O 1:5 \rightarrow 1:4 \rightarrow 1:3). Evaporation and lyophilisation gave 25 (65 mg, 71%) as a greenish yellow solid, which turned brown upon standing at r.t. R_f ($RP-18$ silica gel; MeCN/H₂O 2:1) 0.25. ¹H-NMR (500 MHz, (D₆)DMSO; assignments based on a DQFCOSY and a HSQC spectrum): see Table 1; additionally, 8.26 (dd, J=7.7, 1.3), 8.22 (dd, J=7.7, 1.4) (H-C(4'), H-C(5')); 8.13 (dd, J \approx 7.7, 1.3), 8.03 (dd, $J \approx 7.8$, 1.4) (H-C(2'), H-C(7')); 7.90 (t, J=7.7), 7.89 (t, J=7.9) (H-C(3'), H-C(6')); 4.33 (d, J = 8.6, H – C(3^{LE})); 4.31 (d, J = 7.6, H – C(1^{LE})); 4.28 (d, J = 7.9, H – C(1^{LE})); 4.23 (d, J = 9.4, $H-C(5^{1B}))$; 3.81 (dd, J \approx 12.0, 5.6, H $-C(8^{1E}))$; 3.78 (dd, J \approx 12.0, 5.9, H $-C(10^{1B}))$; 3.74 – 3.60 (m, H – $C(6^{IIB})$, H- $C(6^{IIE})$, H' $-C(10^{I.B})$, H' $-C(8^{I E})$); 3.45–3.30 (m, H' $-C(6^{IIB})$, H' $-C(6^{IIE})$, H- $C(9^{IIB})$, H- $C(7^{1E})$, H-C(8^{1B}), H-C(6^{IE}), H-C(7^{IB}), H-C(5^{IE}), H-C(6^{IB}), H-C(4^{IE})); 3.28–3.20 (m, H-C(5^{IIB}), $H-C(5^{\text{HE}})$); 3.19–3.12 (m, $H-C(3^{\text{HE}})$, $H-C(3^{\text{HE}})$); 3.10–2.99 (m, $H-C(4^{\text{HE}})$, $H-C(4^{\text{HE}})$, $H-C(2^{\text{HE}})$, $H-C(2^{ILE})$). ¹³C-NMR (125 MHz, (D₆)DMSO; assignments based on a HSQC spectrum): see Table 8; additionally, 181.43 (s, C(10')); 179.58 (s, C(9')); 141.94 (d, C(2')); 141.68 (d, C(7')); 135.03 (s, C(9'a)); 133.47 (s, C(8'a)); 133.35 (s, C(4'a)); 133.27 (d, C(3')); 133.22 (d, C(6')); 133.10 (s, C(10'a)); 127.60 (d, $C(4')$); 127.00 $(d, C(5'))$; 122.07 $(s, C(8'))$; 120.17 $(s, C(1'))$. MALDI-TOF: 929.2 $([M + H]^+, C_{44}H_{49}O_{22}^+;$ calc. 929.3), 951.2 $([M + Na]^+, C_{44}H_{48}NaO_{22}^+;$ calc. 951.3), 967.2 $([M + K]^+, C_{44}H_{48}KO_{22}^+;$ calc. 967.2).

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 4)$ -6-O-benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 6)-3,7-anhydro-4,5,8-tris-O-(4-chlorobenzyl)-1,2-dideoxy-1-C-(trimethylsilyl)-D-glycero-D-gulo-oct-1-ynitol (26). A suspension of 13 (6.7 g, 5.9 mmol), AgOTf (9.1 g, 35.4 mmol), and 3-Å molecular sieves (15 g) in dry toluene (135 ml) was stirred under Ar at 23° for 1 h, cooled to -30° to -35° , stirred for 15 min, treated with a soln. of 2 (8.24 g, 12 mmol) in CH₂Cl₂ (20 ml), stirred for 3 h, allowed to warm to r.t., and stirred for 12 h. The mixture was cooled to -35° , treated with an additional batch of 2 (8.26 g, 12 mmol) in CH₂Cl₂ (20 ml), warmed slowly to r.t., and stirred for 48 h. After cooling to -10° , the suspension was treated with Et₃N (6.0 ml), stirred for 15 min, and filtered through a pad of *Celite* (washing of the residue thoroughly with 500 ml of $CHCl₃$). The combined org. layers and washing were dried (MgSO₄) and evaporated. FC (AcOEt/hexane 7:12) gave 26 (8.7 g, 84%). R_f (AcOEt/hexane 2:3) 0.31. M.p. 80.7°. $\left[\alpha\right]_D^{25} = -9.5$ ($c = 1.0$, CHCl₃). IR (CHCl3): 3020m, 2952m, 2916m, 2871w, 2179w, 1756s, 1601w, 1492m, 1460w, 1410w, 1363m, 1162m, 1085s, 1058s, 1020s, 910w, 842m. ¹H-NMR (500 MHz, CDCl₃; assignments based on a HSQC spectrum): see Table 9; additionally, 7.40 – 6.95 (m, 25 arom. H); 5.02 (d, $J=11.9$), 4.95 (d, $J=11.9$), 4.87 (d, $J=11.6$), 4.66 $(d, J=10.8)$, 4.60 $(d, J=12.0)$, 4.54 $(d, J=11.5)$, 4.53 $(d, J=12.2)$, 4.52 $(d, J=12.3)$ (8 PhCH); 4.51 $(d, J=12.3)$ $J=11.5$, 2 PhCH); 4.35 (d, $J=12.2$), 4.27 (d, $J=12.0$) (2 PhCH); 2.07, 2.02, 2.00, 1.99, 1.98, 1.95, 1.89 (7s, 7) ACO); 0.16 (s, Me₃Si). ¹³C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see Table 10; additionally, 170.46, 170.17, 170.11, 169.69, 169.41, 169.29, 168.98 (7s,7C=O); 137.76, 137.72, 137.64, 136.57, 136.50, 136.33 (6s); 133.56, 133.51, 133.42, 132.98, 132.90 (5s); 129.35 – 127.83 (several d); 74.58, 74.33, 74.14, 74.10, 73.40, 72.53 (6t, 6 PhCH₂); 20.64, 20.57, 20.53, 20.50 (4q, 7 Me); -0.30 (q, Me₃Si). HR-MALDI-MS: 1773.4165 (45, $[M+Na]^+$, $C_{85}H_{95}Cl_5NaO_{27}Si^+$; calc. 1773.4170). Anal. calc. for $C_{85}H_{95}Cl_5O_{27}Si$ (1754.02): C 58.21, H 5.46; found: C 58.07, H 5.36.

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Table 9. Selected ¹H-NMR Chemical-Shift Values [ppm] and Coupling Constants [Hz] of the 4-Chlorobenzylated Cellotetraosylalkynes **26, 27, 29, and 30 in CDCl**

 $2,3,4,6$ -Tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 4)$ -6-O-benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 6)-3,7-anhydro-4,5,8-tris-O-(4-chlorobenzyl)-1,2-dideoxy-D-glycero-D-gulo-oct-1-ynitol (27). At 0° , a stirred soln. of 26 (1.0 g, 0.57 mmol) in THF (40 ml) was treated with a soln. of $Bu_4NF \cdot 3 H_2O$ (54 mg, 0.17 mmol) in THF (10 ml), stirred for 5 min, and treated with H₂O (10 ml). Workup (AcOEt) and FC (AcOEt/hexane 2:3) gave 27 (0.83 g, 86%). Solid foam. R_f (AcOEt/hexane 2:3) 0.23. M.p. 82°. $[\alpha]_D^{25} = -4.1$ ($c = 1.0$, CHCl₃). IR (CHCl₃): 3304w, 3033w, 2953w, 2871w, 2124w, 1757s, 1601w, 1492m, 1460w, 1410w, 1363m, 1162m, 1085s, 1058s, 1018s, 910w, 842m. ¹H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY and a HSQC spectrum): see Table 9; additionally, 7.40–6.95 (m, 25 arom. H); 5.04 (d, J=11.8), 4.96 (d, J=11.9), 4.85 (d, $J=10.8$), 4.67 (d, $J=10.8$), 4.60 (d, $J=11.9$), 4.55 (d, $J=11.5$) (6 PhCH); 4.53 (d, $J=11.5$, 2 PhCH); 4.52 $(d, J=12.1), 4.51 (d, J=12.0), 4.34 (d, J=12.3), 4.28 (d, J=12.0) (4 PhCH); 2.07, 2.06, 2.00, 1.99, 1.98,$ 1.95, 1.89 (7s, 7 AcO). ¹³C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see Table 10; additionally, 170.45, 170.17, 170.10, 169.68, 169.41, 169.29, 168.98 (7s,7C=O); 137.75, 136.57, 136.38, 136.34 (4s); 137.63 (2s); 133.60, 133.50, 133.42, 133.04, 132.91 (5s); 129.35 – 127.83 (several d); 74.65, 74.43, 74.16, 74.11, 73.40, 72.59 (6t, 6 PhCH₂); 20.64, 20.57, 20.53, 20.51 (4q, 7 Me). HR-MALDI-MS: 1701.3779 (46, $[M + Na]^+$, C₈₂H₈₇Cl₅NaO₂₇; calc. 1701.3775). Anal. calc. for C₈₂H₈₇Cl₅O₂₇ (1681.84): C 58.56, H 5.21; found: C 58.39, H 5.39.

 $2,3,4,6$ -Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl]₂- $(1 \rightarrow 6)$ -4,5,8-tri-O-acetyl-1,2-dideoxy-D-glycero-D-gulo-oct-1-ynitol (28). At -40° , a soln. of 27 (10.8 g, 6.4 mmol) in Ac₂O (108 ml) was treated with TMSOTf (23.2 ml, 0.13 mol), stirred for 3 h at -40° , warmed

to -20° , stirred for 24 h, warmed to -10° , and stirred for 20 h. After cooling to -20° , the mixture was treated portionwise with sat. aq. NaHCO₃ soln. (40 ml), and stirred for 0.5 h. Workup (AcOEt), evaporation, and FC (AcOEt/hexane 1:2 \rightarrow 3:2) gave a colourless solid (7.0 g). Recrystallisation in AcOEt/ hexane gave 28 (6.0 g, 76%). Colourless needles. R_f (AcOEt/hexane 3:2) 0.20. M.p. 226.4°. $[\alpha]_{\text{D}}^{25}$ = -16.8 (c=1.0, CHCl₃). IR (CHCl₃): 3305w, 3020w, 2952w, 2871w, 2138w, 1757s, 1431w, 1370s, 1160m, 1045s, 902w, 804w. ¹H-NMR (500 MHz, CDCl₃): see *Table 11*; additionally, 2.143, 2.141, 2.13, 2.09, 2.06, 2.03, 2.02, 2.01, 2.00, 1.99, 1.983, 1.979, 1.95 (13s, 13 AcO). ¹³C-NMR (125 MHz, CDCl₃): see Table 12; additionally, 170.50, 170.35, 170.21, 170.20, 170.19, 169.80, 169.75, 169.72, 169.45, 169.33, 169.30, 169.29, 169.11 (13s, 13 C=O); 20.89, 20.67, 20.61, 20.59, 20.54, 20.49, 20.47 (7q, 7 Me); 20.80, 20.56, 20.53 (3q, 6 Me). HR-MALDI-MS: 1243.3527 (100, $[M+Na]^+$, $C_{52}H_{68}NaO_{33}^+$; calc. 1243.3541). Anal. calc. for C₅₂H₆₈O₃₃ (1221.09): C 51.15, H 5.61; found: C 51.29, H 5.61.

2,3,4,6-Tetra-O-acetyl-β-D-glucopyranosyl-(1 \rightarrow 4)-2,3,6-tri-O-acetyl-β-D-glucopyranosyl-(1 \rightarrow 4)-6-O-benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 8)-5,9-anhydro-6,7,10-tris-O-(4-chlorobenzyl)-1,2,3,4-tetradeoxy-1-C-(trimethylsilyl)-D-glycero-D-gulo-deca-1,3-diynitol (29). Under Ar, a suspension of 17 (7.93 g, 6.8 mmol), AgOTf (9.1 g, 35.4 mmol), and 4-Å molecular sieves (16 g) in dry toluene (80 ml) was stirred at 25° for 1 h, cooled to -35° , stirred for 15 min, treated with a soln. of 2 (9.6) g, 13.7 mmol) in CH₂Cl₂ (40 ml), stirred for 3 h at -35° , and left for 24 h at r.t. The mixture was cooled to -35° , treated with an additional batch of 2 (6.5 g, 9.3 mmol) in CH₂Cl₂ (40 ml), gradually brought to r.t., and stirred for 24 h. The mixture was cooled to -10° , treated with Et₃N (20 ml), stirred for 15 min, and filtered through Celite (washing thoroughly with 500 ml of CH₂Cl₂). Evaporation and FC (AcOEt/ hexane 5:9) gave 29 (10 g, 82%). R_f (AcOEt/hexane 2:3) 0.42. M.p. 84.2°. $\left[\alpha\right]_D^{25} = -30.9$ ($c = 1.0$, CHCl3). IR (CHCl3): 3028m, 2960w, 2877w, 2114w, 1756s, 1601w, 1519w, 1491s, 1460w, 1415m, 1363s, 1162m, 1086s, 1051s, 1017s, 924m, 910w, 843s. ¹H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY and a HSQC spectrum): see Table 9; additionally, $7.38-6.95$ (m, 25 arom. H); 5.03 (d, $J=11.8$), 4.96 (d, $J=11.9$), 4.80 (d, $J=11.5$), 4.66 (d, $J=10.9$), 4.60 (d, $J=11.9$), 4.55 (d, $J=11.4$) (6 PhCH); 4.53 (br. d, J=11.2, 2 PhCH); 4.51 (br. d, J=11.3, 2 PhCH); 4.33 (d, J=12.2), 4.28 (d, $J=12.0$) (2 PhCH); 2.07, 2.01, 2.00, 1.99, 1.98, 1.95, 1.89 (7s, 7 AcO); 0.21 (s, Me₃Si). ¹³C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see *Table 10*; additionally, 170.45, 170.16, 170.11, 169.68, 169.40, 169.29, 168.97 (7s, 7C=O); 137.73, 137.60, 137.56, 136.55, 136.32, 136.15 (6s);

	28	31	$32^{\mathrm{a}})$	33^{b})	$35^{\rm a})^{\rm b}$)
					E chain, B chain
$C\equiv CH$	2.49	2.24			
$H - C(1^1)$	4.14	4.21	4.54	$4.54 - 4.51$	4.60, 4.40
$H - C(2^1)$	5.13	5.12	5.30	$5.28 - 5.23$	$5.26 - 5.24$, $5.15 - 5.08$
$H - C(3I)$	5.12	5.11	5.23	$5.28 - 5.23$	$5.26 - 5.24, 5.17$
$H - C(4^1)$	3.75	3.75	3.84	$3.85 - 3.80$	$3.81 - 3.72, 3.82$
$H - C(5^{\rm I})$	3.63	3.64	3.69	3.69	$3.71, 3.67 - 3.55$
$H_a-C(6^I)$	4.47	4.47	4.56	$4.54 - 4.51$	4.56, 4.49
$H_h - C(6^1)$	4.09	4.093	4.13	4.13	$4.15 - 4.06$
$H - C(1H)$	4.44	4.44	4.46	4.45	4.46, 4.54
$H - C(2H)$	4.82	4.815	4.83	4.84	$4.86 - 4.80$
$H-C(3H)$	5.10	5.10	5.13	5.13	$5.15 - 5.08$
$H - C(4H)$	3.74	3.74	3.77	3.77	$3.81 - 3.72$
$H-C(5II)$	3.56	3.56	3.57	3.57	$3.67 - 3.55$
$H_a-C(6^{\text{II}})$	4.40	4.40	4.42	$4.43 - 4.41$	4.41
$Hb-C(6H)$	4.08	4.089	4.11	4.11	$4.15 - 4.06$
$H-C(1III)$	4.46	4.46	4.48	4.48	$4.492 - 4.485$
$H - C(2III)$	4.90	4.90	4.90	4.91	$4.902 - 4.897$
$H - C(3III)$	5.09	5.05	5.12	5.12	$5.15 - 5.08$
$H - C(4III)$	3.74	3.73	3.76	3.77	$3.81 - 3.72$
$H - C(5III)$	3.58	3.58	3.64	3.63	$3.67 - 3.55$
$H_a-C(6III)$	4.39	4.39	4.41	$4.43 - 4.41$	4.41
$H_b-C(6III)$	4.06	4.05	4.10	4.10	$4.15 - 4.06$
$H - C(1^{\rm IV})$	4.48	4.48	4.51	4.51	4.50/4.495
$H - C(2^{\text{IV}})$	4.83	4.82	4.86	4.86	$4.86 - 4.80$
$H - C(3IV)$	5.07	5.10	5.11	5.11	$5.15 - 5.08$
$H - C(4^IV)$	5.05	5.05	5.05	5.05	5.05
$H - C(5IV)$	3.57	3.57	3.60	3.61	$3.67 - 3.55$
$H_a-C(6^{\rm IV})$	4.35	4.35	4.35	4.36	4.36/4.35
$H_b-C(6^{\rm IV})$	4.03	4.03	4.04	4.04	$4.05 - 4.00$
$J(1^I,\equiv CH)$	2.2	0.9	$\overline{}$		$\overline{}$
$J(1^{\rm I},2^{\rm I})$	9.6	9.8	9.8	$\mathbf{c})$	9.7, 9.6
$J(2^{\rm I},3^{\rm I})$	8.8	9.3	9.3	c)	$c)$, 9.2
$J(3^I,4^I)$	9.3	9.3	9.3	c)	$^{\circ}$), 9.2
$J(4^{\rm I},5^{\rm I})$	9.8	9.8	9.8	9.9	9.9, 9.8
$J(5^{\rm I},6{\rm a}^{\rm I})$	1.9	2.0	1.9	1.8	$(1.9, \degree)$
$J(5^{\rm I},6{\rm b}^{\rm I})$	4.8	4.7	5.0	5.2	4.8, \degree)
$J(6a^{\rm I}, 6b^{\rm I})$	12.1	12.0	12.3	12.1	$12.1,$ °)
$J(1^{\rm II},2^{\rm II})$	7.9	7.9	7.9	7.9	7.8/7.8
$J(2^{\text{II}}, 3^{\text{II}})$	9.3	9.3	9.3	9.3	$\mathbf{c})$
$J(3^{\text{II}},4^{\text{II}})$	9.2	9.2	9.3	9.4	c)
$J(4^{\text{II}},5^{\text{II}})$	9.6	9.6	9.8	9.8	$\mathbf{c})$
$J(5^{\text{II}},6\text{a}^{\text{II}})$	2.4	2.4	1.9	2.1	1.9
$J(5^{\text{II}},6b^{\text{II}})$	4.8	5.1	5.1	5.0	$\mathbf{c})$
$J(6a^{\text{II}}, 6b^{\text{II}})$	12.1	12.2	12.1	12.1	12.1
$J(1^{\text{III}},2^{\text{III}})$	7.9	7.9	7.9	7.9	7.8/7.8
$J(2^{III}, 3^{III})$	9.3	9.3	9.3	9.3	9.5/9.3

Table 11. Selected ¹H-NMR Chemical-Shift Values [ppm] and Coupling Constants [Hz] of the Peracetylated Cellotetraosylalkynes $28, 31, 32, 33,$ and 35 in CDCl₃

^a) Assignments based on a DQFCOSY and a HSQC spectrum. ^b) H–C(2¹) and H–C(3¹) of 33 and the E chain of 35 form a narrow AB system leading to virtual couplings with H–C(1¹) and H–C(4¹). ^c) Not assigned.

133.67, 133.48, 133.40, 133.05, 132.89 (5s); 129.62 – 127.83 (several d); 74.68, 74.43, 74.14, 74.09, 73.38, 72.62 (6t, 6 PhCH₂); 20.63, 20.55, 20.51, 20.49 (4q, 7 Me); -0.50 (q, Me₃Si). HR-MALDI-MS: 1797.4162 (45, $[M+Na]^+$, $C_{87}H_{95}Cl_5NaO_{27}Si^+$; calc. 1797.4170). Anal. calc. for $C_{87}H_{95}Cl_5O_{27}Si$ (1778.04): C 58.77, H 5.39; found: C 58.69, H 5.48.

 $2,3,4,6$ -Tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 4)$ -6-O-benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 8)-5,9-anhydro-6,7,10-tris-O-(4-chlorobenzyl)-1,2,3,4-tetradeoxy-D-glycero-D-gulo-deca-1,3-diynitol (30). A soln. of 29 (1.0 g, 0.56 mmol) in THF (40 ml) was cooled to 0° , treated with a soln. of Bu₄NF·3 H₂O (53 mg, 0.17 mmol) in THF (10 ml), stirred for 5 min, treated with H₂O (10 ml), and warmed to r.t. Workup and FC (AcOEt/hexane 2:3) gave 30 (0.90 g, 94%) as a foamy pale yellow solid, which turned pink upon standing. It was immediately used for the next step. R_f (AcOEt/hexane 2:3) 0.26. M.p. 84.3–88.6°. $[\alpha]_D^{25} = -21.9$ ($c = 1.0$, CHCl3). IR (CHCl3): 3305w, 3028w, 2959w, 2877w, 2114w, 1757s, 1601w, 1519w, 1491s, 1460w, 1421m, 1359s, 1162m, 1086s, 1057s, 1016s, 929m, 878w, 843s. ¹H-NMR (500 MHz, CDCl₃): see Table 9; additionally, 7.8–6.95 (m, 25 arom. H); 5.04 (d, J=12.3), 4.96 (d, J=12.3), 4.78 (d, J=10.8), 4.67 (d, J=10.8), 4.60 $(d, J=11.9)$, 4.54 $(d, J=11.4)$, 4.52 $(d, J=10.5)$, 4.51 $(d, J=12.1)$ (8 PhCH); 4.50 $(d, J=11.8, 2$ PhCH); 4.34 (d, J=12.1), 4.27 (d, J=12.0) (2 PhCH); 2.07, 2.02, 2.00, 1.99, 1.98, 1.96, 1.89 (7s, 7 AcO). 13C-NMR (125 MHz, CDCl₃): see Table 10; additionally, 170.49, 170.20, 170.13, 169.71, 169.43, 169.31, 168.99 (7s,7C=O); 137.71, 137.57, 137.513, 136.511, 136.21, 136.00 (6s); 133.80, 133.50, 133.40, 133.10, 132.90 (5s); 129.62-127.83 (several d); 74.76, 74.45, 74.15, 74.11, 73.37, 72.61 (6t, 6 PhCH₂); 20.65, 20.58, 20.53, 20.51 (4q, 7 Me). HR-MALDI-MS: 1725.3758 (49, $[M + Na]^+$, C₈₄H₈₇Cl₅NaO₂₇; calc. 1725.3775). Anal. calc. for C₈₄H₈₇Cl₅O₂₇ (1705.86): C 59.14, H 5.14; found: C 58.97, H 5.36.

 $2,3,4,6$ -Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl]₂- $(1 \rightarrow 8)$ -6,7,10-tri-O-acetyl-5,9-anhydro-1,2,3,4-tetradeoxy-D-glycero-D-gulo-deca-1,3-diynitol (31). A soln. of 30 $(5.00 \text{ g}, 2.90 \text{ mmol})$ in Ac₂O (50 ml) was cooled to -40° , treated with a soln. of TMSOTf (23.2 ml, 58) mmol) in Ac₂O (20 ml), stirred for 3 h, warmed to -20° , stirred for 24 h, warmed to -10° , stirred for 16 h, warmed to -5° , and stirred for 48 h. After cooling to -10° , the mixture was treated portionwise with sat. aq. NaHCO₃ soln. (40 ml) and stirred for 0.5 h. Workup (AcOEt), evaporation, and FC (AcOEt/hexane 1:1 \rightarrow 3:2) gave a pale yellow solid (2.9 g), which was recrystallised in AcOEt/hexane to afford 31 (2.38 g, 65%). R_f (AcOEt/hexane 3:2) 0.22. M.p. 132.3° (dec.). $[a]_D^{25} = -19.0$ ($c = 1.0$, CHCl₃). IR (CHCl₃): 3305w, 3028w, 2969w, 2888w, 2114w, 1757s, 1520w, 1427m, 1369m, 1160m, 1046s, 924m, 878w,

Table 11 (cont.)

	28	31	$32^{\rm a}$)	33	$35^{\rm a}$)	$34a$)	$36a$)
					E chain, B chain		E chain, B chain
C(4')		66.83			$-$, 78.74		$-.78.85$
C(3')		69.86			$-$, 78.29		$-$, 77.33
C(2')	75.38	69.60	85.65	84.94	85.92, 71.97	84.01	84.23, 69.67
C(1')	77.80	71.13	90.16	89.93	90.24, 79.32	94.60	94.04, 84.23
$C(1^I)$	68.30	68.66	69.38	69.34	69.48, 69.08	70.83	70.72, 70.41
$C(2^I)$	71.27	70.99	71.53	71.35	71.20/71.12	73.53	73.57, 73.13
$C(3^I)$	73.02	72.96	73.44	73.11	73.25/73.17	75.91	75.79, 75.67
$C(4^I)$	76.20	76.14	76.29	76.40	76.47/76.34b)	80.11	79.95/79.81
$C(5^I)$	76.89	76.95	76.98	76.88	77.05/76.79	79.00	78.93, 78.80
$C(6^I)$	62.11	62.10c	62.15	62.31°)	62.31/62.20	60.31	60.17
C(1 ^{II})	100.53°)	100.50	100.54	100.54	100.67/100.64	102.71	102.64/102.61
$C(2^{II})$	$71.86d$)	$71.83d$)	71.95c	71.91 ^d	71.92c	$73.06b$)	72.97/72.93
C(3 ^H)	72.80°)	$72.87e$)	$72.91d$)	72.91°)	$72.94 - 72.66$	74.76c	74.64/74.61
C(4 ^H)	76.15 ^f	76.06	76.21°)	76.27 ^f	$76.28/76.13b$)	$80.41d$)	80.21/80.18
C(5 ^H)	72.64°)	72.62^e)	$72.68d$)	72.68°)	$72.94 - 72.66$	74.83	74.71
C(6 ^H)	62.06	62.06°)	62.15	62.20°)	$62.17d$)	60.31	60.20
C(1 ^{III})	100.52°)	100.50	100.54	100.47	100.53	102.82	102.72
C(2 ^{III})	$71.74d$)	$71.73d$)	$71.80c$)	$71.76d$)	71.82c	$72.98b$)	72.89
C(3 ^{III})	72.75°)	72.79e	72.84 ^d	72.80°)	$72.94 - 72.66$	74.83c	74.71
C(4 ^{III})	76.07 ^f	76.06	76.10°)	76.11 ^f	$76.13/76.11b$)	$80.34d$)	80.28
$C(5^{III})$	$72.57e$)	72.55°)	72.64 ^d	72.64°)	$72.94 - 72.66$	74.83	74.71
$C(6^{III})$	62.06	61.92°)	62.10	62.09°)	$62.09d$)	60.31	60.20
$C(1^{\rm IV})$	100.81	100.79	100.82	100.83	100.84/100.82	103.25	103.13
$C(2^{\rm IV})$	$71.56d$)	$71.56d$)	71.61°)	$71.57d$)	71.62	73.25	73.16
$C(3^{\rm IV})$	72.88°)	72.76°)	$72.81d$)	72.76°)	$72.94 - 72.66$	76.47	76.37
$C(4^{\rm IV})$	67.73	67.73	67.79	67.74	67.79	70.04	69.94
$C(5^{\rm IV})$	72.04	72.03	72.07	72.07	72.06	76.80	76.69
$C(6^{\rm IV})$	61.50	61.50	61.54	61.51	61.55	61.04	60.93

Table 12. Selected 13 C-NMR Chemical-Shift Values [ppm] of the Peracetylated Cellotetraosylalkynes 28, 31, 32, 33, and 35 in CDCl₃, and of the Deprotected Cellotetraosylalkynes 34 and 36 in $(D_6)DMSO$

^a) Assignments based on a HSQC spectrum. ^b)^c)^d) Assignments may be interchanged.

849w. ¹H-NMR (500 MHz, CDCl₃): see *Table 11*; additionally, 2.142, 2.140, 2.13, 2.09, 2.06, 2.03, 2.02, 2.01, 2.00, 1.99, 1.983, 1.979, 1.96 (13s, 13 AcO). ¹³C-NMR (125 MHz, CDCl₃): see *Table 12*; additionally, 170.51, 170.34, 170.23, 169.80, 169.77, 169.74, 169.35, 169.30, 169.12 (9s, 9C=O); 170.20, 169.32 (2s, 4 C=O); 20.86, 20.66, 20.58, 20.48, 20.47 (5q, 5 Me); 20.79 (q, 2 Me); 20.55, 20.53 (2q, 6 Me). HR-MALDI-MS:1267.3526 (100, $[M + Na]^+$, $C_{54}H_{68}NaO_{33}^+$; calc. 1267.3541). Anal. calc. for $C_{54}H_{68}O_{33}$ (1245.11): C 52.09, H 5.50; found: C 52.11, H 5.66.

 $2,3,4,6$ -Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl]₂- $(1 \rightarrow 6)$ -4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-1-C-(8-hydroxy-9,10-dioxoanthracen-1-yl)-D-glycero-D-gulooct-1-ynitol (32). Under Ar, a stirred suspension of 20 (0.50 g, 1.20 mmol), $[Pd(PPh₃)₂]C₂$ (42 mg, 0.06 mmol), CuI (34 mg, 0.18 mmol), and Bu₄NI (0.67 mg, 1.81 mmol) in degassed Et₃N/DMF 1:5 (5 ml) was treated dropwise with a soln. of 28 (1.9 g, 1.6 mmol) in Et₃N/DMF 1:5 (5 ml) over a period of 9 h at 24°. After stirring for 24 h, the mixture was treated with solid (NH₄)₂CO₃ (0.50 g, 6.40 mmol), and stirred for 12 h. Workup (AcOEt), evaporation, FC (AcOEt/hexane/CH₂Cl₂ 2 : 3 : 3), and recrystallisation in CH₂Cl₂/MeOH gave 32 (1.3 g, 75%). Bright yellow solid. R_f (AcOEt/hexane 7:4) 0.27. M.p. 217.1^o. $\left[\alpha\right]_{\text{D}}^{25} = -8.9$ (c=1.0, CHCl₃). IR (CHCl₃): 3476w, 3025m, 2978m, 2897w, 1755s, 1674w, 1640w, 1575w,

1518m, 1477w, 1426m, 1368m, 1315w, 1276w, 1044s, 928m, 876w, 848w. ¹H-NMR (500 MHz, CDCl₃; assignments based on a DOFCOSY and a HSOC spectrum): see Table 11; additionally, 12.49 (s, OH); 8.32 (dd, $J=7.8$, 1.4, H-C(4')); 7.89 (dd, $J=7.8$, 1.4, H-C(2')); 7.81 (dd, $J=7.5$, 1.2, H-C(5')); 7.73 $(t, J=7.8, H-C(3'))$; 7.67 (dd, $J=8.2, 7.7, H-C(6'))$; 7.32 (dd, $J=8.4, 1.2, H-C(7'))$; 2.16, 2.155, 2.151, 2.10, 2.09, 2.039, 2.037, 2.03, 2.02, 2.01, 2.00, 1.98, 1.97 (13s, 13 AcO). ¹³C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see Table 12; additionally, 187.34 (s, C(9')); 181.80 (s, C(10')); 170.49, 170.39, 170.19, 169.92, 169.71, 169.34, 169.31, 169.30, 169.10 (9s, 9C=O); 170.21, 169.74 (2s, 4C=O); 162.60 (s, C(8')); 140.88 (d, C(2')); 136.61 (d, C(6')); 134.49 (s, C(9'a)); 133.47 (d, C(3')); 133.37 (s, $C(4'a)$); 132.75 (s, $C(10'a)$); 128.24 (d, $C(4')$); 124.82 (d, $C(7')$); 122.05 (s, $C(1')$); 119.34 (d, $C(5')$); 116.33 (s, C(8'a)); 20.94, 20.89, 20.82, 20.80, 20.66, 20.62, 20.56, 20.51, 20.48 (9q, 9 Me); 20.59, 20.53 $(2q, 4 \text{ Me})$. HR-MALDI-MS: 1465.3852 (86, $[M + Na]^+$, $C_{66}H_{74}NaO_{36}^+$; calc. 1465.3858). Anal. calc. for $C_{66}H_{74}O_{36}$ (1443.29): C 54.92, H 5.17; found: C 54.78, H 5.07.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl]₂- $(1 \rightarrow 6)$ -4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-1-C-{8-[(trifluoromethyl)sulfonyloxy]-9,10-dioxoanthracen-1 y l/-D-glycero-D-gulo-*oct-1-ynitol* (33). A stirred soln. of 32 (1.2 g, 0.83 mmol) in dry CH₂Cl₂ (19.2 ml) was treated with Et₃N (0.23 ml, 1.65 mmol) and cooled to -78° . The resulting red suspension was treated with freshly distilled Tf₂O (0.18 ml, 1.09 mmol), stirred for 1 h, allowed to warm to 0° , and diluted with CH₂Cl₂ (50 ml). Workup, evaporation, and recrystallisation in AcOEt/hexane gave 33 (1.24 g, 95%). Pale yellow solid. R_f (AcOEt/hexane 3:2) 0.22. M.p. 212.2–220.9°. [a] $_D^{25}$ = +9.6 (c = 1.0, CHCl₃). IR (CHCl₃): 3009*m*, 2974m, 2882w, 1755s, 1679m, 1598w, 1518m, 1477w, 1425m, 1367m, 1321m, 1136w, 1044s, 928m, 8 76w, 853w. ¹H-NMR (500 MHz, CDCl₃): see *Table 11*; additionally, 8.37 (dd, $J=7.8$, 1.2, H-C(4')); 8.28 $(dd, J=7.8, 1.3, H-C(5'))$; 7.89 $(dd, J=7.8, 1.3, H-C(2'))$; 7.85 $(t, J=8.0, H-C(6'))$; 7.73 $(t, J=7.8, 1.3)$ $H-C(3')$; 7.63 (dd, J = 8.4, 1.2, $H-C(7')$); 2.16, 2.155, 2.151, 2.11, 2.10, 2.04, 2.01, 2.00, 1.99, 1.98, 1.97 (11s, 11 AcO); 2.03 (s, 2 AcO). ¹³C-NMR (125 MHz, CDCl₃): see *Table 12*; additionally, 181.10 (s, $C(9')$; 179.83 (s, $C(10')$); 170.51, 170.48, 170.25, 170.23, 170.20, 170.01, 169.80, 169.36, 169.12 (9s, 9 C= O); 169.76, 169.31 (2s, 4 C=O); 147.48 (s, C(8')); 140.97 (d, C(2')); 134.92 (s, C(9'a)); 134.78 (d, C(6')); 134.44 (s, C(10'a)); 133.20 (d and s, C(3'), C(4'a)); 128.88 (d, C(4')); 127.71 (d, C(7')); 127.63 (d, $C(5')$); 126.60 (s, $C(8'a)$); 122.30 (s, $C(1')$); 118.77 (q, ¹J(C,F)=320.9, CF₃); 20.87, 20.72, 20.68, 20.62, 20.60, 20.59, 20.57, 20.51, 20.49 (9q, 9 Me); 20.81, 20.54 (2q, 4 Me). ¹⁹F-NMR (282 MHz, CDCl₃): -73.64 (s, CF₃). HR-MALDI-MS: 1597.3336 (100, $[M+Na]^+$, C₆₇H₇₃F₃NaO₃₈S⁺; calc. 1597.3350). Anal. calc. for $C_{67}H_{73}F_3O_{38}S.2 H_2O$ (1610.37): C 49.94, H 4.82; found: C 50.16, H 4.80.

 β -D-Glucopyranosyl-[(1 \rightarrow 4)- β -D-glucopyranosyl]₂-(1 \rightarrow 6)-3,7-anhydro-1,2-dideoxy-1-C-(8-hydroxy-9,10-dioxoanthracen-1-yl)-D-glycero-D-gulo-oct-1-ynitol (34). Under Ar, a stirred suspension of 32 (285 mg, 0.20 mmol) in dry MeOH (21 ml) was treated with a soln. of MeONa (0.296 mmol) in MeOH (21 ml), and stirred for 6.5 h at 24°. The red suspension was treated with H₂O (15 ml, \rightarrow clear soln.), stirred for 4 h, neutralised with *Amberlite IR-120* resin (H^+ form), and filtered. Evaporation of the filtrate gave 34 (169 mg, 95%). Yellow solid. R_f (RP-18 silica gel; MeCN/H₂O 2:1) 0.38. M.p 154.5° (dec., \rightarrow green residue). $\left[\alpha\right]_D^{25} = +7.6$ (c=1.0. DMSO). ¹H-NMR (500 MHz, (D₆)DMSO; assignments based on a DQFCOSY, a TOCSY, and a HSQC spectrum): see *Table 1*; additionally, 12.47 (br. s, $HO-C(8')$); 8.23 (d, J=7.7, H-C(4')); 8.00 (d, J=7.5, H-C(2')); 7.91 (t, J=7.7, H-C(3')); 7.80 (t, J=7.8, H- $C(6')$); 7.70 (d, J = 7.4, H – $C(5')$); 7.40 (d, J = 8.2, H – $C(7')$); 4.39 (d, J = 7.8, H – $C(1^H)$); 4.35 (d, J = 7.9, $H-C(1^{III})$); 4.31 (br. d, J=7.6, H-C(3¹)); 4.26 (d, J=7.8, H-C(1^{IV})); 3.83–3.78 (m, H-C(8¹), H- $C(6^{II-III})$); 3.71 (br. *d*, *J*=9.7, H-C(6^{IV})); 3.66–3.58 (m, H'-C(8^I), H'-C(6^{II-III})); 3.44–3.30 (m, H- $C(4^l)$, H-C(5¹), H-C(6¹), H-C(7¹), H-C(3^{II-III}), H-C(4^{II-III}), H-C(5^{II-III}), H'-C(6^{IV})); 3.24-3.12 (*m*, $H-C(3^{IV})$, $H-C(5^{IV})$; 3.13–3.03 (m, $H-C(2^{II-III})$, $H-C(4^{IV})$); 3.03–2.95 (m, $H-C(2^{IV})$). ¹³C-NMR (125 MHz, (D_6) DMSO; assignments based on a HSQC spectrum): see Table 12; additionally, 186.99 (s, $C(9')$); 181.55 (s, $C(10')$); 161.51 (s, $C(8')$); 141.06 (d, $C(2')$); 136.83 (d, $C(6')$); 134.17 (s, $C(9'a)$); 134.05 (d, C(3')); 132.72 (s, C(4'a)); 132.65 (s, C(10'a)); 127.39 (d, C(4')); 124.35 (d, C(7')); 122.17 (d, $C(1')$); 118.50 (d, $C(5')$); 116.38 (s, $C(8'a)$). MALDI-TOF: 919 ([M+Na]⁺, C₄₀H₄₈NaO₂₃; calc. 919). Anal. calc. for C₄₀H₄₈O₂₃ · 3 MeOH (992.92): C 52.01, H 6.09; found: C 51.93, H 5.98.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl]₂- $(1 \rightarrow 8)$ -6,7,10-tri-O-acetyl-5,9-anhydro-1,2,3,4-tetradeoxy-1-C-(8-{2,3,4,6-tetra-O-acetyl-b-D-glucopyranosyl-[(1 \rightarrow 4)-2,3,6-tri-O-acetyl- β -D-glucopyranosyl] \rightarrow (1 \rightarrow 6)-4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-D-glycero-D-gulo-oct-1-ynitol-1-yl}-9,10-dioxoanthracen-1-yl)-D-glycero-D-gulo-deca-1,3-diynitol (35). Under Ar, a suspension of 33 (300 mg, 0.19 mmol), $[Pd(PPh_3),]C1$, (6.7 mg, 0.01 mmol), CuI (5.5 mg, 0.03 mmol), and Bu₄NI (105 mg, 0.28 mmol) in degassed Et₃N/DMF 1:5 (3 ml) was treated dropwise with a soln. of 31 (0.80 mmol) in Et₃N/DMF 1:5 (3 ml) over a period of 12 h at 24 $^{\circ}$ and stirred for 12 h. Workup (AcOEt), evaporation, and FC (CH₂Cl₂/AcOEt 5:1 \rightarrow 2:1) gave 35 (303 mg, 59%) as a light yellow solid, which was recrystallised in AcOEt/MeOH. R_f (AcOEt/hexane 4:1) 0.40. M.p. $262.3 - 265.4^{\circ}$. $[\alpha]_D^{25} = +0.3$ (c=1.0, CHCl₃). IR (CHCl₃): 3023m, 2976m, 2893w, 1754s, 1685w, 1598w, 1517m, 1476w, 1425m, 1367m, 1337w, 1316w, 1043s, 928m, 876w, 846w. ¹H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY and a HSQC spectrum): see Table 11; additionally, 8.32 (dd, $J=7.8, 1.3$), 8.29 (dd, $J=7.8, 1.4$) (H-C(4'), H-C(5')); 7.94 (dd, $J=7.8, 1.3$), 7.92 (dd, $J=7.5, 1.4$) $(H-C(2'), H-C(7'))$; 7.71, 7.70 $(2t, J=7.8, H-C(3'), H-C(6'))$; 2.19, 2.17, 2.15, 2.14, 2.12, 2.042, 2.038, 2.037, 2.034, 2.00, 1.996, 1.981, 1.980, 1.97, 1.96 (15s, 15 AcO); 2.155, 2.09, 2.03, 2.005 (4s, 8AcO); 2.007 (s, 3 AcO). ¹³C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see Table 12; additionally, 181.77 (s, C(10')); 179.78 (s, C(9')); 170.40, 170.29, 169.81, 169.77, 169.72, 169.46, 169.32, 169.30 (8s,8C=O); 170.49, 169.75, 169.74, 169.37, 169.31, 169.11 (6s, 12 C=O); 170.23, 170.18(2s, 6 C=O); 142.16 (d, C(2')); 141.50 (d, C(7')); 135.37 (s, C(9'a)); 133.73 (s, C(8'a)); 133.69 (s, C(4'a)); 133.50 (s, C(10'a)); 132.89 (d, C(3')); 132.80 (d, C(6')); 128.14 (d, C(4')); 127.78(d, C(5')); 122.05 (s, C(8')); 121.49 (d, C(1')); 20.99, 20.93, 20.90, 20.82, 20.80, 20.73, 20.63, 20.56 (8q, 8Me); 20.67, 20.59, 20.49 (3q, 6 Me); 20.57, 20.56, 20.53, 20.51 (4q, 12 Me). HR-MALDI-MS: 2691.7395 (47, $[M + Na]^+$, C₁₂₀- $H_{140}NaO_{68}^+$; calc. 2691.7395). Anal. calc. for $C_{120}H_{140}O_{68}$ (2670.39): C 53.97, H 5.28; found: C 53.92, H 5.36.

 β -D-Glucopyranosyl-[(1 \rightarrow 4)- β -D-glucopyranosyl]₂-(1 \rightarrow 8)-5,9-anhydro-1,2,3,4-tetradeoxy-1-C-(8- ${\beta}$ -D-glucopyranosyl- $[(1 \rightarrow 4)$ - β -D-glucopyranosyl I_2 - $(1 \rightarrow 6)$ -3,7-anhydro-1,2-dideoxy-D-glycero-D-gulooct-1-ynitol-1-yl}-9,10-dioxoanthracen-1-yl)-D-glycero-D-gulo-deca-1,3-diynitol (36). Under Ar, a suspension of 35 (0.10 g, 0.04 mmol) in 0.1M aq. Bu₄NOH (11 ml) was stirred for 20 h, and neutralised with Amberlite IR-120 resin $(H^+$ form). The supernatant soln. was decanted from the resin and centrifuged (6000 rpm). The soln. was diluted with H_2O and lyophilised to afford 36 (46 mg, 78%). Yellowish brown solid). R_f (RP-18, silica gel; MeCN/H₂O 2:1) 0.50. M.p. 262.9–265.1° (dec.). $[\alpha]_D^{25} = -8.6$ $(c=1.0. \text{ DMSO})$. ¹H-NMR (500 MHz, (D_6) DMSO; assignments based on a DQFCOSY, a TOCSY, and a HSQC spectrum): see *Table 1*; additionally, 8.26 (dd, $J=7.7, 1.3$), 8.23 (dd, $J=7.7, 1.3$) (H- $C(4')$, H- $C(5')$); 8,13 (dd, J=7.8, 1.3), 8.03 (dd, J=7.8, 1.3) (H- $C(2')$, H- $C(7')$); 7.90 (t, J=7.8), 7.89 (t, J=7.9) (H–C(3'), H–C(6')); 4.39 (d, J=7.8, H–C(1^{IIIE})); 4.36 (d, J=7.8, H–C(1^{IIE})); 4.331 (d, $J=7.5, H-C(1^{IB}), H-C(1^{IIIB})),$ 4.329 (d, $J=9.1, H-C(3^{IE}),$ 4.25 (d, $J\approx 8.4, H-C(1^{IVB}), H-C(1^{IVE}))$; 4.23 (d, J \approx 10.0, H – C(5^{IB})); 3.82 – 3.75 (m, H – C(8^{IE}), H – C(10^{IB}), H – C(6^{IIB–IIIB}), H – C(6^{IIE–IIIE})); $3.72-3.65$ (m, H-C(6^{IVB}), H-C(6^{IVE})); $3.64-3.55$ (m, H'-C(8^{IE}), H'-C(10^{IB}), H'-C($6^{\text{IIB-IIIB}}$), H'- $C(6^{IIE-IIE})$); 3.48–3.28 (m, H-C(4^{IE}), H-C(5^{IE}), H-C(6^{IE}), H-C(7^{IE}), H-C(6^{IB}), H-C(7^{IB}), H- $C(8^{1B})$, H-C(9^{IB}), H-C(3^{IIB–IIIB}), H-C(3^{IIE–IIIE}), H-C(4^{IIB–IIIB}), H-C(4^{IIE–IIIE}), H-C(5^{IIB–IIIB}), H- $C(5^{H\acute{E}-H\acute{E}}), H'-C(6^{IVB}), H'-C(6^{IVE}))$; 3.25–3.12 (m, H-C(3^{IVB}), H-C(3^{IVE}), H-C(5^{IVB}), H-C(5^{IVE})); $3.12-2.90$ (m, H-C($2^{\text{IIB-IV}}$), H-C($2^{\text{IIE-IV}}$), H-C(4^{IVB}), H-C(4^{IV}). ¹³C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see Table 12; additionally, 181.44 (s, $C(10')$); 179.58 (s, $C(9')$; 141.96 (d, $C(2')$); 141.71 (d, $C(7')$); 135.02 (s, $C(9'a)$); 133.49 (s, $C(8'a)$); 133.37 (s, $C(4'a)$); 133.27 (d, C(3')); 133.22 (d, C(6')); 133.09 (s, C(10'a)); 127.61 (d, C(4')); 127.00 (d, C(5')); 122.08(s, $C(8')$; 120.19 (s, C(1')). MALDI-TOF: 1599 ([M+Na]⁺, C₆₈H₈₈NaO₄₂; calc. 1599). Anal. calc. for $C_{68}H_{88}O_4$, 4 H₂O (1649.46): C 49.51, H 5.87; found: C 49.33, H 6.39.

Allyl 2,3,4,6-tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 4)$ 4)-2,3,6-tri-O-benzyl- β -D-glucopyranosyl- $(1 \rightarrow 4)$ -2,3,6-tri-O-benzyl- β -D-glucopyranoside [30] (39). A suspension of 38 [30] [34] [36] (5.0 g, 5.4 mmol) and 4-Å mol. sieves in CH₂Cl₂ (190 ml) was stirred at 24° for 1 h and treated with 37 [61] (5.0 g, 6.4 mmol). The mixture was cooled to -40° , stirred for 15 min, treated dropwise with a soln. of TMSOTf (0.23 ml, 1.28 mmol) in CH₂Cl₂ (10 ml), stirred for 2 h, diluted with CH₂Cl₂, allowed to warm to r.t., and filtered through *Celite*. The filtrate was washed with aq. NaHCO₃ soln. Workup, evaporation, and FC (AcOEt/hexane $1:3 \rightarrow 1:2$) gave 39 (7.3 g, 87%). Foamy solid. R_f (AcOEt/hexane 1:1) 0.57. M.p. 230–232° ([30]: 232–234°).

Allyl 2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ 4)-2,3,6-tri-O-benzyl- β -D-glucopyranosyl[$\frac{1}{3}$ - $(1 \rightarrow 4)$ -2,3,6-tri-O-benzyl- β -D-glucopyranoside (44). A sus-

pension of 43 [30] (8.0 g, 4.47 mmol) and 4-Å mol. sieves (8.0 g) in dry CH₂Cl₂ (110 ml) was stirred at 25° for 2 h, treated with 37 (4.2 g, 5.39 mmol), cooled to -60° , stirred for 30 min, treated dropwise with a soln. of TMSOTf (0.16 ml, 0.9 mmol) in CH₂Cl₂ (8 ml), stirred for 12 h, treated with solid NaHCO₃ (0.8 g), stirred for 30 min, warmed to 25° , and filtered through *Celite*. The filtrate was washed with sat. aq. NaHCO₃ soln. Workup and FC (AcOEt/cyclohexane $1:4 \rightarrow 1:2$) gave 44 (10.1 g, 94%). Foam. R_f $(ACOE$ t/hexane 2:3) 0.20. M.p. 71.5–74°. $[a]_D^{25} = -4.4$ (c=1.0, CHCl₃). IR (CHCl₃): 3023*m*, 2872*m*, 1951w, 1881w, 1754s, 1604w, 1595w, 1518m, 1494w, 1454m, 1419m, 1368s, 1309m, 1153s, 1118s, 1066s, 922m, 876w. ¹H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY and a HSQC spectrum): 7.31 – 7.01 (m, 60 arom. H); 5.90 (dddd, J = 17.2, 10.5, 5.9, 5.2, CH=CH₂); 5.28 (dq, J = 17.4, 1.7), 5.14 $(dq, J=10.5, 1.3)$ (CH=CH₂); 5.09 (d, J=11.6), 5.05 (d, J=11.6) (2 PhCH); 5.05 (t, J=9.2, H-C(3^V)); 5.04 (d, J=10.8, PhCH); 5.01 (t, J=9.7, H-C(3^{VI})); 4.92 (d, J=11.4, PhCH); 4.89 (t, J=9.3, H- $C(4^{VI})$); 4.86 (dd, J \approx 9.2, 8.1, H $-C(2^{VI})$); 4.83 (d, J \approx 10.9, PhCH); 4.75 (dd, J = 9.7, 8.1, H $-C(2^{VI})$); 4.68 (d, J=11.5), 4.67 (d, J=11.1) (2 PhCH); 4.66 (br. d, J=11.0, 2 PhCH); 4.64 (br. d, J=10.4, 2 PhCH); 4.61 (d, J=11.0), 4.60 (d, J=11.1), 4.58 (d, J=11.6), 4.54 (d, J=11.6), 4.52 (d, J=12.1) (5 PhCH); 4.48 (d, J=8.1, H-C(1^V)); 4.43 (d, J=11.9, PhCH); 4.42 (d, J=7.8, H-C(1^{IV})); 4.40 (d, $J=8.0, H-C(1^1)$; 4.36 (d, $J=7.7, H-C(1^{II})$; 4.37 – 4.33 (m, $H-C(1^{III-V1})$, 1 allyl. H, 2 PhCH); 4.32 (d, $J=11.2$, PhCH); 4.31 (dd, $J=12.1$, 4.1, $H-C(6^{\circ})$); 4.30 (d, $J=12.1$), 4.20 (d, $J=12.1$), 4.17 (d, $J=12.0$), 4.15 (d, $J=12.0$) (4 PhCH); 4.15 (dd, $J\approx 12.0$, 2.3, H – C(6^{VI})); 4.09 – 4.05 (ddt, $J=13.0$, 5.9, 1.4, 1 allyl. H); 3.96 (dd, $J \approx 11.9$, 2.6, H' $-C(6^V)$); 3.94 (t, $J \approx 9.2$, H $-C(4^L)$); 3.92 (t, $J \approx 9.0$, H $-C(4^{H-HII})$); 3.84 (dd, $J \approx 12.1$, 4.5, H' $-C(6^{VI})$); 3.83 (t, $J \approx 9.1$, H $-C(4^{IV})$); 3.78 (dd, $J \approx 10.9$, 4.1, H $-C(6^{I})$); 3.68 $(dd, J\approx11.0, 3.6, H-C(6^{II}))$; 3.66–3.55 (m, H–C(5^V), H–C(6^{III–IV}), H[–]C(6^{I-III})); 3.62 (t, J \approx 9.4, H– $C(4^V)$; 3.52 (t, J \approx 9.0, H $-C(3^I)$); 3.48 (dd, J \approx 10.9, 3.4, H' $-C(6^{IV})$); 3.39 (t, J \approx 8.6, H $-C(2^I)$, H $C(3^{II})$); 3.34 (t, J \approx 9.0, H $-C(3^{III})$); 3.29 – 3.20 (m, H $-C(2^{II-V})$, H $-C(3^{IV})$, H $-C(5^{I})$); 3.09 (ddd, J = 9.7, 3.4, 1.7, H $-C(5^{II})$); 3.045 (ddd, J = 9.8, 3.5, 1.6, H $-C(5^{III})$); 3.01 (ddd, J = 9.9, 4.3, 2.1, H $-C(5^{VI})$); 2.91 $(ddd, J=9.9, 3.4, 1.8, H-C(5^{IV}))$; 2.03, 1.97, 1.954, 1.946, 1.94, 1.87, 1.85 (7s, 7 AcO). ¹³C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): 170.47, 170.23, 170.18, 169.72, 169.37, 169.29, 168.97 (7s, 7C=O); 139.42, 139.37, 138.26 (3s, 6 arom. C); 138.60, 138.55, 138.42, 138.41, 138.39, 138.02 (6s, 6 arom. C); 134.14 (d, CH=CH₂); 128.52 – 126.99 (several d); 117.13 (t, CH=CH₂); 102.64 $(d, C(1^W))$; 102.54 $(d, C(1^I))$; 102.52 $(d, C(1^H))$; 102.41 $(d, C(1^H))$; 100.81 $(d, C(1^V))$; 99.69 $(d, C(1^V))$; $(83.37\ (d,\,C(3^1));\,83.22\ (d,\,C(3^{II}));\,82.94\ (d,\,C(3^{III}));\,82.82\ (d,\,C(3^{IV}));\,82.04\ (d,\,C(2^1));\,82.02\ (d,\,C(2^{II}));\,$ 81.82 (d, C(2^{IV})); 81.67 (d, C(2^{III})); 77.04 (d, C(4^I)); 76.95 (d, C(4^{IV})); 76.88 (d, C(4^{II})); 76.71 (d, $C(4^{III})$); 76.17 (d, $C(4^{V})$); 75.05, 75.02, 74.98 (3t, 3 PhCH₂); 74.93 (2d), 74.56 (2d) (C(5^{1–IV})); 73.30 (t, PhCH₂); 73.15 (t, 3 PhCH₂); 72.93 (d, C(3^{V–VI})); 72.88 (t, 2 PhCH₂); 72.86 (t, 2 PhCH₂); 72.27 (d, $C(5^{VI})$); 72.05 (d, $C(2^V)$); 71.88 (d, $C(5^V)$); 71.46 (d, $C(2^{VI})$); 70.21 (t, 1 allyl. C); 68.16 (t, $C(6^I)$); 68.04 $(d, C(4^{VI}))$; 67.75 (3t, $C(6^{II-IV})$); 61.82 (t, $C(6^V)$); 61.50 (t, $C(6^{VI})$); 20.67, 20.66, 20.64, 20.53, 20.52 (several q, 7 Me). HR-MALDI-MS: 2427.9859 (49, $[M + Na]^+$, $C_{137}H_{152}NaO_{38}^+$; calc. 2427.9859). Anal. calc. for $C_{137}H_{152}O_{38}$ (2406.69): C 68.37, H 6.37; found: C 68.51, H 6.43.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl $]_4$ - $(1 \rightarrow 4)$ -1,2,3,6-tetra-O-acetyl-a/b-D-glucopyranose (45). A stirred suspension of bis[methyl(diphenyl)phosphine](cycloocta-1,5-diene)iridium(I) hexafluorophosphate (52.7 mg, 0.062 mmol) in dry THF (20 ml) was degassed at 22 $^{\circ}$ and stirred under H₂ for 5 min until the red suspension turned into a pale yellow soln., flushed with Ar, and treated with a soln. of 44 (5.0 g, 2.07 mmol) in THF (50 ml). After stirring for 2 h, the mixture was treated with H₂O (20 ml) and I₂ (1.09 g, 43 mmol), stirred for 3 h, and treated with a chilled 5% aq. Na₂S₂O₃ soln. (29 ml). Workup (AcOEt), evaporation, and FC (AcOEt/cyclohexane 1:4 \rightarrow 1:2) gave the deallylated derivative of 44 (4.1 g, 83%, α/β 45:55). R_f (AcOEt/hexane 1:1) 0.25. ¹³C-NMR (125 MHz, CDCl₃, α/β 45:55): 97.31 (d, C(1¹) of β -anomer); 91.35 (d, C(1¹) of α -anomer). HR-MALDI-MS: 2387.9549 (47, $[M + Na]^+, C_{134}H_{148}NaO_{38}^+$; calc. 2387.9546).

A suspension of the deallylated derivative of 44 (9 g, 3.8mmol) and charcoal (10 g) in AcOEt (200 ml) was stirred for 24 h, and filtered through *Celite*. After evaporation, a soln. of the residue in acetone/ MeOH 1:1 (500 ml) was treated with 20% Pd(OH) $_2$ /C (3.5 g), stirred under 6 bar of H₂ for 3 days, and filtered through *Celite* (washing with 200 ml of pyridine). The combined filtrate and washings were evaporated and co-evaporated with toluene. A soln. of the pale yellow residue $(4.9 g)$ in Ac₂O/pyridine 1:1 (60 ml) was stirred for 24 h at 23°. Evaporation, co-evaporation with toluene, and FC (AcOEt/hexane

 $1:1 \rightarrow 2:1$) gave α/β -D-45 2 : 3 (6.25 g, 90%). White solid. R_f (AcOEt/hexane 4:1) 0.69. M.p. 238.1 – 238.5°. $[\alpha]_{\text{D}}^{25}$ = -12.7 (c=1.0, CHCl₃). IR (CHCl₃): 3028m, 2969m, 2889w, 1751s, 1520m, 1475w, 1421m, 1369s, 1051s, 924m, 878w, 848w. ¹H-NMR (500 MHz, CDCl₃; α/β -D 2:3): 6.24 (d, J=3.7, 0.4 H), 5.65 (d, $J=8.3, 0.6$ H) (H-C(1¹)). ¹³C-NMR (125 MHz, CDCl₃, α/β 2:3): 91.59 (d, C(1¹) of β -anomer); 88.97 (d, C(1¹) of a-anomer). HR-MALDI-MS: 1853.5262 (100, $[M + Na]^+$, $C_{76}H_{102}NaO_{51}^+$; calc. 1853.5280). Anal. calc. for $C_{76}H_{102}O_{51}$ (1831.60): C 49.84, H 5.61; found: C 50.17, H 5.75. For ¹H- and ¹³C-NMR data of α -45, see [63] [64].

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl $]_4$ - $(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl-a/ β -D-glucopyranose [66] (46). A soln. of a/ β -D-45 2:3 (3.41 g, 1.86 mmol) in dry DMF (35 ml) was treated with (NH_4) , CO_3 (0.89 g, 9.28 mmol), stirred at 23° for 36 h, poured into chilled H₂O, and extracted with AcOEt (200 ml). Workup, evaporation, and FC (AcOEt/hexane 1:1 \rightarrow 3:1) gave α/β -**46** 2 : 1 (2.61 g, 78%). White solid. R_f (AcOEt/hexane 4 : 1) 0.24. M.p. 126.8°. [α] $_{\text{D}}^{25}$ = -6.6 (c = 1.0, CHCl₃). IR (CHCl3): 3483w, 3009m, 2974m, 2882w, 1748s, 1672m, 1516w, 1476w, 1424m, 1366s, 1048s, 927m, 8 75w, 850w. ¹H-NMR (500 MHz, CDCl₃, α/β -D-**46** 2:1): 5.49 (t, J=9.7, 0.67 H), 5.21 (t, J=9.4, 0.33 H) (H- $C(3^1)$); 5.37 (t, J = 3.5, 0.67 H), 4.70 (t, J = 8.0, 0.33 H) (H-C(1¹)). HR-MALDI-MS: 1811.5138 (100, $[M+Na]^+$, $C_{74}H_{100}NaO_{50}^+$; calc. 1811.5180). Anal. calc. for $C_{74}H_{100}O_{50}$ (1789.57): C 49.67, H 4.63; found: C 49.80, H 5.82.

 $2,3,4,6$ -Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)-2,3,6$ -tri-O-acetyl- β -D-glucopyranosyl]₄- $(1 \rightarrow 4)-$ 2,3,6-tri-O-acetyl-a/ β -D-glucopyranosyl Trichloroacetamidate [66] (47). A soln. of a/ β -D-46 2 :1 (400 mg, 0.22 mmol) in dry CH₂Cl₂ (8 ml) was treated with Cl₃CCN (0.22 ml, 2.2 mmol) and a soln. of DBU (8.8 μ), 0.05 mmol) in CH₂Cl₂ (2 ml) and stirred for 2 h. Evaporation, FC (silica gel pre-treated with hexane/Et₃N 98:2; AcOEt/hexane 1:1 \rightarrow 2:11), and crystallisation from AcOEt/hexane gave α/β -D-47 94:6 (308 mg, 71%). R_f (AcOEt/hexane 3:1) 0.22. M.p. 126.8°. $[\alpha]_D^{25} = +4.1$ ($c = 1.0$, CHCl₃). IR (CHCl₃): 3345w, 3021m, 2974m, 2882w, 1753s, 1672w, 1516w, 1475w, 1424m, 1366m, 1048s, 927m, 904w, 875w, 849w. ¹H-NMR (500 MHz, CDCl₃, α/β -D-47 94:6; assignments based on a HSOC and a TOCSY spectrum): data of α -D-47: 8.65 (s, NH); 6.48 (d, J=3.8, H-C(1¹)); 5.51 (t, J=9.7, H-C(3¹)); 5.122 (t, J=9.3), 5.103 (t, J=9.2, 2 H), 5.085 (t, J=9.2), 5.083 (t, J=9.2) (H-C(3^{II-VI})); 5.053 (t, J=9.6, H-C(4^{VI})); 5.050 (dd, J = 10.2, 3.9, H – C(2¹)); 4.90 (dd, J = 9.2, 8.0, H – C(2^{V1})); 4.85 (dd, J = 9.2, 7.9, H – C(2^V)); 4.82 (dd, J = 9.2, 7.4), 4.81 (br. dd, J = 9.1, 7.6, 2 H) (H – C(2^{II–IV})); 4.51 (d, J = 7.9, H – C(1^{VI})); 4.47 (d, $J=7.9$, H-C(1^V)); 4.45–4.38 (m, H-C(1^{II–IV}), H-C(6^L^V)); 4.35 (dd, J=12.5, 4.4, H-C(6^{VI})); $4.12-4.06$ $(m, H-C(5^l), H'-C(6^{l-v}))$; 4.03 $(dd, J=12.5, 2.2, H'-C(6^{VI}))$; 3.81 $(t, J=9.7, H-C(4^l))$; 3.76 (t, J = 9.6), 3.75 (t, J = 9.6), 3.735 (t, J = 9.6), 3.727 (t, J = 9.6) (H - C(4^{II-V})); 3.63 (ddd, J = 9.9, 4.3, 2.3, H–C(5^{VI})); 3.58 (ddd, J=9.8, 4.8, 2.0, H–C(5^V)); 3.60–3.54 (m, H–C(5^{II–IV})); 2.143 (s, 4 AcO); 2.115, 2.088, 2.035, 2.025, 2.015, 2.008, 2.002, 1.999, 1.995, 1.982, 1.960 (11s, 11 AcO); 2.021, 1.957 (2s, 4 AcO); data of β -47: 8.69 (s, NH); 5.83 (d, J=7.5, H-C(1¹)); 5.24 (t, J=8.6, H-C(3¹)); 5.19 (dd, $J=8.8, 7.5, H-C(2¹))$. ¹³C-NMR (125 MHz, CDCl₃; α/β -D-**47** 94:6; assignments based on a HSQC and a TOCSY spectrum): data of a-47: 170.49, 170.22, 170.01, 169.81, 169.70, 169.42, 169.34, 169.09 (8s, 8 C=O); 170.18, 169.73, 169.29, 169.27 (4s, 8 C=O); 170.19 (4s, 3 C=O); 160.94 (s, C=NH); 100.82, 100.79, 100.56, 100.53, 100.49 (5d, C(1^{II–VI})); 92.86 (d, C(1¹)); 90.69 (s, CCl₃); 76.19 (d, C(4¹)); 76.09 (br. d, C(4^{II–V})); 72.88, 72.81 (4 C), 72.68, 72.63, 72.54, 72.52 (6d, C(3^{II–VI}), C(5^{II–V})); 72.04 (d, C(5^{VI})); 71.91, 71.84 (2 C), 71.74 (3d, C(2^{II–V})); 71.56 (d, C(2^{VI})); 70.96 (d, C(5^I)); 69.91 (d, C(2^I)); 69.28 (d, $C(3^1)$); 67.72 (d, $C(4^{VI})$); 62.14, 62.05, 62.01 (2 C) (3t, $C(6^{ILV})$); 61.49 (t, $C(6^{VI})$); 61.33 (t, $C(6^1)$); 20.82 (q, 4 Me); 20.79, 20.67, 20.62, 20.61, 20.56, 20.50, 20.46 (7q, 7 Me); 20.58, 20.48 (2q, 6 Me); 20.53 (q, 2 Me); data of β -**47**: 160.74 (s, C=NH); 95.31 (d, C(1¹)). HR-MALDI-MS: 1954.4281 (55, [M+Na]⁺, C₇₆- $H_{100}Cl_3NO_{50}^+$; calc. 1954.4276). Anal. calc. for $C_{76}H_{100}Cl_3NO_{50}$ (1933.96): C 47.20, H 5.21; found: C 47.32, H 5.29

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl-[(1 \rightarrow 4)-2,3,6-tri-O-acetyl- β -D-glucopyranosyl]₅-(1 \rightarrow 4)-6-O-benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 6)-3,7-anhydro-4,5,8-tris-O-(4-chlorobenzyl)-1,2-dideoxy-1-C-(trimethylsilyl)-D-glycero-D-gulo-oct-1-ynitol (48). A suspension of 13 (1.55 g, 1.37 mmol), 47 (1.94 g, 1.0 mmol), and powdered 4-Å molecular sieves (3.00 g) in dry CH₂Cl₂ (25 ml) was stirred under Ar at 24° for 2 h, cooled to -18° (ice/NaCl bath), treated dropwise with BF₃ · OEt₂ (0.15 ml, 1.15 mmol) over 10 min, stirred for 2 h, treated with solid NaHCO₃ (0.06 g), warmed to r.t., and filtered through *Celite*. Workup, evaporation, and FC (toluene/acetone $5:1 \rightarrow 4:1$) gave **48** (2.58)

g, 88%). R_f (toluene/acetone 3:2) 0.73. M.p. 128.7–129.7°. $[a]_D^{25} = -14.8$ ($c = 1.0$, CHCl₃). IR (CHCl₃): 3030w, 2929w, 2872w, 2179w, 1756s, 1600w, 1492w, 1367m, 1160m, 1088s, 1054s, 1017m, 905w, 8 45w, 810w. ¹H-NMR (500 MHz, CDCl₃; assignments based on a HSQC and a TOCSY spectrum): 7.39-7.12 $(m, 23 \text{ arom. H})$; 6.97 (d, J=7.5, 2 arom. H); 5.115 (t, J=9.3, H-C(3^{VIII})); 5.091 (t, J=9.1), 5.077 (t, $J=9.1$), 5.066 (t, $J=9.1$), 5.051 (t, $J=9.2$) (H-C(3^{IV-VII})); 5.043 (t, $J=9.6$, H-C(4^{VIII})); 5.00 (d, $J=11.9$), 4.939 (d, $J=12.2$), 4.865 (d, $J=10.5$) (3 PhCH); 4.932 (t, $J=9.2$, H-C(3^{III})); 4.890 (dd, $J=9.2, 7.9, H-C(2^{VIII}))$; 4.813 (dd, $J=9.1, 7.8$), 4.801 (dd, $J=9.1, 7.9$), 4.794 (dd, $J=9.1, 7.8$), 4.783 $(dd, J=9.1, 7.9), 4.773 (dd, J=9.5, 7.9) (H-C(2^{III-VII})); 4.65 (d, J=10.8), 4.59 (d, J=11.9), 4.537 (d, J=11.9),$ $J=11.6$), 4.523 (d, $J=12.5$), 4.513 (d, $J\approx$ 12.5), 4.500 (d, $J\approx$ 10.5), 4.489 (d, $J\approx$ 11.6) (7 PhCH); 4.480 $(d, J \approx 7.7), 4.469$ $(d, J = 7.7), 4.437$ $(d, J = 7.7), 4.424$ $(d, J = 7.7), 4.418$ $(d, J = 7.8)$ $(H - C(1^{IV-VIII}))$; 4.42–4.32 (m, H–C($6^{\text{IV-VIII}}$)); 4.346 (d, J=7.9, H–C(1^{III})); 4.330 (d, J=7.7, H–C(1^{II})); 4.342 (d, $J=11.8$), 4.265 (d, $J=12.0$) (2 PhCH); 4.236 (dd, $J=12.5$, 2.0, H-C(6^{III})); 4.105-4.03 (m, H'-C(6^{IV–VII})); 4.03 (dd, J = 12.6, 2.2, H'–C(6^{VIII})); 3.97 (d, J = 9.6, H–C(3¹)); 3.92 (t, J = 9.4, H–C(6¹)); 3.885 (dd, $J \approx 12.5$, 4.6, H'–C(6^{III})); 3.85 (t, J=9.3, H–C(4^{II})); 3.746 (t, J=9.3), 3.727 (t, J=9.3), 3.722 (t, J=9.3), 3.708 (t, J=9.1) $(H-C(4^{IV-VII}))$; 3.75–3.71 (m, H-C(8¹)); 3.628 (t, J=9.5, H-C(4^{II1})); 3.625 (ddd, $J=9.8$, 4.4, 2.4, $H-C(5^{\text{VIII}}))$; 3.616 (dd, $J\approx 12.5$, 2.0, $H'-C(8^1)$); 3.59–3.47 (m, 2 $H-C(6^{11})$, $H-C(5^{\text{IV-VII}}))$; 3.495 (t, J = 9.3, H – C(4¹)); 3.42 (t, J = 8.9, H – C(5¹)); 3.27 (t, J = 9.0, H – C(3^{I1})); 3.24 (ddd, J = 9.5, 3.4, 1.7, H-C(7¹)); 3.17 (dd, J=9.1, 7.9, H-C(2^{II})); 3.10 (ddd, J=9.9, 4.2, 2.1, H-C(5^{III})); 3.04 (ddd, $J=10.0, 3.4, 1.7, H-C(5^{II}))$; 2.133, 2.129, 2.126, 2.109, 2.079, 2.026, 2.015, 2.009, 1.999, 1.993, 1.989, 1.987, 1.973, 1.950, 1.947, 1.944, 1.938, 1.934, 1.882 (19s, 19 AcO); 0.16 (s, Me3Si). 13C-NMR (125 MHz, CDCl3 ; assignments based on a HSQC and a TOCSY spectrum): 170.44, 170.17, 170.07, 169.63, 169.33, 169.29, 169.22, 169.15, 169.04 (9s, 9 C=O); 170.14 (s, 4 C=O); 169.68, 169.66, 169.25 (3s, 6 C=O); 137.70, 137.69, 137.66, 136.53, 136.48, 136.31 (6s, 6 arom. C); 133.53, 133.48, 133.39, 132.93, 132.88 (5s, 5 arom. C); 129.31, 129.20, 128.96, 128.82, 128.61, 128.53, 128.41, 128.32, 128.28, 128.24, 128.22, 127.76 $(12d, 24 \text{ arom. C})$; 128.10 $(d, 1 \text{ arom. C})$; 102.29 $(d, C(1^{\text{II}}))$; 102.17 (s, $C \equiv CSi$); 100.78, 100.64, 100.52, 100.49, 100.43 $(5d, C(1^{III-VII}))$; 99.77 $(d, C(1^{VIII}))$; 91.39 $(s, C \equiv CSi)$; 84.09 $(d, C(5^1))$; 82.99 $(d, C(3^{II}))$; 81.65 (d, C(2^{II})); 81.61 (d, C(4^I)); 79.17 (d, C(7^I)); 76.40 (d, C(4^{II})); 76.23 (d, C(6^I)); 76.04 (br. d, $C(4^{III-VII})$); 74.67 (d, $C(5^{II})$); 74.54, 74.29, 74.11, 74.07, 73.36, 72.63 (6t, 6 PhCH₂); 72.88, 72.82 (br.), 72.80, 72.71 (4d, C(3^{III–VIII}), C(5^{IV–VII})); 72.55 (d, C(5^{III})); 72.51 (d, C(5^{VIII})); 72.49, 72.44, 72.04, 71.86, 71.76, 71.58 (6d, C(2^{III–VIII})); 70.24 (d, C(3¹)); 67.81 (t, C(8¹)); 67.77 (d, C(4^{VIII})); 67.62 (t, C(6^{II})); 62.07, 62.05, 62.00, 61.67, 61.51 (2 C) (5t, $C(6^{III-VIII})$); 20.76–20.45 (several q, 19 Me); -0.33 (q, Me₃Si). HR-MALDI-MS: 2925.8295 (4, $[M + Na]^+$, C₁₃₃H₁₅₉Cl₅NaO₅₉Si⁺, calc. 2925.7551). Anal. calc. for C₁₃₃H₁₅₉Cl₅-O59Si (2907.03): C 54.95, H 5.51; found: C 54.88, H 5.65.

 $2,3,4,6$ -Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)-2,3,6$ -tri-O-acetyl- β -D-glucopyranosyl]₅- $(1 \rightarrow 4)-$ 6-O-benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 6)-3,7-anhydro-4,5,8-tris-O-(4-chlorobenzyl)-1,2-dideoxy-D-glycero-D-gulo-oct-1-ynitol (49). A stirred soln. of 48 (0.40 g, 0.14 mmol) in THF (15 ml) was cooled to 0° , treated with a soln. of Bu₄NF·3 H₂O (15 mg, 48 µmol) in THF (1.5 ml), stirred for 15 min, and treated with sat. aq. NH₄Cl soln. (5 ml). Workup and FC gave 49 (0.36 g, 92%). R_f (AcOEt/hexane 2:1) 0.20. M.p. 122.2°. $\left[\alpha\right]_D^{25} = -11.7$ (c=1.0, CHCl₃). IR (CHCl₃): 3307w, 3029w, 3015w, 2872w, 1757s, 1600w, 1492w, 1411w, 1367m, 1160m, 1088s, 1058s, 1017m, 905w, 843w, 810w. ¹H-NMR (300 MHz, CDCl₃): 7.35 – 7.10 (*m*, 23 arom. H); 6.96 (*d*, $J = 7.8$, 2 arom. H); 5.113 (*t*, $J = 9.3$, H- $C(3^{VIII})$); 5.084 (t, J=9.0), 5.070 (t, J=9.0), 5.065 (t, J=9.0), 5.045 (t, J=9.3) (H-C(3^{IV-VII})); 5.040 (t, $J=9.6$, H-C(4^{VIII})); 4.982 (d, J=12.0), 4.938 (d, J=12.0), 4.873 (d, J=11.1) (3 PhCH); 4.923 (t, $J=9.3, H-C(3^{III}))$; 4.892 (t, $J\approx 8.4, H-C(2^{VIII}))$; 4.85 – 4.73 (m, H – C(2^{III–VII})); 4.66 (d, J = 11.4), 4.585 $(d, J=12.3)$ (2 PhCH); 4.55 – 4.40 (m, H – C(1^{II–VIII}), H – C(6^{IV–VIII}), 6 PhCH); 4.256 (d, J = 12.0, PhCH); 4.236 (br. $d, J = 12.3$, H $- C(6^{III})$); 4.12–3.99 (m, H' $- C(6^{IV-VIII})$); 3.963 (dd, J=9.1, 2.1, H $- C(3^{I})$); 3.927 $(t, J \approx 9.4, H-C(6^1)); 3.92-3.84$ $(m, H'-C(6^{III})); 3.857$ $(t, J=9.0, H-C(4^{II})); 3.78-3.66$ $(m, H-C(4^{IV-VI}));$ $3.66 - 3.45$ (m, $2 H - C(8^1)$, $2 H - C(6^{11})$, $H - C(4^{111})$, $H - C(5^{IV-VIII})$); 3.51 (t, $J = 9.3$, $H - C(4^1)$); 3.41 (t, $J=9.0, H-C(5^1)$; 3.275 (t, $J=9.0, H-C(3^{\text{II}})$); 3.235 (br. d, $J\approx 9.6, H-C(7^1)$); 3.17 (dd, $J=9.0, 7.8, H-J$ $C(2^{II})$; 3.08 (br. *d*, *J* \approx 9.9, H-C(5^{III})); 3.04 (br. *d*, *J* \approx 9.9, H-C(5^{II})); 2.52 (*d*, *J*=2.1, C=CH); 2.13–1.87 (several s, 19 AcO). HR-MALDI-MS: 2853.7135 (32, $[M + Na]^+$, $C_{130}H_{151}Cl_5NaO_{59}^+$; calc. 2853.7150). Anal. calc. for C₁₃₀H₁₅₁Cl₅O₅₉ (2834.85): C 55.08, H 5.37; found: C 55.19, H 5.61.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl J_6 - $(1 \rightarrow 6)$ - $4,5,8-tri-O-acetvl-1,2-dideoxv-D-glycero-D-gulo-oct-1-vnitol$ (50). A soln. of 49 (0.60 g, 0.21 mmol) in Ac₂O (6 ml) was cooled to -18° (ice/NaCl bath), treated with BF₃ \cdot OEt₂ (0.54 ml, 4.23 mmol), stirred for 1 h, warmed to 24° , stirred for additional 2 d, poured into chilled sat. aq. NaHCO₃ soln. (30 ml), and stirred vigorously for 0.5 h. Workup (AcOEt), evaporation, and FC (AcOEt/hexane $1:1 \rightarrow 5:2$) gave 50 (0.41 g, 82%). White solid. R_f (toluene/acetone 3:2) 0.44. M.p. 263.7° (dec.). $\left[\alpha\right]_D^{25} = -18.8$ $(c=1.0, CHCl₃)$. IR (CHCl₃): $3307w$, $3029w$, $2958w$, $2872w$, $1756s$, $1600w$, $1429w$, $1368m$, $1162m$, $1051s$, 952w, 902w, 836w. ¹H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY and a HSQC spectrum): see Table 13; additionally, 2.139, 2.137, 2.125, 2.084, 2.053, 2.031, 2.020, 2.008, 2.004, 1.991, 1.952 (11s, 11 AcO); 2.133, 1.948(2s, 6 AcO); 2.013 (s, 4 AcO); 1.978, 1.950 (2s, 4 AcO). 13C-NMR (125 MHz, CDCl₃; assignments based on a HSQC spectrum): see *Table 14*; additionally, 170.41, 170.26, 169.70, 169.62, 169.37, 169.26, 169.01 (7s, 7C=O); 170.13, 170.12, 169.21 (3s, 9C=O); 170.11, 169.19 $(2s, 4 \text{ C}=0); 169.66 \text{ (}s, 5 \text{ C}=0); 20.81, 20.59, 20.53 \text{ (}3q, 3 \text{ Me}); 20.73, 20.50 \text{ (}2q, 8 \text{ Me}); 20.71, 20.48,$ 20.46 (3q, 9 Me); 20.40 (q, 5 Me). HR-MALDI-MS: 2395.6891 (90, $[M + Na]^+$, $C_{100}H_{132}NaO⁺₆₅$; calc. 2395.6921). Anal. calc. for C₁₀₀H₁₃₂O₆₅ (2374.10): C 50.59, H 5.60; found: C 50.68, H 5.76.

 $2,3,4,6$ -Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ - $2,3,6$ -tri-O-acetyl- β -D-glucopyranosyl] ζ - $(1 \rightarrow 4)$ -6-O-benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 8)-5,9-anhydro-6,7,10-tris-O-(4-chlorobenzyl)-1,2,3,4-tetradeoxy-1-C-(trimethylsilyl)-D-glycero-D-gulo-deca-1,3-diynitol (51). A suspension of 17 (0.72 g, 0.62 mmol), α/β -47 8:1 (1.09 g, 0.56 mmol) and powdered 4-Å mol. sieves (1.0 g) in dry CH₂Cl₂ (16 ml) was stirred under Ar at 28° for 2 h, cooled to -40° , treated dropwise with BF₃ · OEt₂ (82 µl, 0.65 mmol) over 10 min, slowly warmed to -20° , stirred for 1 h, treated with solid NaHCO₃ soln. (0.12 g) , and filtered through *Celite*. The filtrate was washed with aq. NaHCO₃ soln. Workup, evaporation, and FC (AcOEt/cyclohexane 1:2 \rightarrow 1:1) gave 51 (1.35 g, 72%). R_f (toluene/acetone 3:2) 0.71. M.p. 126.7° (dec.). $\left[\alpha\right]_D^{25} = -28.4$ (c=1.0, CHCl₃). IR (CHCl₃): 3029w, 3014w, 2872w, 2112w, 1756s, 1711m, 1492w, 1414w, 1366m, 1227s, 1160m, 1088m, 1055s, 1017w, 908w, 845w, 810w. ¹H-NMR (300 MHz, CDCl₃): 7.37 – 7.13 (m, 23 arom. H); 6.96 (d, J = 7.2, 2 arom. H); 5.113 (t, J = 9.3, H – C(3^{VIII})); 5.11 – 4.99 (m, H – C(3^{IV–VII}), H – C(4^{VIII}), 1 PhCH); 4.936 (d, J = 11.1, PhCH); 4.922 (t, J = 9.3, H – $C(3^{III})$); 4.891 (t, J \approx 8.4, H – $C(2^{VIII})$); 4.84 – 4.72 (m, H – $C(2^{III-VII})$, 1 PhCH); 4.644 (d, J = 11.1), 4.580 $(d, J=12.3)$ (2 PhCH); 4.54–4.28 (m, H–C(1^{II–VIII}), H–C(6^{IV–VIII}), 6 PhCH); 4.255 (d, J = 12.0, PhCH); 4.235 (br. d, J = 12.3, H – C(6^{III})); 4.11 – 3.98 (m, H' – C(6^{IV–VIII})); 3.995 (d, J = 9.0, H – C(5^I)); 3.91 – 3.83 $(m, H'-C(6^{III}));$ 3.895 $(t, J=9.3, H-C(8^I));$ 3.855 $(t, J=9.3, H-C(4^{II}));$ 3.78–3.65 $(m, H-C(10^I), H C(4^{IV-VIII})$); 3.66–3.42 (m, H'–C(10¹), 2H–C(6^{II}), H–C(4^{III}), H–C(5^{IV–VIII})); 3.470 (t, J=9.0, H– $C(6^1)$); 3.425 (t, J=9.0, H-C(7¹)); 3.276 (t, J=9.0, H-C(3^{II})); 3.25-3.19 (m, H-C(9^I)); 3.167 (dd, $J=9.0, 8.1, H-C(2^H))$; 3.07 (br. d, $J\approx 10.0, H-C(5^{III})$); 3.04 (br. d, $J\approx 10.0, H-C(5^H))$; 2.137, 2.134, 2.130, 2.113, 2.082, 2.029, 2.018, 2.012, 2.002, 1.996, 1.993, 1.988, 1.976, 1.950, 1.947, 1.945, 1.938 (6 H), 1.881 (18s, 19 AcO); 0.20 (s, Me₃Si). ¹³C-NMR (75 MHz, CDCl₃): 170.29–168.90 (several s, 19 C=O); 137.53, 137.46, 137.39, 136.35, 136.13, 135.96 (6s, 6 arom. C); 133.54, 133.34, 133.26, 132.89, 132.75 (5s, 5 arom. C); 129.50, 129.15 (2d, 4 arom. C); 128.91 – 128.12 (several d, 19 arom. C); 127.66 (d, 2 arom. C); 102.26 (d, C(1^{II})); 100.74 (2 C), 100.60, 100.48 (2 C) (3d, C(1^{III-VII})); 99.68 (d, C(1^{VIII})); 88.32 (s, $C \equiv CSi$; 86.93 (s, C $\equiv CSi$); 84.15 (d, C(7¹)); 82.92 (d, C(3¹¹)); 81.55 (d, C(2¹¹)); 81.01 (d, C(6¹)); 79.24 $(d, C(9^I))$; 76.28, 76.24 (2d, C(8¹), C(4^{II}))); 76.04 (br. d, C(4^{III–VII})); 74.68 (d, C(5^{II})); 74.60, 74.42, 74.12 (2 C) , 73.32, 72.58 (5t, 6 PhCH₂); 72.84, 72.76 (br.), 72.50 (br.) (3d, C(3^{III–VIII}), C(5^{III–VIII})); 72.00 (2 C), 71.78 (2 C), 71.70 , 71.52 (4d, C($2^{\text{III-VIII}}$)); 70.17 (d, C(5^{I})); 67.71 (t, C(10^{I})); 67.67 (d, C(4^{VIII})); 67.62 (t, $C(6^{II})$; 62.02 (br., 4 C), 61.59, 61.47 (3t, $C(6^{III-VIII})$; 20.89–20.62 (several q, 19 Me); -0.37 (q, Me₃Si); ss of C≡CSi hidden by other signals. HR-MALDI-MS: 2949.7831 (20, $[M+Na]^+$, C₁₃₅H₁₅₉Cl₅NaO₅₉Si⁺; calc. 2949.7551). Anal. calc. for C₁₃₅H₁₅₉Cl₅O₅₉Si (2931.05): C 55.32, H 5.47; found: C 55.31, H 5.60.

 $2,3,4,6$ -Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)-2,3,6$ -tri-O-acetyl- β -D-glucopyranosyl]₅- $(1 \rightarrow 4)-$ 6-O-benzyl-2,3-bis-O-(4-chlorobenzyl)- β -D-glucopyranosyl-(1 \rightarrow 8)-5,9-anhydro-6,7,10-tris-O-(4-chlorobenzyl)-1,2,3,4-tetradeoxy-D-glycero-D-gulo-deca-1,3-diynitol (52). A soln. of 51 (0.82 g, 0.28 mmol) in THF (25 ml) was cooled to 0° , treated with a soln. of Bu₄NF·3 H₂O (30.8 mg, 0.10 mmol) in THF (2.5 ml), stirred for 30 min, and treated with sat. aq. NH4Cl soln. (10 ml). Workup and FC (AcOEt/cyclohexane 1:1 \rightarrow 3:2) gave 52 (0.66 g, 83%). Foamy solid. R_f (toluene/acetone 2:1) 0.55. M.p. 118.2° (dec.). $[\alpha]_{\text{D}}^{25}$ = -26.3 (c=1.0, CHCl₃). IR (CHCl₃): 3306w, 3030w, 2958w, 2872w, 1756s, 1600w, 1492w, 1367m,

Table 13. Selected ¹H-NMR Chemical-Shift Values [ppm] and Coupling Constants [Hz] of the Peracetylated Cellooctaosylalkynes 50, 53-55, and 58 in CDCl₃ Table 13. Selected 1H-NMR Chemical-Shift Values [ppm] and Coupling Constants [Hz] of the Peracetylated Cellooctaosylalkynes 50, 53 –55, and 58 in CDCl3

724 Helvetica Chimica Acta – Vol. 89 (2006)

	$50a$)	53	54	58	57	59
				E chain, B chain		E chain, B chain
C(4')		66.82	$\overline{}$	$-$, 78.68		$-.78.80$
C(3')		69.80	$\qquad \qquad -$	$-$, 78.31		$-$, 77.33
C(2')	75.31	69.48	85.90	85.92, 71.97	83.89	84.23, 69.68
C(1')	77.72	71.12	88.93	90.20, 79.31	94.53	94.05, 84.23
$C(1^1)$	68.22	68.66	69.49	69.45, 69.05	70.74	70.72, 70.41
C(1 ^H)	100.73		100.73 100.81	100.81	103.13	103.13 $(2 C)$
$C(1^{\text{III-VIII}})$	100.48, 100.43		100.43 100.56, 100.51	100.81 (2C), 100.56	103.13, 102.72	103.13 (2C),
	$(4 C)$, 100.37		$(4 C)$, 100.49	$(4 C)$, 100.52 $(6 C)$	$(4 C)$, 102.61	102.72 $(10 C)$
$C(2^I)$	71.20	70.95	71.41	71.13/71.08	73.44	73.57/73.14
$C(2^{II-VII})$	71.74 (5 C),	71.77	71.88, 71.83	71.82 $(8 C)$,	73.15, 72.97,	73.14 (2 C),
	71.65	$(5C)$, 71.71	(4 C), 71.73	71.73 $(4 C)$	72.88 (4 C)	72.87 (10 C)
$C(2^{VIII})$	71.48	71.52	71.57	71.56	73.15	73.14 (2C)
$C(3^I)$	72.94	72.86	73.18	73.21/73.12	75.84	75.79/75.68
$C(3^{II-VII})$	72.72 (5 C),	72.76	72.52 (br.)	72.70 $(3 C)$,	74.65 (6 C)	74.65 (12 C)
	72.67			72.54 (9 C)		
$C(3^{VIII})$	72.80	72.81	72.63	72.63	76.38	76.38 (2 C)
$C(4^{I-VII})$	76.11, 76.05,	76.03	76.39, 76.19,	76.09	$80.27(6 \text{ C}),$	80.27 (12 C),
	76.00 (5 C)		76.08 (4 C)		80.01	79.95/79.82
$C(4^{VIII})$	67.64	67.67	67.72	67.71	69.94	69.94 (2 C)
$C(5^I)$	76.81	77.20	76.92	76.48, 76.20	78.91	78.94/78.80
$C(5II-VII)$	72.55, 72.44	72.55	72.88, 72.81	72.88 (3 C),	74.72 (6 C)	74.71 (12C)
	(5C)	(2 C),	(4 C), 72.79	72.80 (9 C)		
		72.47				
		(4C)				
$C(5^{VIII})$	71.96	72.00	72.04	72.03	76.70	76.69 (2 C)
$C(6^{I-VII})$	62.05, 61.98	61.99	62.25, 62.18,	62.25, 62.01	60.94, 60.20	60.93 $(2 C)$,
	(3 C), 61.94 (3 C)		62.05 $(5 C)$	(8 C), 61.48 (3 C)	(6C)	60.19 (12C)
$C(6^{VIII})$	61.41	61.46	61.49	61.48	60.94	60.93 $(2 C)$

Table 14. Selected ¹³C-NMR Chemical-Shift Values [ppm] of the Peracetylated Cellooctaosylalkynes **50**, **53**, 54, and 58 in CDCl₃, and of the Deprotected Cellooctaosylalkynes 57 and 59 in $(D_6)DMSO$

1160w, 1088s, 1055s, 1017m, 904w. ¹H-NMR (500 MHz, CDCl₃): 7.38–7.10 (m, 23 arom. H); 6.96 (d, $J=7.8$, 2 arom. H); 5.108 (t, $J=9.3$, H-C(3^{VIII})); 5.10–4.99 (m, H-C($3^{\text{IV-VII}}$), H-C(4^{VIII}), 1 PhCH); 4.936 (d, J \approx 10.8, PhCH); 4.918 (t, J = 9.3, H – C(3^{III})); 4.887 (t, J \approx 8.4, H – C(2^{VIII})); 4.84 – 4.72 (m, H – $C(2^{III-VII})$, 1 PhCH); 4.647 (d, J=10.8), 4.580 (d, J=12.0) (2 PhCH); 4.56–4.28 (m, H-C(1^{II–VIII}), H- $C(6^{IV-VIII})$, 6 PhCH); 4.25 (d, J=12.3, PhCH); 4.23 (br. d, J=12.3, H-C(6^{III})); 4.12-3.98 (m, H'- $C(6^{IV-VIII})$; 3.98 (br. d, J=9.3, H-C(5¹)); 3.90–3.83 (m, H'-C(6^{III})); 3.89 (t, J=9.3, H-C(8¹)); 3.85 (t, $J=9.3$, H $-C(4^{II})$); 3.78–3.65 (m, H $-C(10^I)$, H $-C(4^{IV-VII})$); 3.66–3.45 (m, H' $-C(10^I)$, 2 H $-C(6^{II})$, H $C(4^{III})$, $H-C(5^{IV-VIII})$); 3.48 (t, $J=8.7$, $H-C(6^{1})$); 3.43 (t, $J=8.7$, $H-C(7^{1})$); 3.265 (t, $J=9.0$, $H-C(3^{II})$); $3.26 - 3.18$ (m, H-C(9¹)); 3.16 (dd, $J \approx 8.5$, H-C(2^{I1})); 3.08 (br. d, $J \approx 10.0$, H-C(5^{III})); 3.03 (br. d, $J \approx 10.0$, H-C(5^{II})); 2.22 (br. s, C \equiv CH); 2.12–1.86 (several s, 19 AcO). ¹³C-NMR (75 MHz, CDCl₃): 170.29 – 168.89 (several s, 19 C=O); 137.52, 137.45, 137.36, 136.34, 136.04, 135.86 (6s, 6 arom. C); 133.60, 133.38, 133.26, 132.91, 132.75 (5s, 5 arom. C); 129.45, 129.18, 128.84, 128.76, 128.52, 128.22 (6d, 12 arom. C); 128.44 (d, 4 arom. C); 128.15 (d, 6 arom. C); 128.05 (d, 1 arom. C); 127.65 (d, 2 arom. C); 102.26 (d, C(1^{II})); 100.73, 100.59, 100.47 (2 C), 100.41 (4d, C(1^{III–VII})); 99.68 (d, C(1^{VIII})); 84.13 (d, C(7¹)); 82.92 (d, $C(3^H)$); 81.54 (d, $C(2^H)$); 80.88 (d, $C(6^H)$); 79.23 (d, $C(9^H)$); 76.25 , 76.22 (2d, $C(8^H)$, $C(4^H)$); 76.04 (br. d, $C(4^{III-VII})$); 74.73 (d, $C(5^{II})$); 74.60, 74.42, 74.12 (2 C), 73.32, 72.57 (5t, 6 PhCH₂); 72.85, 72.75 (br.), 72.49 (br.) $(3d, C(3^{III-VIII}), C(5^{III-VIII}))$; 71.99 (2 C), 71.78 (2 C), 71.70, 71.52 (4d, C(2^{III–VIII})); 70.17 (d, $C(5^I)$); 70.00 (s, C=C-C=CH); 67.67 (t, C(10^I)); 67.67 (d, C(4^{VIII})); 67.62 (t, C(6^{II})); 67.21 (s, C=C- $C \equiv CH$); 62.03 (br., 4 C), 61.63, 61.47 (3t, $C(6^{III-VIII})$); 20.89–20.66 (several q, 19 Me); s of $C \equiv C - C \equiv$ CH hidden by other signals. Anal. calc. for $C_{132}H_{151}Cl_5O_{59}$ (2858.87): C 55.46, H 5.32; found: C 55.49, H 5.46.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl $J_{\alpha}(1 \rightarrow 8)$ -5,9-anhydro-6,7,10-tri-O-acetyl-1,2,3,4-tetradeoxy-D-glycero-D-gulo-deca-1,3-diynitol (53). A soln. 52 (2.13 g, 0.75 mmol) in Ac₂O (20 ml) was cooled to -18° (ice/NaCl bath), treated with BF₃ ·OEt₂ (1.90) ml, 15.01 mmol), warmed to 31°, stirred for 3 d, poured into a chilled sat. aq. NaHCO₃ soln. (60 ml), and stirred vigorously for 0.5 h. Workup (AcOEt), evaporation, and FC (AcOEt/hexane $1:1 \rightarrow 5:2$) gave 53 (1.26 g, 70%). Pale yellow solid. R_f (toluene/acetone 3:2) 0.49. M.p. 165.2° (dec.). $[\alpha]_{\text{D}}^{25}$ = -20.6 (c = 1.03, CHCl₃). IR (CHCl₃): 3300w, 3029w, 2969w, 2119w, 1756s, 1602w, 1430w, 1368m, 1162w, 1054s, 904w, 846w. ¹H-NMR (300 MHz, CDCl₃): see *Table 13*; additionally, 2.135, 2.002, 1.975 (3s, 6 AcO); 2.130 (s, 5 AcO); 2.082, 2.060, 2.029, 2.018, 1.988 (5s, 5 AcO); 2.011 (s, 4 AcO); 1.944 (br. s, 5 AcO). ¹³C-NMR (75 MHz, CDCl₃): see *Table 14*; additionally, 170.30, 170.10, 168.90 (3s, 3 C=O); 170.00, 169.54 (2s, 14 C=O); 169.16 (s, 2 C=O); 169.10 (s, 6 C=O); 20.89 (br. q, 5 Me); 20.75 (q, 2 Me); 20.66 (br. q , 13 Me); 20.89 $(q, 5 \text{ Me})$. HR-MALDI-MS: 2419.6950 $(83, [M + Na]^+, C_{102}H_{132}NaO_6^+;$ calc. 2419.6921). Anal. calc. for $C_{102}H_{132}O_{65}$ (2398.13): C 51.09, H 5.55; found: C 51.08, H 5.78.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl J_6 - $(1 \rightarrow 6)$ -4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-1-C-(8-acetoxy-9,10-dioxoanthracen-1-yl)-D-glycero-D-gulooct-1-ynitol (54). Under Ar, a suspension of 20 (0.38 g, 0.93 mmol), $[Pd(PPh_3)_2]C_2$ (29.6 mg, 0.04 mmol), CuI (24.1 mg, 0.13 mmol), and Bu₄NI (0.47 g, 1.26 mmol) in degassed Et₃N/DMF 1:5 (20 ml) was cooled to 0° , treated dropwise with a soln. of 50 (2.0 g, 0.84 mmol) in Et₃N/DMF 1:5 (20 ml) over a period of 10 h (with a syringe pump), warmed to 28° , stirred for 6 h, and poured into ice/H₂O (400 ml). Workup (AcOEt), evaporation, and FC (toluene/acetone 3:1) gave 54 (1.85 g, 83%). Bright yellow solid. R_f (toluene/acetone 3:2) 0.49. M.p. 219.6° (dec.). $[a]_D^{25} = -9.4$ ($c = 1.03$, CHCl₃). IR (CHCl₃): 3029w, 2951w, 2863w, 1756s, 1598w, 1434w, 1368m, 1323w, 1205m, 1160w, 1054m, 902w, 8 46w. ¹ H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY spectrum): see *Table 13*; additionally, 8.27 (dd, $J = 7.8$, 1.4, $H-C(4')$; 8.23 (dd, J = 7.8, 1.3, H - $C(5')$); 7.83 (dd, J = 7.8, 1.4, H - $C(2')$); 7.78 (t, J = 7.9, H - $C(6')$); 7.68 (t, J = 7.8, H - C(3')); 7.43 (dd, J = 8.0, 1.3, H - C(7')); 2.50 (s, AcO - C(8')); 2.153, 2.150, 2.144, 2.140, 2.139, 2.137, 2.136, 2.086, 2.085, 2.034, 2.032, 2.023, 2.022, 2.020, 2.007, 2.005, 1.993, 1.979, 1.962 (19s, 19 AcO); 2.016, 1.954 (2s, 4 AcO); 1.953 (br. s, 2 AcO). ¹³C-NMR (125 MHz, CDCl₃): see Table 14; additionally, 181.99 (s, C(9')); 180.56 (s, C(10')); 170.49, 170.37, 169.86, 169.70, 169.68, 169.34, 169.10 (7s, 7C=O); 170.22 (s, 4C=O); 170.20 (s, 3C=O); 169.74 (s, 5C=O); 169.30 (s, 7C=O); 150.17 (s, C(8')); 140.79 (d, C(2')); 134.72 (s, C(9'a)); 134.62 (d, C(3')); 134.46 (s, C(4'a)); 133.53 (s, C(10'a)); 132.75 (d, C(6')); 130.25 (d, C(4')); 127.72 (d, C(7')); 125.51 (d, C(5')); 125.46 (s, C(1')); 121.77 (s, C(8'a)); 21.22 (q, MeC(=O)-C(8')); 20.94, 20.79, 20.70, 20.67, 20.61, 20.56 (7q, 7 Me); 20.81, 20.48 (2q, 10 Me); 20.58 (q, 6 Me); 20.54 (q, 2 Me). HR-MALDI-MS: 2659.7390 (36, $[M + Na]^+$, C₁₁₆H₁₄₀- $NaO₆₉⁺$; calc. 2659.7344). Anal. calc. for $C₁₁₆H₁₄₀O₆₉$ (2638.34): C 52.81, H 5.35; found: C 52.94, H 5.43.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl J_6 - $(1 \rightarrow 6)$ -4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-1-C-(8-hydroxy-9,10-dioxoanthracen-1-yl)-D-glycero-D-gulooct-1-ynitol (55). A stirred soln. of 54 (1.63 g, 0.62 mmol) in DMF (30 ml) was treated with $(NH₄)₂CO₃$ (0.18 g, 1.92 mmol), stirred for 10 h at 26° , and poured into chilled H₂O (300 ml). Workup, evaporation, and FC (AcOEt/cyclohexane $3:1 \rightarrow 3:2$) gave 55 (1.42 g, 88%). White solid. R_f (toluene/acetone 3:2) 0.48. M.p. 227° (dec.). ¹H-NMR (300 MHz, CDCl₃): see *Table 13*; additionally, 12.50 (s, OH); 8.33 (dd, $J=7.8, 1.2, H-C(4'))$; 7.89 (dd, $J=7.8, 1.2, H-C(2'))$; 7.81 (dd, $J=7.5, 1.2, H-C(5'))$; 7.73 (t, $J=7.7$, $H-C(3')$); 7.67 (t, J=7.8, H-C(6')); 7.32 (dd, J=8.4, 1.2, H-C(7')); 2.151, 2.134, 2.030, 2.019, 2.016 (5s, 10 AcO); 2.137, 2.013, 1.949 (3s, 9 AcO); 2.106, 2.082, 2.003, 1.989, 1.977, 1.958 (6s, 6 AcO). HR-MALDI-MS: 2617.7270 (40, $[M + Na]^+, C_{114}H_{138}NaO_{68}^+$; calc. 2617.7238).

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl J_6 - $(1 \rightarrow 6)$ -4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-1-C-{8-[(trifluoromethyl)sulfonyloxy]-9,10-dioxoanthracen-1 y l/-D-glycero-D-gulo-oct-1-ynitol (56). A soln. of 55 (140 mg, 0.05 mmol) in dry CH₂Cl₂ (1.2 ml) was treated with Et₃N (22 μ l, 0.16 mmol) and cooled to -78° . The resulting red suspension was treated with Tf₂O (18 μ l, 0.11 mmol), stirred for 1 h, diluted with CH₂Cl₂, and warmed to r.t. Workup, evaporation, and crystallisation from CH₂Cl₂/MeOH gave 56 (81 mg, 59%). Yellow solid. R_f (toluene/acetone 3:2) 0.48. M.p. 237° (dec.). HR-MALDI-MS: 2749.6744 (26, $[M+Na]^+$, $C_{115}H_{137}F_3NaO_{70}S^+$; calc. 2749.6731).

 β -D-Glucopyranosyl-[(1 \rightarrow 4)- β -D-glucopyranosyl]₆-(1 \rightarrow 6)-3,7-anhydro-1,2-dideoxy-1-C-(8-hy $droxy-9,10-dioxoanthracen-1-yl)$ -D-glycero-D-gulo-*oct-1-ynitol* (57). A suspension of 55 (180 mg, 0.07) mmol) in 0.1M aq. Bu₄NOH (2.7 ml, 2.7 mmol) was sonicated for 20 min, stirred for 2 d at 26 $^{\circ}$, and neutralised with *Amberlite IR-120* resin $(H^+$ form). The supernatant was decanted, centrifuged (4000 rpm), and diluted with H₂O. Lyophilisation gave 57 (46 mg, 43%). Yellowish brown solid. M.p. 138° (dec., \rightarrow green residue). IR (KBr): 3400s, 2892m, 2119w, 1664m, 1636m, 1579w, 1454m, 1364m, 1317m, 1284m, 1240m, 1200m, 1158s, 1071s, 896m, 746m, 665m, 611w. ¹H-NMR (600 MHz, (D₆)DMSO; assignments based on a DQFCOSY spectrum): see Table 1; additionally, 12.48 (br. s, HO-C(8')); 8.24 (d, $J=7.7$, H-C(4')); 8.01 (d, J=6.7, H-C(2')); 7.91 (t, J=7.7, H-C(3')); 7.81 (t, J=8.4, H-C(6')); 7.73 (d, $J=7.9, H-C(5'))$; 7.41 $(d, J=8.5, H-C(7'))$; 4.38 $(d, J=7.8, H-C(1^{VI1}));$ 4.32 (br. $d, J \approx 7.5, H-C(3^I),$ $H - C(1^{H-VI})$); 4.24 (d, J = 7.8, H - C(1^{VIII})); 3.79 (br. s, H - C(8^I), H - C(6^{II–VII})); 3.70 (br. d, J = 9.8, H - $C(6^{VIII})$); 3.57 (br. s, H'– $C(8^1)$, H'– $C(6^{II-VII})$); 3.42–3.26 (m, H–C(4¹), H–C(5¹), H–C(6¹), H–C(7¹), $H'-C(6^{VIII})$, $H-C(3^{II-VII})$, $H-C(4^{II-VII})$, $H-C(5^{II-VII})$; 3.26–3.13 (m, $H-C(3^{VIII})$, $H-C(5^{VIII})$); 3.07 – 2.89 (m, H – C(2^{II–VIII}), H – C(4^{VIII})). ¹³C-NMR (151 MHz, (D₆)DMSO): see *Table 14*; additionally, 186.94 (s, C(9')); 181.50 (s, C(10')); 161.36 (s, C(8')); 141.95 (d, C(2')); 136.74 (d, C(6')); 134.11 (s, $C(9'a)$; 133.95 (d, C(3')); 132.63 (s, C(4'a)); 132.60 (s, C(10'a)); 127.28 (d, C(4')); 124.20 (d, C(7')); 122.10 (d, C(1')); 118.60 (d, C(5')); 116.32 (s, C(8'a)). MALDI-TOF: 1567.5 ([M+Na]⁺, C₆₄H₈₈NaO₄₃; calc. 1567.6). Anal. calc. for $C_{64}H_{88}O_{43} \cdot 2 H_{2}O$ (1581.39): C 48.61, H 5.86; found: C 48.79, H 6.09.

2,3,4,6-Tetra-O-acetyl- β -D-glucopyranosyl- $[(1 \rightarrow 4)$ -2,3,6-tri-O-acetyl- β -D-glucopyranosyl J_6 - $(1 \rightarrow 8)$ -6,7,10-tri-O-acetyl-5,9-anhydro-1,2,3,4-tetradeoxy-1-C-(8-{2,3,4,6-tetra-O-acetyl-b-D-glucopyranosyl-[(1 \rightarrow 4)-2,3,6-tri-O-acetyl- β -D-glucopyranosyl[₆-(1 \rightarrow 6)-4,5,8-tri-O-acetyl-3,7-anhydro-1,2-dideoxy-D-glycero-D-gulo-oct-1-ynitol-1-yl}-9,10-dioxoanthracen-1-yl)-D-glycero-D-gulo-deca-1,3-diynitol (58). Under Ar, a soln. of 56 (0.43 g, 0.16 mmol), $[Pd(PPh₃)_2]Cl₂$ (5.5 mg, 7.9 µmol), CuI (4.5 mg, 5.5 µmol), and Bu₄NI (88 mg, 55 µmol) in degassed DMF/Et₃N 5 : 1 (1.8 ml) was cooled to -18° (ice/NaCl bath), treated dropwise with a soln. of 53 (0.42 g, 0.18 mmol) in degassed DMF/Et₃N 5:1 (1.8 ml) over 10 h (with a syringe pump), warmed to 28° , stirred for 16 h, and poured into chilled H₂O (80 ml). Workup (AcOEt), evaporation, and FC (toluene/acetone $3:1 \rightarrow 2:1$) gave 58 (0.31 g, 40%). Yellow solid. R_f (toluene/acetone 1:1) 0.56. M.p. 166.2° (dec.). $\left[\alpha\right]_D^{25} = -8.1$ (c=1.0, CHCl₃). IR (CHCl₃): 3029w, 2957w, 2873w, 2142w, 1756s, 1677w, 1566w, 1431w, 1368m, 1316w, 1161w, 1052s, 953w, 903w. ¹ H-NMR (500 MHz, CDCl₃; assignments based on a DQFCOSY spectrum): see *Table 13*; additionally, 8.32 (dd, $J=7.8, 1.3$), 8.29 (dd, $J=7.8, 1.3$) (H-C(4'), H-C(5')); 7.95 (dd, $J=7.7, 1.3$), 7.92 (dd, $J=7.7, 1.3$) $(H-C(2'), H-C(7'))$; 7.72 (t, J=7.8), 7.70 (t, J=7.8) (H-C(3'), H-C(6')); 2.185, 2.166, 2.119, 1.962 (4s, 4 AcO); 2.144, 2.033, 2.020, 2.006, 1.993, 1979 (6s, 18AcO); 2.141, 2.137 (2s, 10 AcO); 2.085, 2.027, 2.001 (3s, 6 AcO); 2.023, 2.017, 1.953 (3s, 12 AcO). ¹³C-NMR (125 MHz, CDCl₃): see *Table 14*; additionally, 181.75 (s, C(10')); 179.76 (s, C(9')); 170.50 – 169.09 (several s, 50 C=O); 142.24 (d, C(2')); 141.55 (d, $C(7')$; 135.29 (s, $C(9'a)$); 133.69 (s, $C(8'a)$); 133.61(s, $C(4'a)$); 133.46 (s, $C(10'a)$); 132.90 (d, $C(3')$); 132.81 (d, C(6')); 128.15 (d, C(4')); 127.77 (d, C(5')); 122.35 (s, C(8')); 121.46 (d, C(1')); 21.00, 20.93, 20.91, 20.74 (4q, 4 Me); 20.81 (q, 9 Me); 20.79 (q, 4 Me); 20.67 (q, 3 Me); 20.58(q, 10 Me); 20.56, 20.53 (2q, 12 Me); 20.47 $(q, 8 \text{ Me})$. HR-MALDI-MS: 4996.4282 $(7, [M + Na]^+, C_{216}H_{268}NaO_{132}^+$; calc. 4996.4156). Anal. calc. for $C_{216}H_{268}O_{132}$ (4976.41): C 52.13, H 5.43; found: C 52.15, H 5.57.

 β -D-Glucopyranosyl-[(1 \rightarrow 4)- β -D-glucopyranosyl]₆-(1 \rightarrow 8)-5,9-anhydro-1,2,3,4-tetradeoxy-1-C-(8- ${\beta}$ -D-glucopyranosyl- $[(1 \rightarrow 4)$ - β -D-glucopyranosyl]₆- $(1 \rightarrow 6)$ -3,7-anhydro-1,2-dideoxy-D-glycero-D-gulooct-1-ynitol-1-yl}-9,10-dioxoanthracen-1-yl)-D-glycero-D-gulo-deca-1,3-diynitol (59). A suspension of 58 $(150 \text{ mg}, 0.03 \text{ mmol})$ in 0.1m aq. Bu₄NOH $(30 \text{ ml}, 3.0 \text{ mmol})$ was sonicated for 20 min, stirred for 2 d at 26°, and neutralised with *Amberlite IR-120* resin (H⁺ form). The supernatant was decanted and centrifuged (4000 rpm, washing with H₂O). Lyophilisation gave 59 (48 mg, 55%). Yellowish brown solid. M.p. 268° (dec., \rightarrow green residue). IR (KBr): 3368s, 2899m, 2143w, 1673m, 1633m, 1572w, 1430m, 1372m, 1334s, 1256m, 1199m, 1160s, 1067s, 897w, 801m, 741m, 662s, 599s. ¹H-NMR (600 MHz, (D_6) DMSO; assignments based on a DQFCOSY and a HSQC spectrum): see *Table 1*; additionally, 8.26 (d, J = 7.9), 8.23 (d, J = 8.4) (H-C(4'), H-C(5')); 8,13 (d, J = 7.5), 8.03 (dd, J = 7.4) (H-C(2'), H- $C(7)$); 7.90 (t, J = 7.4), 7.89 (t, J = 8.4) (H - C(3'), H - C(6')); 4.33 – 4.27 (m, H - C(3^{LE}), H - C(5^{LB}), H - $C(1^{11-VIIIB/E})$); 3.80–3.79 (m, H–C(8^{IE}), H–C(10^{IB}), H–C(6^{II–VIIB/E})); 3.70–3.69 (m, H–C(6^{VIIIB/E})); 3.57 (br. s, H'–C(8^{LE}), H'–C(10^{LB}), H'–C($6^{\text{II-VIIB/E}}$)); 3.45–3.33 (m, H–C(4^{LE}), H–C(5^{LE}), H–C(6^{LE}), $\rm{H-C(7^{IE}),\ H-C(6^{IB}),\ H-C(7^{IB}),\ H-C(8^{IB}),\ H-C(9^{IB}),\ H-C(3^{II-VIIB/E}),\ H-C(4^{II-VIIB/E}),\ H-C(5^{II-VIIIB/E}),}$ $H'-C(6^{VBE})$); 3.20–3.15 (m, H-C(3^{VIIIB/E}), H-C(5^{VIIIB/E})); 3.07–2.99 (m, H-C(4^{VIIIB/E}), H-C- $(2^{II-VIIIB/E})$). ¹³C-NMR (151 MHz, CDCl₃): see *Table 14*; additionally, 181.44 (s, C(10')); 179.57 (s, $C(9')$; 141.97 (d, $C(2')$); 141.71 (d, $C(7')$); 135.02 (s, $C(9'a)$); 133.49 (s, $C(8'a)$); 133.37 (s, $C(4'a)$); 133.28(d, C(3'), C(6')); 133.09 (s, C(10'a)); 127.58(d, C(4')); 127.00 (d, C(5')); 122.09 (s, C(8')); 120.19 (s, C(1')). MALDI-TOF: 2899.8 $([M+Na+4 H]^+, C_{116}H_{172}NaO₈₂$; calc. 2899.9). Anal. calc. for $C_{116}H_{168}O_{82} \cdot 8$ H₂O (3018.65): C 46.15, H 6.14; found: C 46.06, H 6.24.

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